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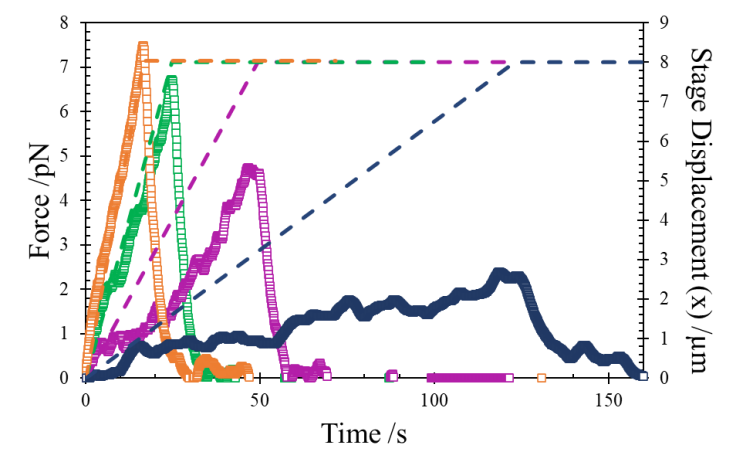
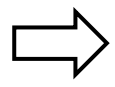
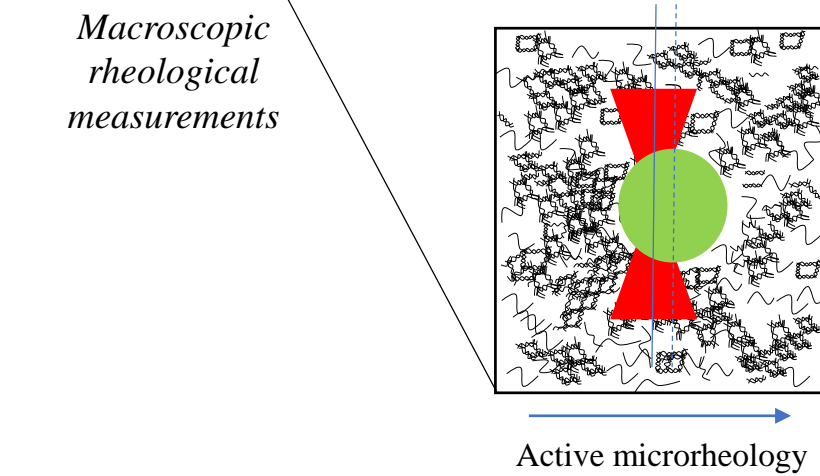
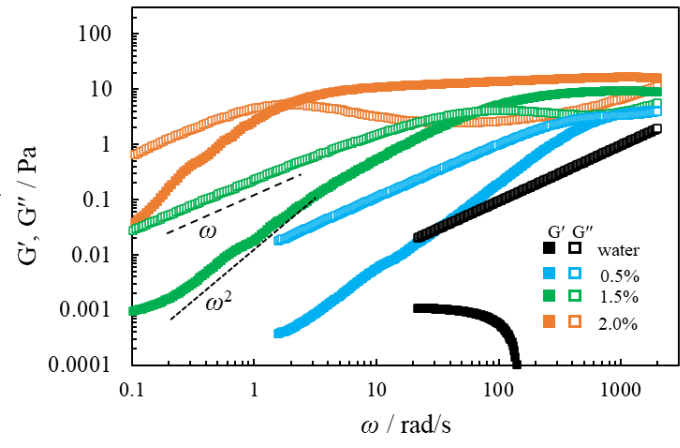
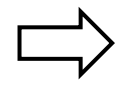
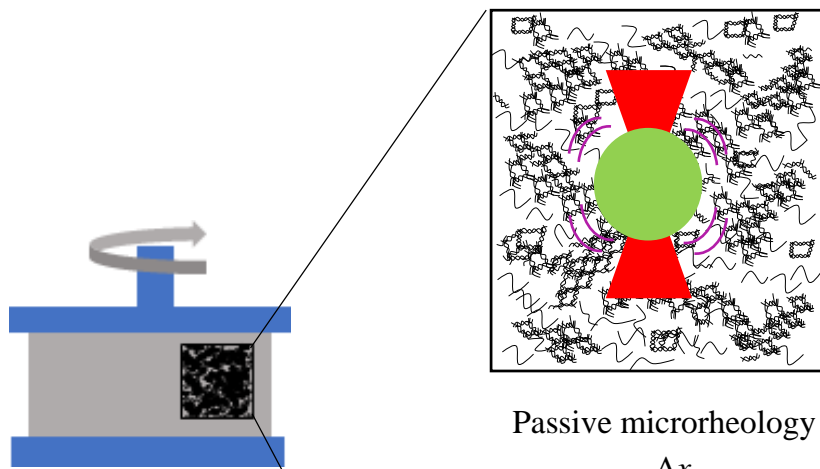
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*Macroscopic
rheological
measurements*

1 **Microrheological properties and local structure of ι-carrageenan gels**
2 **probed by using optical tweezers**

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23 **Abstract**

24 Iota carrageenan (IC) is one of the most important gelling carrageenans in food and
25 biotechnological applications; however, some of its potential applications have been
26 constrained by its weak gelation ability. In order to better understand the origin of the weak
27 mechanical response of IC gels, the hierarchical network structure based on meso- and
28 microrheological properties of IC gels at 20 °C were probed via passive and active
29 microrheology both performed by using optical tweezers (OT). Passive microrheology
30 captures a wide spectral content of IC viscoelastic properties, revealing a rubbery plateau of
31 the elastic modulus at relatively high frequencies for all IC concentrations; thus, uncovering
32 the dynamics of the network. Moreover, different microstructures of IC gels were inferred
33 by analyzing the concentration-dependent stress response when applying large deformation
34 to the network. At low IC concentration, yielding was observed as indicated by the stress-
35 independent response vs. strain, implying the structural rearrangement and disentanglement.
36 As the IC concentration increased, the yielding diminished with increasing strain rate due
37 to increased entanglement density, which limits the rearrangement of clusters. Therefore,
38 this study presents novel insights into the meso- and microscale properties of IC gels that
39 otherwise would not be accessible by conventional bulk rheology and submicroscopic probe
40 diffusion measurements.

41

42 **Keywords:** Optical tweezers-based microrheology/ carrageenan gel / mesoscale properties/
43 local large deformation

44 **1. Introduction**

45 Carrageenans are a family of water-soluble, linear sulfated polysaccharides extracted
46 from various species of red algae. It is composed of alternating units of 1,3-linked β -D-
47 galactopyranose (G) and 1,4-linked 3,6-anhydro- α -D- galactopyranose (AG) units. Owed to
48 its nontoxicity and biocompatibility, carrageenans are widely used in the food industry
49 (Mundo, Zhou, Tan, Liu, & McClements, 2021; Roesch, Cox, Compton, Happe, & Corredig,
50 2004; Simeone, Tassieri, Sibillo, & Guido, 2005); and they have the potential of being used
51 for developing products in medical and biotechnological applications (Bai & Tuvikene,
52 2021; Barabanova et al., 2008; Cao et al., 2021; Necas & Bartosikova, 2013). In industry,
53 gel-forming carrageenans are the most utilized carrageenan, including κ -carrageenan (KC)
54 and ι -carrageenan (IC). These carrageenans only differ in the degree of sulfation: one (G4S-
55 DA) for KC and two (G4S-DA2S) for IC (Azevedo, Torres, Sousa-Pinto, & Hilliou, 2015;
56 Usov, 1992), which results in a noticeable difference in the gelation properties: i.e., KC
57 forms a brittle and robust gel, whereas IC forms a soft and weak gel (Chronakis, Doubler,
58 & Piculell, 2000; Robal et al., 2017). Still, the weak gel properties of the IC gels limit their
59 potential applications as compared to the hard and brittle KC gels.

60 The gel formation and network structure of carrageenan have been associated to the
61 conformational changes of the molecules from random coil to helix structure (Rees, Steele,
62 & Williamson, 1969; Rochas, Rinaudo, & Landry, 1989). This conformational transition
63 gives rise to the viscoelastic properties of carrageenan gels, whereas the extensive
64 aggregation of KC helix structures is generally accepted to generate stiff and brittle KC gels
65 (Hu, Du, & Matsukawa, 2016; Liu, Chan, & Li, 2015; Rochas et al., 1989). However, despite

66 the small difference in the molecular structure between KC and IC, an apparent difference
67 in the physical property and gelation mechanism has been realized. Hence, the network
68 structure of IC gels has been the subject of research interest in the past decades. The physical
69 property and network structure formation of IC gels has been investigated using different
70 techniques to elucidate the underlying gelation mechanisms from different perspectives
71 (Geonzon, Descallar, Du, Bacabac, & Matsukawa, 2020). Bulk rheological properties and
72 differential scanning calorimetry (DSC) measurements revealed the gelation of IC gels and
73 suggested that IC formed only a few IC helical aggregates (Chronakis et al., 2000; Du,
74 Brenner, Xie, & Matsukawa, 2016; Takemasa, Chiba, & Date, 2001). On the other hand,
75 several studies utilized imaging techniques to infer the network structure of IC gels using
76 electron microscopy and confocal laser scanning microscopy (Amici, Clark, Normand, &
77 Johnson, 2001; Bui, Nguyen, Nicolai, & Renou, 2019; Thrimawithana, Young, Dunstan, &
78 Alany, 2010). Although the methods provide direct information on the microstructure, the
79 invasive sample treatment may alter the actual gel structure. Furthermore, polymer diffusion
80 measurements using NMR and FRAP have demonstrated the gradual formation of IC
81 aggregates, where some free polymers were not connected to the aggregates (Bui et al., 2019;
82 Zhao & Matsukawa, 2012). Moreover, submicroscopic measurements using particle
83 tracking have suggested the formation of IC aggregates or the loosely connected network of
84 IC chains (Geonzon, Bacabac, & Matsukawa, 2019b). However, the diffusion measurements
85 come with some limitations, as they rely on the thermal fluctuations of the polymers, and
86 therefore the tracer particles cannot probe the materials' response at larger length scales.
87 Thus, the network structure of weak IC gels could not be resolved. Moreover, there exists

88 a ‘size domain’ (i.e., length scale) that is often overlooked in experimental investigations of
89 IC gels, specifically at meso- and microscopic scales, which are not accessible by
90 conventional bulk and diffusion measurements. Therefore, to gain a full picture of the
91 hierarchical structure of IC gels, it is imperative to shorten further the gap between
92 macroscopic and submicroscopic investigations, as for the aim of this work by utilizing
93 techniques capable of elucidating meso- and microscopic properties.

94 The optical tweezers (OT) are extremely versatile tools used in a variety of different
95 applications within the Natural Sciences (Ashkin, Dziedzic, & Bjorkholm, 1986; Berg-
96 Sørensen & Flyvbjerg, 2004; Geonzon, Kobayashi, & Adachi, 2021; Geonzon, Kobayashi,
97 Sugimoto, & Adachi, 2022; Svoboda, Schmidt, Schnapp, & Blck, 1993; Tassieri, 2016).
98 Among the many applications, OT have been successfully used in the field of rheology
99 because of its ability to access the local mechanical properties of materials within both the
100 linear and the nonlinear regimes. Linear microrheological measurements can be performed
101 in two modes, either in passive or active microrheology (Mizuno, Head, MacKintosh, &
102 Schmidt, 2008; Shindel & Furst, 2015). The first analyses the thermally driven motion of
103 the probe particles to infer the local rheological properties of the surrounding media by
104 exploiting the generalized Stokes-Einstein relationship (Mason, 2000; Tassieri, 2016; Vyas
105 et al., 2016; Waigh, 2016). The key advantage of passive microrheology over conventional
106 bulk rheology is that it can access a wider frequency spectrum. Additionally, only a few
107 studies use OT to probe the microrheological properties of polysaccharide gels. Del Giudice
108 et al. studied rheology and nonlinear rheology of hydroxyethyl cellulose through different
109 rheological techniques, including OT, to provide new insights into the morphological

110 structure of HEC molecules. Meanwhile, Tassieri et al. measured the microrheology of
111 glucuronoxylomannan using passive microrheology by OT. Moreover, multiple particle
112 tracking and OT-based microrheology can reveal inherent heterogeneities in the network
113 structure and scale-dependent rheology, which are not accessible by macroscopic
114 measurements (Geonzon et al., 2019a; Hori, Penaloza, Shundo, & Tanaka, 2012; Maki &
115 Annaka, 2020; Van Mameren et al., 2006; Yang, Lv, Jia, Nishinari, & Fang, 2017). Notably,
116 OT-based microrheology can access the materials' nonlinear regimes by driving the probe
117 particles to relatively large deformations (Chapman & Robertson-Anderson, 2014; Falzone,
118 Blair, & Robertson-Anderson, 2015; Gomez-Solano & Bechinger, 2014; Gomez-Solano &
119 Bechinger, 2015; Robertson-Anderson, 2018). This approach can give access to the
120 materials' mechanical properties at large length scales, i.e., within the mesoscale domain,
121 and extract near macroscopic properties (Robertson-Anderson, 2018). Therefore, by using
122 OT to probe the materials' linear and nonlinear regimes, we can address the knowledge gap
123 between the macroscopic and the submicroscopic rheological perspectives of the
124 hierarchical structure of IC gels. Although this is not a direct microstructural analysis, this
125 approach also provided pivotal information on the network structure of IC gels.

126 In this study, we have adopted optical tweezers to investigate the microscopic wideband
127 viscoelastic properties of IC gels, which have revealed interesting features at relatively high
128 frequencies. Notably, the outcomes have been corroborated by conventional bulk rheology
129 measurements. Moreover, the mesoscopic properties of IC gels were investigated by means
130 of active microrheology measurements, which were achieved by entraining an optically
131 trapped particle with a flow generated by moving the microscope stage. Our collective

132 results provide new insights into the hierarchical structure of IC gels, as elucidated hereafter.

133 **2. Materials and Methods**

134 **2.1 Materials**

135 The ι- carrageenan (IC) powder was purchased from Tokyo Chemical Industry Co., Ltd.
136 (Tokyo, Japan). The IC was dialyzed against ~200 mM NaCl solution and afterward against
137 deionized water. The ion concentrations of the dialyzed samples were measured by
138 inductively coupled plasma atomic emission, wherein 0.63% for Na⁺ and 0.11% for K⁺,
139 while no Mg²⁺ or Ca²⁺ was detected. The molecular weight, M_w , of dialyzed IC samples was
140 around 360 kDa, obtained via pullulan (Hayashibara Co., Ltd., Japan) converted molecular
141 weight (see Supplementary Fig. S1) (Brenner, Tuvikene, Parker, Matsukawa, & Nishinari,
142 2014; Geonzon et al., 2019b). The stock solution of KCl was prepared by dissolving the
143 powder in deionized water (Wako Pure Chemicals Ind., Ltd.).

144 **2.2 Sample preparation**

145 The samples for bulk rheology and microrheology experiments were prepared by using
146 a simplified procedure described in the following references (Geonzon et al., 2019b;
147 Geonzon, Zhuang, Santoya, Bacabac, Xie, & Matsukawa, 2020). The IC solutions, KCl
148 solutions, deionized water (DI), and silica particles were mixed to obtain the desired
149 concentration of carrageenan with 10 mM KCl, and ~0.001% w/w probe particles (for
150 microrheology). The probe particles were non-functionalized silica with a net negative
151 charge (diameter $2a=1.0$ μm, Nippon Shokubai, Japan) and suspended in DI (Elix
152 Advantage 5, Millipore, Tokyo, Japan) (Kobayashi, Skarba, Galletto, Cakara, & Borkovec,
153 2005). The solution was then heated to 90 °C for 30 min with constant stirring. The hot

154 sample solutions were placed in a microscope chamber with double-sided tape as a spacer
155 and sealed with vacuum sealant. Afterward, the sample was quenched to 5 °C to equilibrate
156 and stored for 1-1.5 hours before measurement at 20 °C. Different carrageenan
157 concentrations were studied from 0.5 wt. %-2.0 wt.%.

158 **2.3 Bulk rheology**

159 Bulk dynamic viscoelastic properties were measured using an OHNR1 rheometer
160 (Model No.: OHNR1, OHNATECH, Japan) equipped with a concentric cylinder tool,
161 having a gap of 1.65 mm. The temperature of the sample was regulated by using a
162 temperature-controlled water bath. Hot sample solutions were loaded at ~70 °C. Prior to
163 performing the measurements at 20 °C, the sample was quenched to ~15 °C to accelerate
164 the gelation process and kept for 40-50 min at such temperature to fully equilibrate. A
165 frequency sweep from 0.6 to 125 rad/s was performed at a constant strain amplitude of
166 $\gamma=0.01$ to evaluate the frequency-dependent viscoelastic properties of the sample.

167 **2.4 Passive microrheology**

168 The microheological characterization protocols are schematically shown in Fig. 1.
169 Passive microrheology was performed using a single-laser optical tweezer (wavelength
170 $\lambda=976$ nm, OTKB/M, Thorlabs) as described in (Geonzon et al., 2021, 2022). The trapping
171 laser was focused using a 100× oil immersion objective with a high numerical aperture (NA
172 1.25, WD 0.23 mm, Nikon) and then collected by an air condenser (10×, NA 0.25, WD 7
173 mm, Nikon). The fluctuation of the trapped particle was detected using a quadrant position
174 detector (QPD) connected to the force measurement module (OTKBFM-CAL, Thorlabs).

175 The calibration factor of the voltage-to-position response function of the QPD was circa
176 0.80 V/ μm .

177 The x - and y - coordinates of the optically trapped particle were recorded at a rate of 2
178 kHz for 700-800 seconds using the force acquisition module (OTKB-Cal, Thorlabs). The
179 trap stiffness was set as low as possible (i.e., $k_p=2.05\times 10^{-6}$ N/m) to allow thermal
180 fluctuations of the probe particle. The position data were obtained by reformatting the
181 measurable ones into time-series data, which consisted of time and converted x - and y -
182 displacement expressed in micrometers by using a calibration factor (Mathematica 12.1,
183 Wolfram Research, Inc., Champaign, IL). The particle positions were then analyzed, and in
184 the case of linear viscoelastic measurements, the materials' properties were determined by
185 using the analytical method described in (Tassieri, Evans, Warren, Bailey, & Cooper, 2012).
186 Also, there is no significant changes in the thermal fluctuation of the trapped particle
187 within the observation period implying that possible heating at the laser focus has a
188 negligible effect in our study (Peterman, Gittes & Schmidt, 2003). For each sample,
189 measurements were performed for at least five different particles.

190 In general, the thermal fluctuation of an optically trapped particle is affected by the
191 response of the surrounding media and the trapping force (Mizuno et al., 2008; Tassieri,
192 2016). The thermal fluctuation of the probe particle can be analyzed by using the normalized
193 position autocorrelation function, $NPAF(\tau)$, as described in (Tassieri, 2016; Tassieri et al.,
194 2015):

$$195 \quad NPAF(\tau) \equiv \frac{\langle \vec{x}(t)\vec{x}(t+\tau) \rangle_t}{\langle x^2 \rangle_t} \quad (1)$$

196 where τ is the lag-time, $\vec{x}(t)$ is the particle position at time t , and $\langle x^2 \rangle$ is the variance of the

197 particle trajectory. For sufficiently long measurements of the particle's thermal fluctuation
 198 suspended into a Newtonian fluid with viscosity η , the $NPAF(\tau)$ can be described by a single
 199 exponential decay function (Del Giudice et al., 2017; Tassieri, 2016)

$$200 \quad NPAF(\tau) = e^{-\lambda\tau} \quad (2)$$

201 where $\lambda = k_p/\gamma$, k_p is the trap stiffness, and $\gamma = 6\pi a\eta$ is the friction coefficient (Tassieri, 2016;
 202 Tassieri et al., 2015).

203 In order to measure the linear viscoelastic properties of IC samples, we have employed
 204 the analytical method developed by (Tassieri et al., 2012), which relates the materials' shear
 205 complex modulus ($G^*(\omega)$) to the Fourier transform of the above-mentioned time-averaged
 206 fluctuation (Mason, Gang, Weitz, 1996; Pommella et al., 2013; Tassieri et al., 2010, 2015):

$$207 \quad G^*(\omega) = G'(\omega) + iG''(\omega) = \frac{\hat{A}(\omega)}{\hat{\Pi}(\omega)} \quad (3)$$

208 Where $\hat{A}(\omega)$ and $\hat{\Pi}(\omega)$ are the Fourier transforms of $NPAF(\tau)$ and $NMSD(\tau)$, respectively.

209 **2.5 Entraining flow measurement**

210 The local structure of the gels was probed by imposing a large displacement, x_{disp} , of
 211 the particle embedded in the gel network (Fig. 1c), by following the procedure in reference
 212 (Chapman & Robertson-Anderson, 2014; Khan, Regan, & Robertson, 2019; Robertson-
 213 Anderson, 2018). The position of the optical trap was kept fixed in space while the sample
 214 was moved at different speeds, v , using a 3-dimension translational stage (NanoMax 300,
 215 Thorlabs), which was controlled by custom software (LabVIEW). The applied stage
 216 movement, x_{stage} , and particle displacement relative to the optical trap centre, x_{disp} , were
 217 acquired at 10 Hz for 15 s as schematically shown in (Fig. 1). The x_{disp} was used to calculate
 218 the force exerted on the gel network, $F = k_A x_{disp}$; where the trap stiffness was set to a higher

219 value ($k_A=2.25\times 10^{-5}$ N/m) than the one used for passive microrheology measurements.
220 Before each measurement, a particle was trapped and allowed to equilibrate for 10-20
221 seconds. At least five different particles were measured for each speed at different regions
222 throughout the sample. Repeats of each particle were not performed as the application of a
223 large formation could have destroyed the gel network. The applied stage movement, x_{stage} ,
224 and the measured force, F , were converted to strain and stress by using the following
225 relationships, $\gamma=x_{stage}/2a$ and $\sigma=F/\pi a^2$ (Robertson-Anderson, 2018). The applied strain rates
226 were calculated by using the following relationship $\dot{\gamma} = 3v/\sqrt{2}a$. The details of these
227 parameters are described in the Results and Discussion section. The experiments were all
228 performed in a temperature-controlled room set at 20 °C.

229 **3. Results and Discussion**

230 **3.1 Linear viscoelastic measurements**

231 The thermal fluctuation of 1.0 μm diameter optically trapped particles embedded in the
232 IC samples were used to determine the local rheological properties of the samples. In Fig. 2
233 (top) are reported the $NPAF(\tau)$ of the optically trapped particles embedded into five
234 different samples. Whereas, in Fig. 2 (bottom) are reported the corresponding normalized
235 mean square displacement curves, $NMSD(\tau)$; which are simply related to the $NPAF(\tau)$:
236 $NMSD(\tau) = 1 - NPAF(\tau)$ (Preece et al., 2011). Notably, in the case of water, a good
237 agreement between experiments and theoretical prediction (i.e., Eq. 2) is reported. Moreover,
238 it is interestingly to notice that, at short time scales, the $NMSD(\tau)$ of the suspended particle
239 in water shows a similar behaviour as for a freely diffusing particle, with a linear increase
240 as function of the lag-time. At long lag-times, the $NMSD(\tau)$ approaches a plateau due to the

241 presence of trapping potential (Tassieri et al., 2010). Interestingly, in the case of particles
242 embedded in IC samples, the $NPAF(\tau)$ does not follow a single relaxation mode (i.e., Eq.
243 2), which is highlighted by the lower value of the slope (i.e., <1) of the $NMSD(\tau)$ curves at
244 short time scales. The decrease in the slope of $NMSD(\tau)$ at early time scale with increasing
245 IC concentration can be attributed to the significant sub-diffusive behaviour of the particles
246 governed by the increase of the viscoelastic nature of the sample. This behaviour is indeed
247 typical of a non-Newtonian fluid and here it is due to the viscoelastic response provided by
248 either the IC helices and aggregates in the solvent or the formation of the network structure
249 of IC gels.

250 In Fig. 3 are reported the elastic modulus, $G'(\omega)$, and viscous modulus, $G''(\omega)$, as a
251 function of frequency. Moreover, in order to reduce the amount of noise in the graphs, the
252 moduli shown in Fig. 3 have been obtained by transforming a smooth fitting function of the
253 experimental data shown in Fig. 2 (Tassieri et al., 2012). Nonetheless, the viscoelastic
254 moduli obtained from a direct transformation of the experimental raw data were plotted for
255 comparison, for which a good agreement can be observed. The crossover frequency is
256 shown in Table 1. Moreover, the moduli obtained from the analysis of a measurement
257 performed in water are also shown as a reference for the effective calibration of the passive
258 microrheology measurements.

259 In Fig. 3, we also report the bulk rheological properties as a means of comparison with
260 standard rheological measurements. In the case of IC samples, an apparent difference
261 between the bulk and microrheological data can be observed. It appears that microrheology
262 measurements underestimate the moduli by a factor of three to four times the bulk rheology

263 measurements. This kind of discrepancies have been observed also in other systems and has
264 been attributed to either the differences between the length scales of the techniques or to the
265 electrostatic interactions between the probe and polymer (Caggioni, Spicer, Blair,
266 Linderberg, & Weitz, 2007; Cardinaux, Cipelletti, Scheffold, & Schurtenberger, 2002;
267 Piechocka, Bacabac, Potters, Mackintosh, & Koenderink, 2010). An additional justification
268 could be given by assuming the formation of a depletion layer surrounding the probe
269 particles, given that both the probe particle and ι-carrageenan are negatively charged. This
270 would result in a slight increase in the thermal fluctuation of the probe particle (Caggioni et
271 al., 2007). Nonetheless, both techniques generally captured the behaviour of the frequency-
272 dependent viscoelastic response of IC gels, with a good agreement between the two
273 techniques in detecting the crossover frequency of the viscoelastic moduli of IC gels.

274 Moreover, from Fig. 3, the microrheological data obtained using optical tweezers show
275 a wider viscoelastic spectrum than that bulk rheology could probe (Mizuno et al., 2008;
276 Nishi, Kilfoil, Schmidt, & Mackintosh, 2018; Vyas et al., 2016; Yao et al., 2009). In
277 particular, from Fig. 3, it is interesting to notice the change in behaviour of the viscoelastic
278 moduli from IC at 1.5 wt.%, for which they show a viscous-dominated response, to the IC
279 at 2.0 wt.% sample, for which they show a more elastic-dominated response. This is also
280 corroborated by the time sweep measurements performed at 20 °C, as reported in Fig. S2 of
281 the Supplementary. Nonetheless, a previous study shows that the onset of gelation for IC is
282 expected at temperatures above 20 °C (Du et al., 2016). Therefore, we would expect the
283 formation of helical aggregates of IC in our samples. However, the viscoelastic moduli of
284 2.0 wt.% IC are relatively low in magnitude, with a significant viscous component (i.e.,

285 $G''(\omega) > G'(\omega)/10$). Hence based on the consideration on the mechanical spectra of weak
286 gels wherein the moduli are slightly frequency dependent with $G''(\omega) > G'(\omega)/10$, it
287 implies that the IC gels considered in this study behave as weak-gels at this temperature
288 (Chronakis, Picullel, & Borgström, 1996; Ikeda & Nishinari, 2001).

289 In Fig. 4, we report the complex viscosity, $|\eta^*(\omega)| = |G^*(\omega)|/\omega$, as a function of
290 angular frequency at 20 °C obtained by reorganizing the data shown in Fig. 3. In the same
291 figure, the viscosity of water is also plotted as a means of comparison. Interestingly, at low
292 frequencies, all the IC samples show a frequency-independent behavior, with the onset of
293 the Newtonian plateau that shifts towards lower frequencies with increasing carrageenan
294 concentration. The plateau value of the complex viscosity provides a measure of the zero-
295 shear viscosity, η_0 , which increases with polymer concentration due to the increase in the
296 number of entanglements within the network. At high frequencies, the drop of the $\eta^*(\omega)$ has
297 also been reported and attributed to the non-Newtonian behaviour of the IC, i.e., shear
298 thinning behaviour (Liu et al., 2015).

299 **3.2 Local strain-induced response**

300 In Fig. 5, we report the force exerted on the trapped particle during the stage movement
301 at different speeds. In order to interpret these measurements, we can recall the force response
302 for two extreme material properties, i.e., purely elastic solid and purely viscous fluid
303 (Chapman & Robertson-Anderson, 2014; Ricketts et al., 2019; Robertson-Anderson, 2018).
304 In particular, it is expected that, for a purely elastic material, the force would increase
305 linearly with applied stage movement, with the slope value providing a measure of the
306 network elastic properties. On the other hand, for a purely viscous solution, the force

307 response is expected to reach a plateau almost instantaneously, with force magnitude
308 proportional to the strain rate (Ricketts et al., 2019).

309 In Fig. 5, the force responses for both 1.5 wt.% and 2.0 wt.% IC samples show an
310 intermediate behaviour between the extreme cases, in agreement with the viscoelastic nature
311 of the materials. In particular, in the case of 1.5 wt.% IC, the force increases rapidly at short
312 times (i.e., high frequencies), and then at relatively long times (i.e., low frequencies), it
313 approaches a plateau. Therefore, the initial increase is a measure of the local elastic
314 properties of the material, whereas, at long times (i.e., by imposing large deformation), the
315 material yields and behaves like a viscous fluid. It is interesting to notice that the yielding
316 force increases with the increase of stage velocity or equivalently the strain rate. The
317 magnitude of the yielding force, F_y , provides a measure of the elasticity of the material. This
318 is confirmed by the response of the 2.0 wt.% IC, which shows a different force response
319 than the 1.5 wt.% IC. Although the 2.0 wt.% IC is considered to be a weak gel, but the fact
320 that G' is larger than G'' implies the presence of the highly entangled network. Hence, the
321 sample shows larger resistive forces and does not exhibit any yielding within the explored
322 stage velocities, thus reflecting the formation of a highly entangled network of IC gels. This
323 is further corroborated by the speed at which the force response curve of 2.0 wt.% IC returns
324 to zero when the stage stops, as the elastic component of the material would complement
325 the restoring force by the optical trap.

326 In order to better understand the materials' behavior in terms of the force response curve,
327 the measured force, F , and applied stage displacement, x_{stage} , were converted into stress, σ ,
328 and strain, γ , by means of the following two relationships: $\sigma = F/\pi a^2$ and $\gamma = x_{stage}/2a$

329 (Chapman & Robertson-Anderson, 2014; Robertson-Anderson, 2018), and reported in Fig.
330 6. For all samples, the initial increase in the σ with increasing γ can be attributed to the
331 elastic character of the material at high frequencies, which is mostly related to the stiffness
332 (i.e. $G'(\omega)$) of the network (Fig. 7a). For a 1.5 wt.% IC, yielding was observed for all strain
333 rates, demonstrating a more viscous response. Also, the yield strain roughly increases with
334 the increase in strain rate. The yielding behavior of 1.5 wt.% ICgf implies permanent
335 deformation with the applied large strains. In the linear rheological measurements (Fig. 3),
336 the 1.5 wt.% sample showed viscoelastic fluid properties suggesting the formation of a non-
337 permanent network of IC helices at these conditions of polymer concentration and
338 temperature. Furthermore, the same yielding phenomena were observed in the case of the
339 concentrated DNA and F-actin network studied using the same technique, attributed to the
340 release of entanglements in the network (Chapman & Robertson-Anderson, 2014; Falzone
341 et al., 2015). Therefore, the yielding can be attributable to the irreversible microstructure
342 rearrangement, possibly due to the disentanglement of IC helices, which are responsible for
343 the viscoelastic properties of IC.

344 On the contrary, the 2.0 wt.% IC exhibited the gel-like behavior in agreement with the
345 linear rheological measurements (see Fig. 3), suggesting a highly entangled network of IC
346 helices. Moreover, a clear dependence on the strain rates was observed for 2.0 wt.% IC.
347 While a plateau region in the stress at intermediate strain was observed at lower strain rates
348 (below 0.16 $\mu\text{m/s}$), the stress continuously increased with strain at higher strain rates. The
349 plateau value of the stress at lower rates could be attributed to the possible rearrangement
350 of the IC aggregates to accommodate the applied deformation. After the plateau, an increase

351 in stress was also demonstrated with a further increase in strain. The plateau diminished
352 with increasing the strain rate. This diminished strain could be explained as follows: as the
353 strain rate increases, the time for the microstructure rearrangement is too short. Moreover,
354 highly crosslinked microtubules demonstrated similar elastic responses. This was
355 considered to originate from the degree of crosslinking in the microtubule network (Ricketts
356 et al., 2019). Hence, it can be considered that the increase in the polymer concentration
357 contributes to more entanglement that enhances the elastic properties and restrains possible
358 rearrangement of the network with large deformation.

359 The rheological parameters were calculated from the σ and γ curves as a function of $\dot{\gamma}$,
360 as shown in Fig. 7. Fig. 7a presents the initial slope of σ and γ , $(d\sigma/d\gamma)_{\gamma \rightarrow 0}$ as a function of
361 $\dot{\gamma}$, calculated at the initial increase of σ . As mentioned earlier, the $d\sigma/d\gamma$ is comparable to the
362 stiffness of the network or differential modulus, K (Robertson-Anderson, 2018). For a 2.0
363 wt.% IC, the $d\sigma/d\gamma$ is almost independent of $\dot{\gamma}$ showing a gel-like behavior. On the contrary,
364 a clear dependence of $d\sigma/d\gamma$ on $\dot{\gamma}$ was seen for 1.5 wt.% IC. Furthermore, the $d\sigma/d\gamma$ showed
365 a tendency to approach a constant value with increasing $\dot{\gamma}$. The same behaviour was also
366 seen for F-actin gels using the same methodology (Falzone et al., 2015). We consider that
367 the dependence of $d\sigma/d\gamma$ on $\dot{\gamma}$ can be attributed to the network properties and rearrangement
368 of IC helices with the applied local deformation. By giving enough time at low strain rates,
369 the possible network rearrangement contributes to stress dissipation. Moreover, with
370 increasing strain rates, the time required for network rearrangement may not be enough,
371 contributing to the elastic response of the material, which originates from the entangled
372 network. Fig. 7b presents the average yield stress, σ_y against $\dot{\gamma}$. From the σ_y , the apparent

373 viscosity, η , was calculated following the Stokes relationship based on the magnitude of the
374 yielding force, F_y , as shown in the following equation: $\eta = F_y / 6\pi a v$, and considering $\sigma_y = F_y$
375 $/\pi a^2$ (Robertson-Anderson, 2018). The apparent viscosity, η against $\dot{\gamma}$ are shown in Fig. 7c.
376 For all concentrations, the η decreases with strain rate indicating a shear thinning behaviour
377 of IC in agreement with the macroscopic measurements (Chronakis et al., 2000). The shear
378 thinning exponents is higher than -1.0, which is an apparent limit for well-entangled
379 polymer solutions (Teixeira et al., 2007). Moreover, the $\eta^*(\omega)$ obtained using passive
380 microrheology (Fig. 4) exhibited shear thinning behaviour at high frequencies with
381 exponent ≈ -1.0 and a plateau at low frequencies. Despite the difference in the methodologies,
382 the viscosities obtained from both passive and large deformation microrheology are within
383 the same orders of magnitude. Furthermore, it has been found that dilute polymer solutions
384 demonstrate shear thinning exponent even up to circa -0.5. This suggests that our samples
385 can be considered within the entangled regime, and the material properties comes from the
386 entanglement of rigid IC helices (Ikeda & Nishinari, 2001; Morris et al., 1999).

387 **3.3 Proposed structure developed from different investigations**

388 Based on the above discussion and previous results from different investigations, the
389 network of IC gels is schematically drawn in Fig. 8. Previous study on the gelation
390 mechanism and network structure of IC gels at submicroscopic viewpoints suggest the
391 formation of clusters of IC aggregates (Geonzon *et al.*, 2019b). This submicroscopic
392 investigation supports the macroscopic measurements using bulk rheology of IC gels (Du
393 *et al.*, 2016; Hu *et al.*, 2016; Takemasa *et al.*, 2001). The microrheological results presented
394 in this study, both in linear and nonlinear regimes, corroborate the macroscopic and

395 submicroscopic results. Moreover, the wide spectral content captured by using passive
396 microrheology suggests that the viscoelastic properties of IC originates from the
397 entanglement of rigid IC helices. This is further confirmed by the large deformation
398 measurements, which infer that the yielding at low IC concentration comes from the
399 rearrangement of clusters and disentanglement of rigid IC helices. Increasing the IC
400 concentration and entanglement density suppresses the yielding and produces a gel-like
401 behaviour, which may suggest that the systems may be composed of non-percolating
402 clusters of IC chain aggregates. These aggregated IC chains contribute to IC gels'
403 rheological properties, even without forming permanent networks and allow particle
404 diffusion, as seen in the diffusion measurements. Thus, we infer the schematic illustration
405 shown in Fig. 8 as heterogenous clusters of IC helices developed from different perspectives.

406 **4. Conclusion**

407 We have performed optical tweezers-based microrheology to gain new insights into the
408 hierarchical network structure of IC gels. Two microrheological techniques were employed
409 to probe the network structure: passive microrheology and local large deformation
410 measurements. Comparison between bulk and passive microrheology showed slight
411 differences attributed to the difference in detecting length scale, but captured the behaviour
412 of the materials' frequency-dependent viscoelastic properties. By applying large
413 deformation to the local network structure of IC gels, we gained further information on the
414 hierarchical structure of IC gels. In particular, we have found a concentration-dependent
415 nonlinear responses of the IC gels at the mesoscopic scales. At concentrations <2.0 wt.%,

416 IC samples yield with the application of large deformation because of the rearrangement of
417 clusters and disentanglement of the IC helices. In contrast, at 2.0 wt.%, the high
418 entanglement density suppresses the rearrangement of clusters, and the sample exhibits a
419 gel-like behaviour. To the best of our knowledge, this is the first time to explore the
420 mesoscopic properties of the IC gels.

421 **Author Contributions**

422 **Lester Geonzon:** Conceptualization, Methodology, Software, Visualization, Formal
423 Analysis, Investigation, Writing-Original Draft, Writing - Review & Editing. **Motoyoshi**
424 **Kobayashi:** Supervision; Conceptualization; Methodology; Formal analysis; Validation;
425 Visualization; Writing-original draft; Writing-review & editing; Resources; Funding
426 acquisition; Project administration. **Manlio Tassieri:** Formal analysis; Software; Writing-
427 review & editing. **Rommel Bacabac:** Formal Analysis; Writing - Review & Editing.
428 **Yasuhisa Adachi:** Funding acquisition; Project administration. **Shingo Matsukawa:**
429 Formal Analysis; Resources; Writing - Review & Editing.

430 **Conflicts of interest**

431 There are no conflicts of interest to declare.

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438 **References**

- 439 Amici, E., Clark, A. H., Normand, V., & Johnson, N. B. (2001). Interpenetrating network
440 formation in agarose-sodium gellan gel composites. *Carbohydrate Polymers*, 46(4),
441 383–391. [https://doi.org/10.1016/S0144-8617\(00\)00342-8](https://doi.org/10.1016/S0144-8617(00)00342-8)
- 442 Ashkin, A., Dziedzic, D., Bjorkholm, J. E., & Chu, S. (1986). Observation of a single-beam
443 gradient force optical trap for dielectric particles. *Optics Letters*, 11(5), 288–290.
- 444 Azevedo, G., Torres, M. D., Sousa-Pinto, I., & Hilliou, L. (2015). Effect of pre-extraction
445 alkali treatment on the chemical structure and gelling properties of extracted hybrid
446 carrageenan from *Chondrus crispus* and *Ahnfeltiopsis devoniensis*. *Food*
447 *Hydrocolloids*, 50, 150–158. <https://doi.org/10.1016/j.foodhyd.2015.03.029>
- 448 Bai, R. G., & Tuvikene, R. (2021). Potential antiviral properties of industrially important
449 marine algal polysaccharides and their significance in fighting a future viral pandemic.
450 *Viruses*, 13(9). <https://doi.org/10.3390/v13091817>
- 451 Barabanova, A. O., Shashkov, A. S., Glazunov, V. P., Isakov, V. V., Nebylovskaya, T. B.,
452 Helbert, W., Solov'eva, T. F., & Yermak, I. M. (2008). Structure and properties of
453 carrageenan-like polysaccharide from the red alga *Tichocarpus crinitus* (Gmel.) Rupr.
454 (Rhodophyta, Tichocarpaceae). *Journal of Applied Phycology*, 20(6), 1013–1020.
455 <https://doi.org/10.1007/s10811-007-9295-z>
- 456 Berg-Sørensen, K., & Flyvbjerg, H. (2004). Power spectrum analysis for optical tweezers.
457 *Review of Scientific Instruments*, 75(3), 594–612. <https://doi.org/10.1063/1.1645654>
- 458 Brenner, T., Tuvikene, R., Parker, A., Matsukawa, S., & Nishinari, K. (2014). Rheology and

459 structure of mixed kappa-carrageenan/iota-carrageenan gels. *Food Hydrocolloids*, 39,
460 272–279. <https://doi.org/10.1016/j.foodhyd.2014.01.024>

461 Bui, V. T. N. T., Nguyen, B. T., Nicolai, T., & Renou, F. (2019). Mobility of carrageenan
462 chains in iota- and kappa carrageenan gels. *Colloids and Surfaces A: Physicochemical
463 and Engineering Aspects*, 562(October 2018), 113–118.
464 <https://doi.org/10.1016/j.colsurfa.2018.11.017>

465 Caggioni, M., Spicer, P. T., Blair, D. L., Lindberg, S. E., & Weitz, D. A. (2007). Rheology
466 and microrheology of a microstructured fluid: The gellan gum case. *Journal of
467 Rheology*, 51(5), 851–865. <https://doi.org/10.1122/1.2751385>

468 Cao, W., Jin, J., Wu, G., Bravenboer, N., Helder, M. N., Pathak, J. L., Zandieh-Doulabi, B.,
469 Hogervorst, J. M. A., Matsukawa, S., Geonzon, L. C., Bacabac, R. G., Schulten, E. A.
470 J. M., & Klein-Nulend, J. (2021). K-carrageenan stimulates pre-osteoblast proliferation
471 and osteogenic differentiation: A potential factor for the promotion of bone
472 regeneration? *Molecules*, 26(20). <https://doi.org/10.3390/molecules26206131>

473 Cardinaux, F., Cipelletti, L., Scheffold, F., & Schurtenberger, P. (2002). Microrheology of
474 giant-micelle solutions. *Europhysics Letters*, 57(5), 738–744.
475 <https://doi.org/10.1209/epl/i2002-00525-0>

476 Chapman, C. D., & Robertson-Anderson, R. M. (2014). Nonlinear microrheology reveals
477 entanglement-driven molecular-level viscoelasticity of concentrated DNA. *Physical
478 Review Letters*, 113(9), 1–5. <https://doi.org/10.1103/PhysRevLett.113.098303>

479 Chronakis, I. S., Doublier, J. L., & Piculell, L. (2000). Viscoelastic properties for kappa-
480 and iota-carrageenan in aqueous NaI from the liquid-like to the solid-like behaviour.

481 *International Journal of Biological Macromolecules*, 28(1), 1–14.
482 [https://doi.org/10.1016/S0141-8130\(00\)00141-0](https://doi.org/10.1016/S0141-8130(00)00141-0)

483 Chronakis, I. S., Piculell, L., & Borgström, J. (1996). Rheology of kappa-carrageenan in
484 mixtures of sodium and cesium iodide: Two types of gels. *Carbohydrate Polymers*,
485 31(4), 215–225. [https://doi.org/10.1016/S0144-8617\(96\)00117-8](https://doi.org/10.1016/S0144-8617(96)00117-8)

486 Del Giudice, F., Tassieri, M., Oelschlaeger, C., & Shen, A. Q. (2017). When Microrheology,
487 Bulk Rheology, and Microfluidics Meet: Broadband Rheology of Hydroxyethyl
488 Cellulose Water Solutions. *Macromolecules*, 50(7), 2951–2963.
489 <https://doi.org/10.1021/acs.macromol.6b02727>

490 Du, L., Brenner, T., Xie, J., & Matsukawa, S. (2016). A study on phase separation behavior
491 in kappa/iota carrageenan mixtures by micro DSC, rheological measurements and
492 simulating water and cations migration between phases. *Food Hydrocolloids*, 55(July
493 2018), 81–88. <https://doi.org/10.1016/j.foodhyd.2015.11.004>

494 Falzone, T. T., Blair, S., & Robertson-Anderson, R. M. (2015). Entangled F-actin displays
495 a unique crossover to microscale nonlinearity dominated by entanglement segment
496 dynamics. *Soft Matter*, 11(22), 4418–4423. <https://doi.org/10.1039/c5sm00155b>

497 Geonzon, L. C., Bacabac, R. G., & Matsukawa, S. (2019a). Microscopic Characterization
498 of Phase Separation in Mixed Carrageenan Gels Using Particle Tracking. *Journal of*
499 *The Electrochemical Society*, 166(9), B3228–B3234.
500 <https://doi.org/10.1149/2.0351909jes>

501 Geonzon, L. C., Bacabac, R. G., & Matsukawa, S. (2019b). Network structure and gelation
502 mechanism of kappa and iota carrageenan elucidated by multiple particle tracking.

503 *Food Hydrocolloids*, 92, 173–180. <https://doi.org/10.1016/j.foodhyd.2019.01.062>

504 Geonzon, L. C., Descallar, F. B. A., Du, L., Bacabac, R. G., & Matsukawa, S. (2020).
505 Gelation mechanism and network structure in gels of carrageenans and their mixtures
506 viewed at different length scales – A review. *Food Hydrocolloids*, 108, 106039.
507 <https://doi.org/10.1016/j.foodhyd.2020.106039>

508 Geonzon, L. C., Kobayashi, M., & Adachi, Y. (2021). Effect of shear flow on the
509 hydrodynamic drag force of a spherical particle near a wall evaluated using optical
510 tweezers and microfluidics. *Soft Matter*, 17(34), 7914–7920.
511 <https://doi.org/10.1039/d1sm00876e>

512 Geonzon, L. C., Kobayashi, M., Sugimoto, T., & Adachi, Y. (2022). Study on the Kinetics
513 of Adsorption of Poly(ethylene oxide) Onto A Silica Particle Using Optical Tweezers
514 and Microfluidics. *Colloids and Surfaces A: Physicochemical and Engineering Aspects*,
515 642(February), 128691. <https://doi.org/10.1016/j.colsurfa.2022.128691>

516 Geonzon, L. C., Zhuang, X., Santoya, A. M., Bacabac, R. G., Xie, J., & Matsukawa, S.
517 (2020). Gelation mechanism and network structure of mixed kappa
518 carrageenan/lambda carrageenan gels studied by macroscopic and microscopic
519 observation methods. *Food Hydrocolloids*, 105, 105759.
520 <https://doi.org/10.1016/j.foodhyd.2020.105759>

521 Gomez-Solano, J. R., & Bechinger, C. (2014). Probing linear and nonlinear microrheology
522 of viscoelastic fluids. *Epl*, 108(5). <https://doi.org/10.1209/0295-5075/108/54008>

523 Gomez-Solano, Juan Ruben, & Bechinger, C. (2015). Transient dynamics of a colloidal
524 particle driven through a viscoelastic fluid. *New Journal of Physics*, 17(10).

525 <https://doi.org/10.1088/1367-2630/17/10/103032>

526 Hori, K., Penaloza, D. P., Shundo, A., & Tanaka, K. (2012). Time-dependent heterogeneity
527 in viscoelastic properties of worm-like micelle solutions. *Soft Matter*, 8(28), 7361–
528 7364. <https://doi.org/10.1039/c2sm25549a>

529 Hu, B., Du, L., & Matsukawa, S. (2016). NMR study on the network structure of a mixed
530 gel of kappa and iota carrageenans. *Carbohydrate Polymers*, 150, 57–64.
531 <https://doi.org/10.1016/j.carbpol.2016.04.112>

532 Ikeda, S., & Nishinari, K. (2001). “ *Weak Gel* ” -Type Rheological Properties of Aqueous
533 *Dispersions of Nonaggregated K -Carrageenan Helices*. 4436–4441.

534 Khan, M., Regan, K., & Robertson-Anderson, R. M. (2019). Optical Tweezers
535 Microrheology Maps the Dynamics of Strain-Induced Local Inhomogeneities in
536 Entangled Polymers. *Physical Review Letters*, 123(3), 38001.
537 <https://doi.org/10.1103/PhysRevLett.123.038001>

538 Kobayashi, M., Skarba, M., Galletto, P., Cakara, D., & Borkovec, M. (2005). Effects of heat
539 treatment on the aggregation and charging of Stöber-type silica. *Journal of Colloid and*
540 *Interface Science*, 292(1), 139–147. <https://doi.org/10.1016/j.jcis.2005.05.093>

541 Liu, S., Chan, W. L., & Li, L. (2015). Rheological Properties and Scaling Laws of κ -
542 Carrageenan in Aqueous Solution. *Macromolecules*, 48(20), 7649–7657.
543 <https://doi.org/10.1021/acs.macromol.5b01922>

544 Maki, Y., & Annaka, M. (2020). Gelation of fish gelatin studied by multi-particle tracking
545 method. *Food Hydrocolloids*, 101(November 2019), 105525.
546 <https://doi.org/10.1016/j.foodhyd.2019.105525>

547 Mason, T. G. (2000). Estimating the viscoelastic moduli of complex fluids using the
548 generalized Stokes-Einstein equation. *Rheologica Acta*, 39(4), 371–378.
549 <https://doi.org/10.1007/s003970000094>

550 Mason, T. G., Gang, H., & Weitz, D. A. (1996). Rheology of complex fluids measured by
551 dynamic light scattering. *Journal of Molecular Structure*, 383(1–3), 81–90.
552 [https://doi.org/10.1016/S0022-2860\(96\)09272-1](https://doi.org/10.1016/S0022-2860(96)09272-1)

553 Mizuno, D., Head, D. A., MacKintosh, F. C., & Schmidt, C. F. (2008). Active and passive
554 microrheology in equilibrium and nonequilibrium systems. *Macromolecules*, 41(19),
555 7194–7202. <https://doi.org/10.1021/ma801218z>

556 Morris, V. J., Kirby, A. R., & Gunning, A. P. (1999). A fibrous model for gellan gels from
557 atomic force microscopy studies. *Progress in Colloid and Polymer Science*, 114, 102–
558 108. https://doi.org/10.1007/3-540-48349-7_15

559 Mundo, J. L. M., Zhou, H., Tan, Y., Liu, J., & McClements, D. J. (2021). Enhancing
560 emulsion functionality using multilayer technology: Coating lipid droplets with
561 saponin-polypeptide-polysaccharide layers by electrostatic deposition. *Food Research*
562 *International*, 140(April), 109864. <https://doi.org/10.1016/j.foodres.2020.109864>

563 Necas, J., & Bartosikova, L. (2013). Carrageenan: A review. *Veterinarni Medicina*, 58(4),
564 187–205. <https://doi.org/10.17221/6758-VETMED>

565 Nishi, K., Kilfoil, M. L., Schmidt, C. F., & Mackintosh, F. C. (2018). A symmetrical method
566 to obtain shear moduli from microrheology. *Soft Matter*, 14(19), 3716–3723.
567 <https://doi.org/10.1039/c7sm02499a>

568 Peterman, E. J. G., Gittes, F., & Schmidt, C. F. (2003). Laser-induced heating in optical traps.

569 *Biophysical Journal*, 84(2 I), 1308–1316. <https://doi.org/10.1016/S0006->
570 3495(03)74946-7

571 Piechocka, I. K., Bacabac, R. G., Potters, M., Mackintosh, F. C., & Koenderink, G. H. (2010).
572 Structural hierarchy governs fibrin gel mechanics. *Biophysical Journal*, 98(10), 2281–
573 2289. <https://doi.org/10.1016/j.bpj.2010.01.040>

574 Pommella, A., Preziosi, V., Caserta, S., Cooper, J. M., Guido, S., & Tassieri, M. (2013).
575 Using optical tweezers for the characterization of polyelectrolyte solutions with very
576 low viscoelasticity. *Langmuir*, 29(29), 9224–9230. <https://doi.org/10.1021/la4015948>

577 Preece, D., Warren, R., Evans, R. M. L., Gibson, G. M., Padgett, M. J., Cooper, J. M., &
578 Tassieri, M. (2011). Optical tweezers: Wideband microrheology. *Journal of Optics*,
579 13(4). <https://doi.org/10.1088/2040-8978/13/4/044022>

580 Rees, D. A., Steele, I. W., & Williamson, F. B. (1969). Conformational analysis of
581 polysaccharides. III. The relation between stereochemistry and properties of some
582 natural polysaccharide sulfates (1). *Journal of Polymer Science Part C: Polymer*
583 *Symposia*, 28(1), 261–276. <https://doi.org/10.1002/polc.5070280121>

584 Ricketts, S. N., Francis, M. L., Farhadi, L., Rust, M. J., Das, M., Ross, J. L., & Robertson-
585 Anderson, R. M. (2019). Varying crosslinking motifs drive the mesoscale mechanics
586 of actin-microtubule composites. *Scientific Reports*, 9(1), 1–12.
587 <https://doi.org/10.1038/s41598-019-49236-4>

588 Robal, M., Brenner, T., Matsukawa, S., Ogawa, H., Truus, K., Rudolph, B., & Tuvikene, R.
589 (2017). Monocationic salts of carrageenans: Preparation and physico-chemical
590 properties. *Food Hydrocolloids*, 63, 656–667.

591 <https://doi.org/10.1016/j.foodhyd.2016.09.032>

592 Robertson-Anderson, R. M. (2018). Optical Tweezers Microrheology: From the Basics to
593 Advanced Techniques and Applications. *ACS Macro Letters*, 7(8), 968–975.
594 <https://doi.org/10.1021/acsmacrolett.8b00498>

595 Rochas, C., Rinaudo, M., & Landry, S. (1989). Relation between the molecular structure
596 and mechanical properties of carrageenan gels. *Carbohydrate Polymers*, 10(2), 115–
597 127. [https://doi.org/10.1016/0144-8617\(89\)90061-1](https://doi.org/10.1016/0144-8617(89)90061-1)

598 Roesch, R., Cox, S., Compton, S., Happek, U., & Corredig, M. (2004). κ -Carrageenan and
599 β -lactoglobulin interactions visualized by atomic force microscopy. *Food*
600 *Hydrocolloids*, 18(3), 429–439. <https://doi.org/10.1016/j.foodhyd.2003.08.001>

601 Shindel, M. M., & Furst, E. M. (2015). Frequency modulated microrheology. *Lab Chip*,
602 15(11), 2460–2466. <https://doi.org/10.1039/C5LC00351B>

603 Simeone, M., Tassieri, M., Sibillo, V., & Guido, S. (2005). Effect of sol-gel transition on
604 shear-induced drop deformation in aqueous mixtures of gellan and κ -carrageenan.
605 *Journal of Colloid and Interface Science*, 281(2), 488–494.
606 <https://doi.org/10.1016/j.jcis.2004.08.095>

607 Svoboda, K., Schmidt, C. F., Schnapp, B. J., & Block, B. S. M. (1993). Direct observation
608 of kinesin stepping by OT interferometry. *Nature*, 365(October), 721–727.

609 Takemasa, M., Chiba, A., & Date, M. (2001). Gelation mechanism of κ - and ι -carrageenan
610 investigated by correlation between the strain-optical coefficient and the dynamic shear
611 modulus. *Macromolecules*, 34(21), 7427–7434. <https://doi.org/10.1021/ma0102924>

612 Tassieri, M. (2016). *Microrheology with Optical Tweezers: Principles and Applications*. Pan

613 Stanford Publishing Pte. Ltd.

614 Tassieri, M., Del Giudice, F., Robertson, E. J., Jain, N., Fries, B., Wilson, R., Glidle, A.,
615 Greco, F., Netti, P. A., Maffettone, P. L., Bicanic, T., & Cooper, J. M. (2015).
616 Microrheology with optical tweezers: Measuring the relative viscosity of solutions “at
617 a glance.” *Scientific Reports*, 5, 1–6. <https://doi.org/10.1038/srep08831>

618 Tassieri, M., Evans, R. M. L., Warren, R. L., Bailey, N. J., & Cooper, J. M. (2012).
619 Microrheology with optical tweezers: Data analysis. *New Journal of Physics*, 14.
620 <https://doi.org/10.1088/1367-2630/14/11/115032>

621 Tassieri, M., Gibson, G. M., Evans, R. M. L., Yao, A. M., Warren, R., Padgett, M. J., &
622 Cooper, J. M. (2010). Measuring storage and loss moduli using optical tweezers:
623 Broadband microrheology. *Physical Review E - Statistical, Nonlinear, and Soft Matter*
624 *Physics*, 81(2), 1–5. <https://doi.org/10.1103/PhysRevE.81.026308>

625 Teixeira, R. E., Dambal, A. K., Richter, D. H., Shaqfeh, E. S. G., & Chu, S. (2007). The
626 individualistic dynamics of entangled DNA in solution. *Macromolecules*, 40(7), 2461–
627 2476. <https://doi.org/10.1021/ma062932e>

628 Thrimawithana, T. R., Young, S., Dunstan, D. E., & Alany, R. G. (2010). Texture and
629 rheological characterization of kappa and iota carrageenan in the presence of counter
630 ions. *Carbohydrate Polymers*, 82(1), 69–77.
631 <https://doi.org/10.1016/j.carbpol.2010.04.024>

632 Usov, A. I. (1992). Sulfated polysaccharides of the red seaweeds. *Topics in Catalysis*, 6(1),
633 9–23. [https://doi.org/10.1016/S0268-005X\(09\)80055-6](https://doi.org/10.1016/S0268-005X(09)80055-6)

634 Van Mameren, J., Modesti, M., Kanaar, R., Wyman, C., Wuite, G. J. L., & Peterman, E. J.

635 G. (2006). Dissecting elastic heterogeneity along DNA molecules coated partly with
636 Rad51 using concurrent fluorescence microscopy and optical tweezers. *Biophysical*
637 *Journal*, 91(8), L78–L80. <https://doi.org/10.1529/biophysj.106.089466>

638 Vyas, B. M., Orpe, A. V., Kaushal, M., Joshi, Y. M., Joshi, Y. M., Cicuta, P., Donald, A. M.,
639 Squires, T. M., Mason, T. G., Puertas, A. M., Voigtmann, T., Schultz, K. M., Furst, E.
640 M., Abdala, A. A., Amin, S., Zanten, J. H. van, Khan, S. A., Moschakis, T., Fielding, S.
641 M., ... Sasso, A. (2016). Passive microrheology in the effective time domain: analyzing
642 time dependent colloidal dispersions. *Soft Matter*, 5, 181–202.
643 <https://doi.org/10.1039/C6SM00829A>

644 Waigh, T. A. (2016). Advances in the microrheology of complex fluids. *Reports on Progress*
645 *in Physics*, 074601, 1–101. <https://doi.org/10.1088/0034-4885/79/7/074601>

646 Yang, N., Lv, R., Jia, J., Nishinari, K., & Fang, Y. (2017). Application of Microrheology in
647 Food Science. *Annual Review of Food Science and Technology*, 8(January), 493–521.
648 <https://doi.org/10.1146/annurev-food-030216-025859>

649 Yao, A., Tassieri, M., Padgett, M., & Cooper, J. (2009). Microrheology with optical tweezers.
650 *Lab on a Chip*, 9(17), 2568–2575. <https://doi.org/10.1039/b907992k>

651 Zhao, Q., & Matsukawa, S. (2012). Estimation of the hydrodynamic screening length in -
652 carrageenan solutions using NMR diffusion measurements. *Polymer Journal*, 44(8),
653 901–906. <https://doi.org/10.1038/pj.2012.107>

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Table 1. The crossover frequency of G' and G'' obtained using both techniques

	0.5 wt. %	1.0 wt. %	1.5 wt. %		2.0 wt. %	
			Bulk	OT	Bulk	OT
ω /rad/s	875.3	257.5	125.7	71.6	2.0	1.9

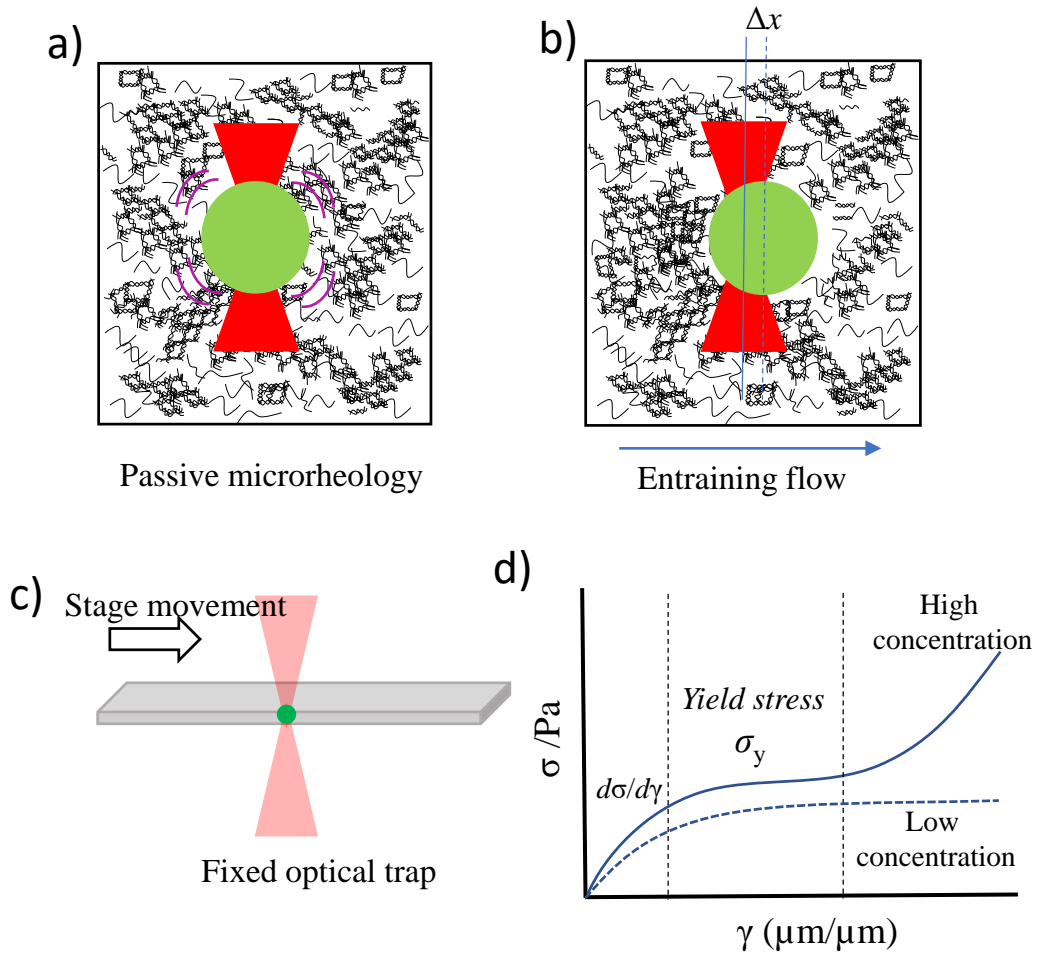


Fig. 1. Protocols for the microrheological characterization of the ι -carrageenan gels using optical tweezers. a) Passive microrheology and b) entraining flow for large deformation measurement. c) For the entraining flow, the sample stage is moved at different rates over a distance x and the force F the sample exerts on the trapped particle is measured. d) Schematic illustration of the stress, σ , and strain, γ , curve. **The green balls in Figs. 1 a-c represent the probe particle. Meanwhile, the black curves/lines in Figs. 1 a-b represent the IC polymers and the aggregates. The purple curves in Fig. 1a represent the fluctuation of the trapped particle for the passive microrheology measurements. Finally, the red trapezoids in Figs. 1 a-c represent the trapping laser that is highly focused after passing through the objective lens.**

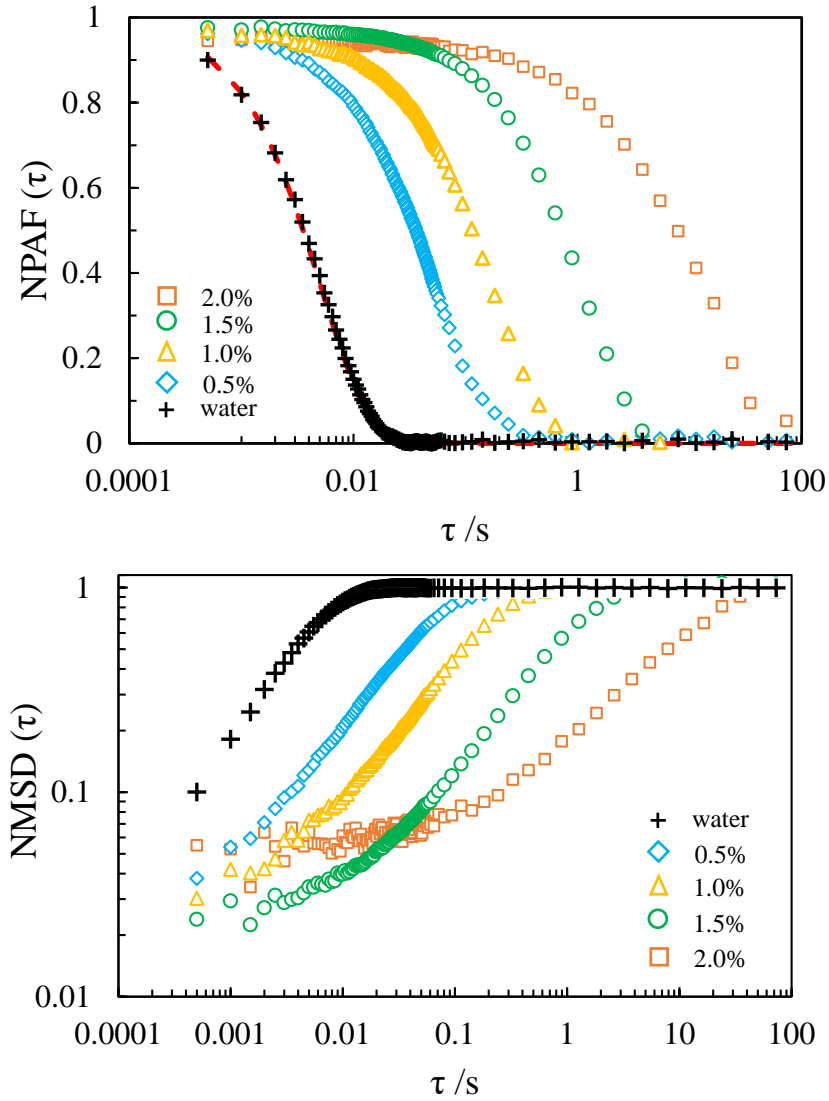


Fig. 2. (Top) Normalized position autocorrelation function, $NPAF(\tau)$, and (bottom) normalized mean square displacement, $NMSD(\tau)$, plots of an optically trapped particle ($2a = 1 \mu\text{m}$) in IC gels measured at 2 kHz for a period of 700-800 s at laser current of 50 mA ($k \sim 2.05 \times 10^{-6} \text{ N/m}$). Water data was only measured for 300 s.

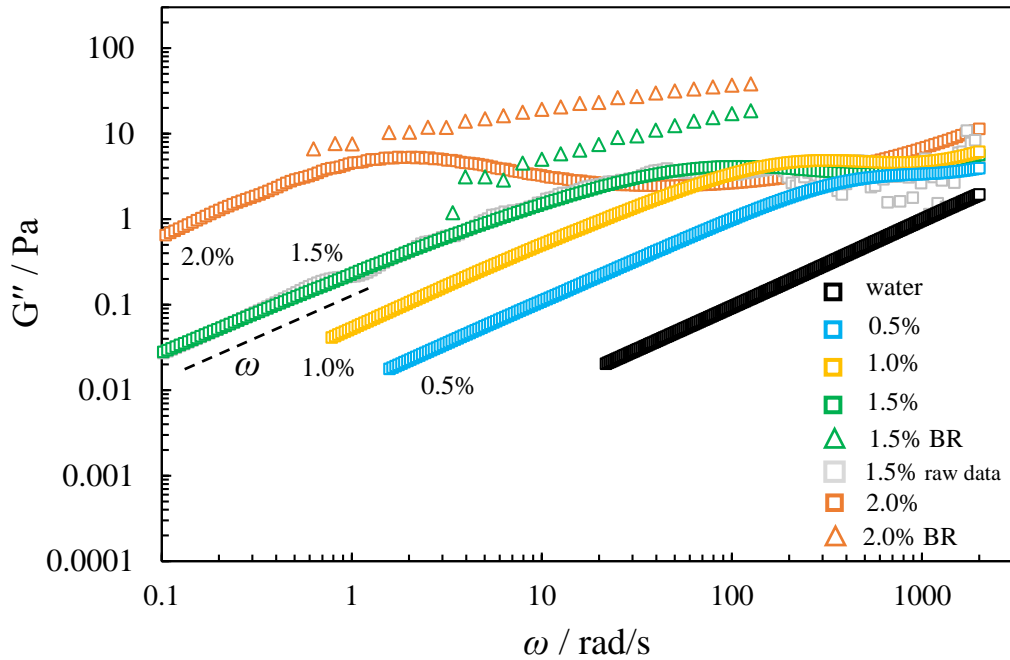
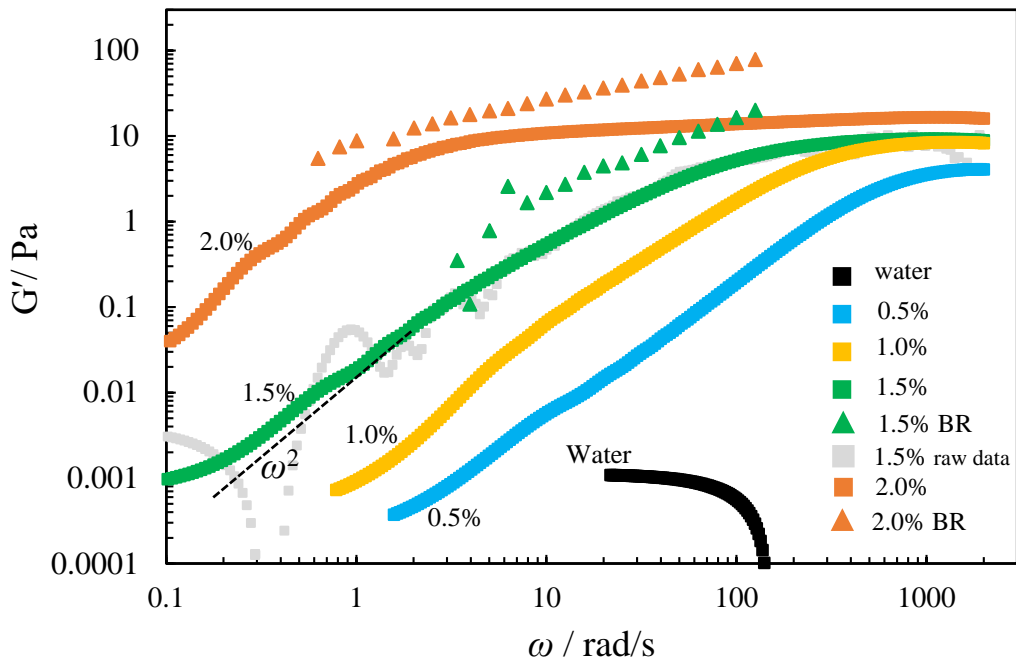


Fig. 3. Plots of viscoelastic moduli (top) G' (filled) and (bottom) G'' (open) as a function of frequency obtained using passive microrheology (square) and bulk rheology (BR) (triangle) measurement. The microrheology data using OT were obtained using the fitting function from the raw experimental data (i.e., 1.5% raw data).

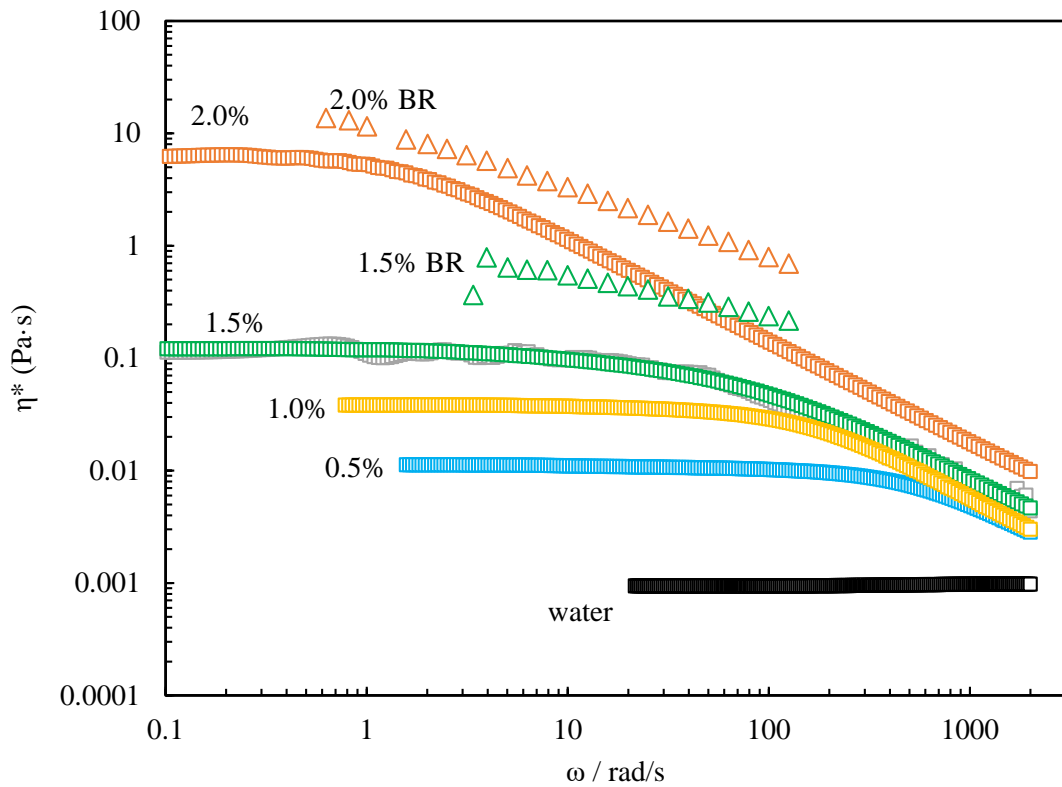


Fig. 4. Plots of complex viscosity as a function of frequency for different IC concentrations obtained by using passive microrheology (square) and bulk rheology (BR) (triangle) . The data were obtained by using the fitting function of the raw experimental data (i.e., black line for the case 1.5%).

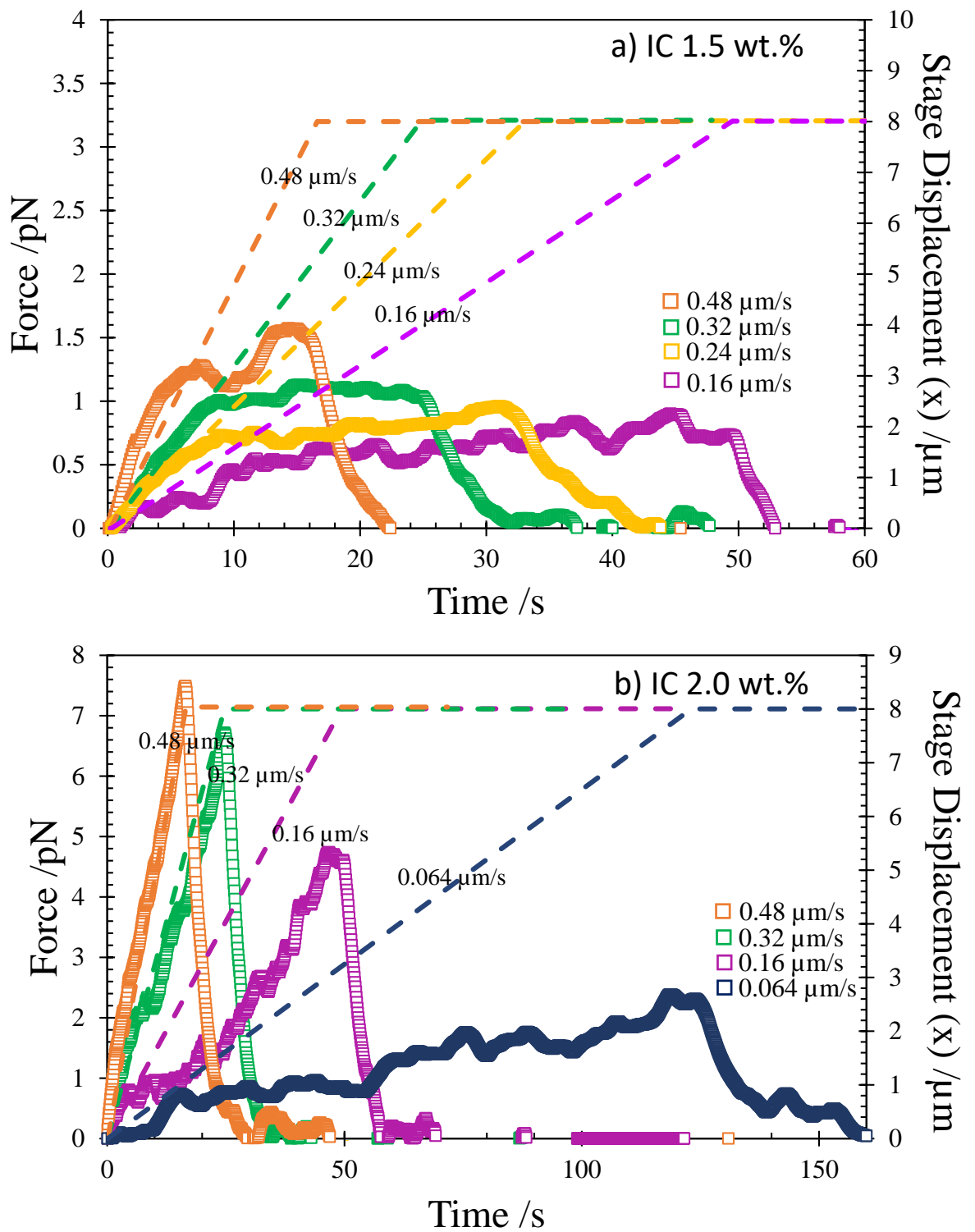


Fig. 5. Mesoscale rheology of IC gels at different strain rates using OT. The measured force response (symbols) as a function of stage movement at different speeds. The dashed line represent the stage displacement.

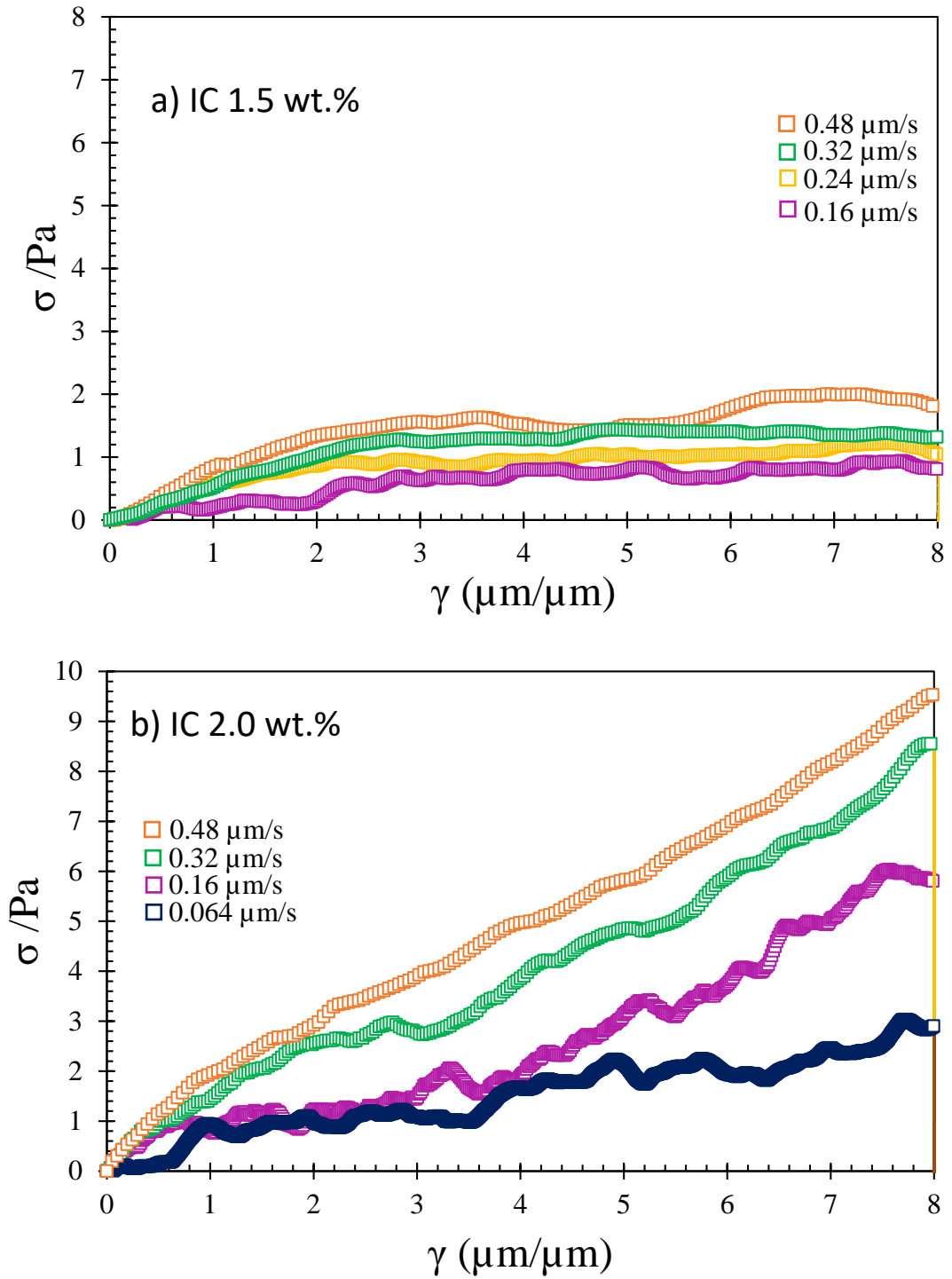


Fig. 6. Plots of σ as function γ , where $\sigma=F/\pi R^2$ and $\gamma=x/2R$, for different stage velocities.

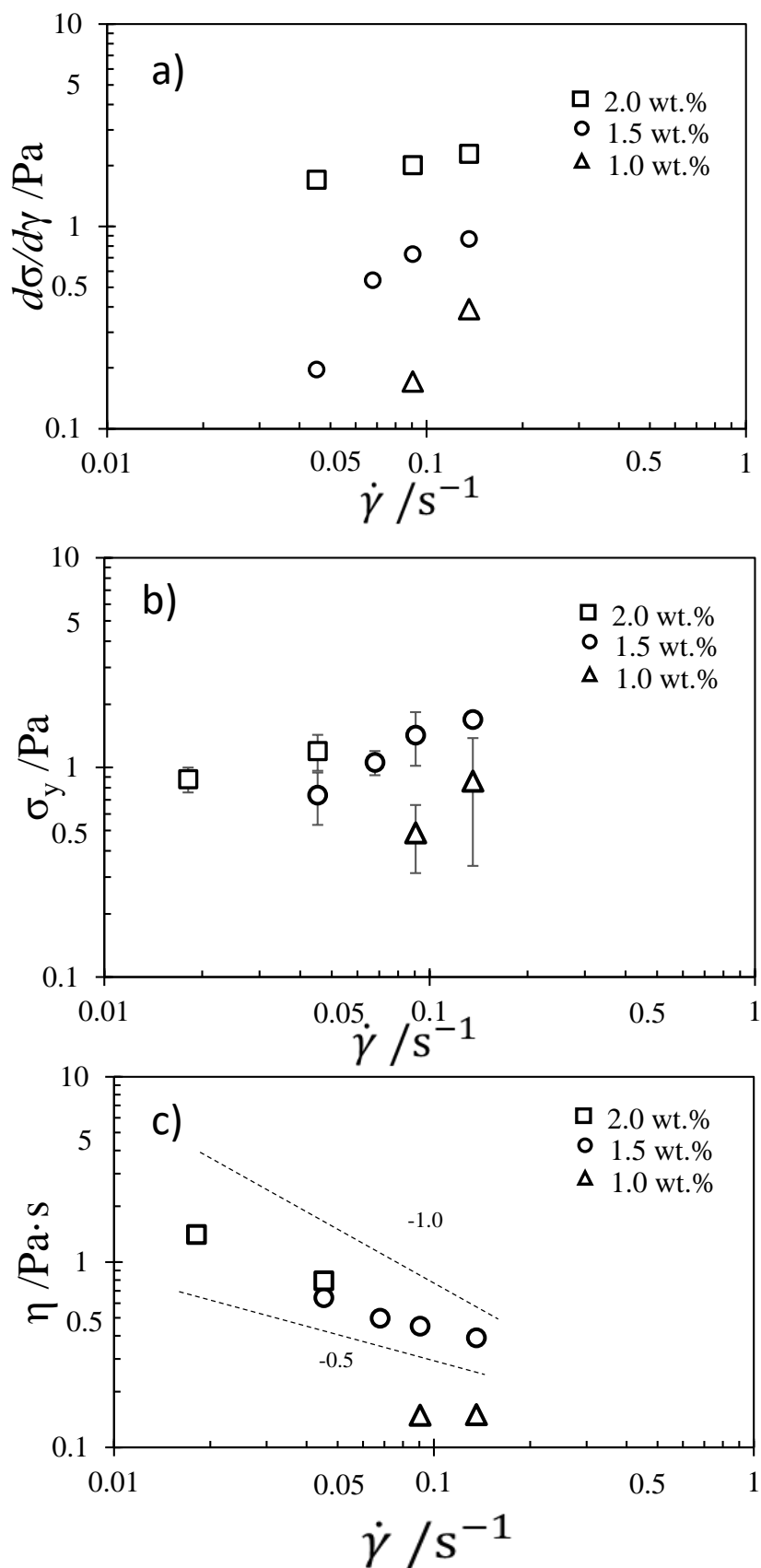


Fig. 7. Plots of the calculated rheological parameters as a function of strain rate for different IC concentrations: a) $d\sigma/d\gamma$, b) yield stress, σ_y and c) apparent viscosity η curves. A line with slope of -1 and -0.5 are drawn as guides for the gradients.

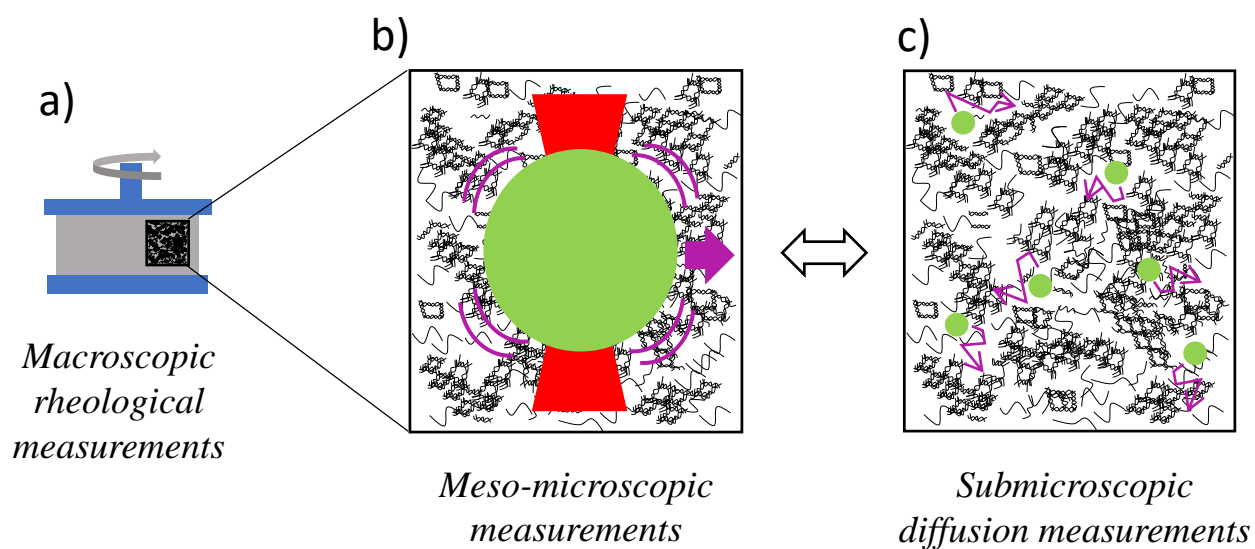


Fig. 8. Schematic representation of IC gel network developed from different investigations a) macroscopic, b) meso- to microscopic, c) submicroscopic. **Figs. b-c are drawn assuming similar magnification.** The arrow in the optical tweezers measurements represents the mode of deformation by the probe particle, while those in the diffusion measurements represents the thermal movement of tracer particles.

Highlights:

- Optical tweezers-based microrheology provides access to the mesoscopic properties of iota carrageenan (IC) gels
- Passive microrheology captures a wide spectral content of IC viscoelastic properties uncovering the dynamics of the network
- New insights into the hierarchical structure of IC gels were provided, especially at the mesoscopic scale

There are no conflicts of interest to declare.

Lester Geonzon: Conceptualization, Methodology, Software, Visualization, Formal Analysis, Investigation, Writing-Original Draft, Writing - Review & Editing. **Motoyoshi Kobayashi:** Supervision; Conceptualization; Methodology; Formal analysis; Validation; Visualization; Writing-original draft; Writing-review & editing; Resources; Funding acquisition; Project administration. **Manlio Tassieri:** Formal analysis; Software; Writing-review & editing. **Rommel Bacabac:** Formal Analysis; Resources; Writing - Review & Editing. **Yasuhisa Adachi:** Formal analysis; Resources; Funding acquisition; Project administration. **Shingo Matsukawa:** Formal Analysis; Resources; Writing - Review & Editing.