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1	Comparison of $\gamma$ -ray spectrometry and ICP-MS methods for
2	measuring heat-producing radioactive elements of rocks: a
3	case study on borehole samples from the Sichuan basin, China
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16	Abbreviated title: Comparison of different methods for measuring heat-producing
17	radioactive elements
18	
19	Abstract: We compared the consistency of $\gamma$ -ray spectrometry and inductively coupled
20	plasma mass spectrometry (ICP-MS) by analyzing measurement results of the radioactive
21	heat-producing elements U, Th, and K from borehole samples. This analysis was based
22	on 49 samples obtained from mudstone, siltstone, and carbonate rock, and 11 of the 15
23	control groups showed great consistency. The radioactive heat production(RHP) of
24	carbonate rocks was relatively low (0.23-0.63 $\mu W~m^{\text{-3}})$ and was mainly contributed by U.
25	Mudstone and siltstone have higher RHPs, which was 1.73 $\pm 0.46~\mu W~m^{\text{-3}}$ and 2.04 $\pm$
26	0.49 $\mu$ W m <sup>-3</sup> , respectively.
27	Keywords: γ-ray spectrometry; ICP-MS; Radioactive heat production; Radioactive
28	elements; Sichuan basin
29	

# 30 Introduction

31 Heat production rate of rocks is an important and essential parameter for studies on 32 terrestrial heat flow, deep thermal conditions and lithospheric thermal structure. It is also 33 essential for the research on the thermal history of basins. Acquisition of accurate rock 34 heat-production data is of great significance for studies involving present- and paleo-35 geothermal field calculations [1, 2], continental lithosphere structure [3-7], tectonic 36 activity and lithospheric evolution [8, 9] and thermal evolution simulation of hydrocarbon 37 source rocks and oil and gas resource assessment. The earth crust contains over 60 38 unstable nuclides, among which three elements (uranium, thorium, and potassium) can 39 provide large amounts of thermal energy from radioactive isotopes. The temporal and 40 spatial distributions of radioactive elements have great influences on the earth internal 41 thermal field and their contribution to the earth surface heat flow can exceed 30-40% [7, 42 10, 11].

43 The heat production rate of crustal rocks is primarily determined by the geochemical 44 method, which involves directly measuring the content of the radioactive heat-producing 45 elements U, Th, and K in rock samples. Compared with the seismic wave speed method 46 and geophysical logging method, the geochemical method produces higher quality data 47 and is now the most common method used to determine the heat production rate of rocks. 48 Currently, the primary methods for measuring the radioactive heat-producing element 49 content are inductively coupled plasma mass spectrometry (ICP-MS) [12] and  $\gamma$ -ray 50 spectrometry [13-21]. This paper analysed the differences between the two, based on 51 measurement results obtained from sedimentary basin borehole core samples.

# 52 Geological background and sample preparation

## 53 Geological background

54 The diamond-shaped Sichuan Basin, located in southwest China, is surrounded by 55 mountain ranges (Fig. 1a). The Sichuan Basin began to develop at the base of the Upper 56 Yangtze Platform and has undergone two evolutionary stages: an oceanic cratonic basin 57 stage and a continental foreland basin stage. The end of the Middle Triassic was the 58 Upper Yangtze region's oceanic-continental sediment convergence period. Before this, 59 the deposits of this region consisted primarily of marine carbonate rocks, sandstones and 60 mudstones. After this period, the deposits consisted primarily of clastic sedimentary 61 rocks (Fig. 1b). During the Cambrian to Early Ordovician, which was a period of cratonic basin subsidence, marine transgressive sand-shale deposits began forming, with carbonate rocks and evaporites towards the top. Then the Yangtze Plate converged with the Cathaysian Block in the Middle Ordovician, resulting in the basin's uplifting. The basin's Central Uplift expanded in the Silurian, and sandy shale with limestone intercalation deposits as well as sheet stone deposits developed at the edge of the craton, causing the Silurian, Devonian, and Carboniferous to disappear in the basin.

68 The Permian formed on top of the Carboniferous. Carbonate rocks and evaporites 69 formed during the Early and Middle Triassic, and the Feixianguan Formation developed 70 on top of the Permian. The Sichuan Basin went through a marine-continental transition 71 period during the Late Triassic, and the deposits transformed from marine carbonate-72 evaporite rock to continental fluvial-deltaic deposits. The Jurassic was characterized by 73 the red clastic rock deposits and lacustrine carbonate at the center of the lake. Then the 74 lake basin shrank in the Cretaceous Period, occupying mainly southern and western 75 Sichuan, and contained sandstone and mudstone deposits.

## 76 Sample preparation

The borehole samples were collected from the drill cores in the subsurface strata. These drill cores are marked with depth and their startigraphical properties can be judged by geophysical prospecting and well logging. Therefore, the borehole sample is the most direct approach to study the earth interior. It is frequently used in mineral exploration and geological research. A data column can be set up based on the analysis result of the borehole samples.

The samples in this study come from boreholes in the Sichuan basin, with the exception of boreholes KQ3 and HC1 (located at the edge of the present basin area), and consist mainly of mudstones, sandstones, siltstones, and carbonate rocks (Fig. 1b). The samples were pulverized by the self-developed rock crushers in the State Key Laboratory of Lithospheric Evolution, Institute of Geology and Geophysics, Chinese Academy of Sciences.

## 89 Experimental Procedure

#### 90 Calculation method of the heat production rate

91 During heat production rate measurements, the content of U, Th, and K in the rock 92 samples is determined and the radioactive heat production(RHP) can be calculated with 93 empirical formulas. Although many formulas had been presented in recent years [22-24],

94 the coefficients from Rybach (1988) [25] are the most used in literature[26]:

95  $A = \rho (9.52 C_U + 2.56 C_{Th} + 3.48 C_K) 10^{-5}$ (1)

96 where A is the rock's radioactive heat production (in  $\mu$ W m<sup>-3</sup>),  $\rho$  is the rock density (in kg 97 m<sup>-3</sup>), and  $C_U$ ,  $C_{Th}$ , and  $C_K$  represent the rock's U (in  $\mu$ g g<sup>-1</sup>), Th (in  $\mu$ g g<sup>-1</sup>), and K (in

98 weight%) contents, respectively.

# 99 Experimental approaches

We have two groups of samples which are labeled as "S" and "C". Each sample in 100 101 both groups is also labeled with a number. The samples who have the same number are 102 considered as the same sample because they were collected from the same lithological 103 section in the same strata of a borehole. The comparison used in this study intentionally 104 incorporates conditions that could produce inconsistent results. For example, while the 105 compared samples came from the same boreholes and had identical stratigraphy and 106 lithology, the sampling depths were not entirely consistent (Table 1). If the measurement 107 results are still identical, this proves not only the comparability between the two methods 108 but also the consistency in content of the radioactive heat-producing elements with the 109 same lithology and from the same stratigraphy.

110 S-samples (measured using the ICP-MS methods) were dissolved by the mixed acid closed digestion method. <sup>238</sup>U and <sup>232</sup>Th were measured by PerkinElmer, Type Elan 111 112 DCR-e ICP-MS, using the standard mode (neither KED nor DRC mode was used), while  $^{40}$ K was measured by PerkinElmer Type 5300DV ICP-OES, under conditions of 20  $^{\circ}$ C 113 114 and relative humidity of 30%. The dilution factor was 500 (dilute 0.1g to a constant 115 volume of 50 ml). The experiment was conducted according to the QC reference material 116 GBW07106. The blank values of both U and Th were adopted as 0.003ng/ml. The 117 Relative Standard Deviation(RSD) of the measured data was estimated as: RSD<5%, 118 while the concentration > 1ppm; RSD<10%, while the concentration < 1ppm.

119 The other set of samples (labelled C-samples, measured by  $\gamma$ -ray spectroscopy 120 method) were analyzed at the Lab of Cosmogenic Nuclide Chronology, Institute of 121 Geology and Geophysics, Chinese Academy of Sciences, using a hyper-pure  $\gamma$ -ray 122 detector (HPGe) with an efficiency of around 30%. Each sample was measured during an 123 accumulating time of 24 h. An empty cylindrical plastic container was placed in the

124 detection system for a counting period of 24 h, in order to collect the background count 125 rates. The naturally occurring radionuclides considered in the present analysis of the measured  $\gamma$ -ray spectra are: <sup>212</sup>Pb (with a main gamma energy at 239 keV and a gamma 126 yield of 43.1%), <sup>214</sup>Pb (352 keV, 37.1%), <sup>214</sup>Bi (609 keV, 46.1%), <sup>228</sup>Ac (911 keV, 29.0%) 127 and <sup>40</sup>K (1461 keV, 10.7%). Under the assumption that secular equilibrium was reached 128 between <sup>232</sup>Th and <sup>238</sup>U and their decay products, the concentration of <sup>232</sup>Th was 129 determined from the average concentrations of <sup>212</sup>Pb and <sup>228</sup>Ac in the samples, and that of 130 <sup>238</sup>U was determined from the average concentrations of the <sup>214</sup>Pb and <sup>214</sup>Bi decay 131 productsm[15, 21]. As a result, the radionuclide concentrations of <sup>232</sup>Th and <sup>238</sup>U were 132 obtained. <sup>40</sup>K concentration was achieved through a direct measurement. The systems 133 134 calibration was performed by using a reference material of GBW08304. When sampling <sup>137</sup>Cs, <sup>210</sup>Pb, <sup>226</sup>Ra, <sup>238</sup>U, <sup>232</sup>Th, and <sup>40</sup>K nuclides with this spectrometer for 24 h, the 135 confidence level exceeds 99%, meaning that the gross counts >  $3C_{BK}^{1/2}$  ( $C_{BK}$  indicates 136 background counts) and the minimum detectable activities are 0.5, 3.3, 0.85, 4.0, 0.45, 137 138 and 7.0 dpm, respectively. Details of energy and efficiency calibration methods, and 139 quality control follow the method described by Foster et al [20]. In addition, the precise 140 experiment of this method is not the major motivation for the ICP-MS measurements. We 141 focused on examining the uniformity for the same samples (samples collected from the 142 same lithological section of the drill cores) and this work was not limited in comparison 143 of the precisions of the two methods.

# 144 **Results and discussion**

145 Table 2 and Table 3 show the measurement results. The sample data are divided into 146 three stages for comparison. The direct measurement results of individual samples are 147 presents in the first stage. The second stage consists of samples taken from identical 148 boreholes and having identical stratigraphy and lithology (divided into S-group and C-149 group based on their measurement method). In the third stage, we consider the S-samples 150 and C-samples that taken from identical boreholes and having identical stratigraphy and lithology as one group. The standard deviation have not been given for the A<sub>S</sub> or A<sub>C</sub> data 151 in the third stage, because some of them were calculated from multi- individual 152 153 measurement results but the other ones used the measurement results directly.

154 Figure 2 shows that 11 out of the 15 sample groups, with the exception of C1, C7, 155 C12, and C14, have similar heat production rate values. It is hard to determine the reason 156 behind the larger discrepancies in the sample testing results, however, since 73.3% of the 157 sample heat production rate values fall within a range that allows for their comparison. 158 Therefore, our preliminary conclusion is that the results obtained using the two testing 159 methods can be used as the basis for further analysis. In order to facilitate comparison, 160 we selected the 11 sets of C and S samples with relatively close thermal conductivity and 161 calculated their average values. Figure 3 shows the averaged heat production rate values for the most similar samples. The margin of error between the C and S samples fell 162 between 0.4% (S5: 1.96 µW m<sup>-3</sup>; C5: 1.95 µW m<sup>-3</sup>) and 18% (S10: 2.13 µW m<sup>-3</sup>; C10: 163 2.60  $\mu$ W m<sup>-3</sup>), except for the No.11 group, for which the difference reached 38%, due to 164 the very low values (S11: 0.17  $\mu$ W m<sup>-3</sup>; C11: 0.23  $\mu$ W m<sup>-3</sup>) with greater uncertainties. 165

When taking into account sample lithology, the heat production rate values vary 166 greatly with different lithologies, ranging from  $1.73 \pm 0.46 \,\mu\text{W m}^{-3}$  for siltstone (G2, G8), 167  $2.04 \pm 0.49 \ \mu\text{W} \text{ m}^{-3}$  for mudstone (G3, G4, G5, G9, and G10), and  $0.50 \pm 0.12 \ \mu\text{W} \text{ m}^{-3}$  for 168 169 dolomite. The statistical results conform to the general relationship between heat 170 production rates and lithology, meaning that carbonate rocks have a lower heat 171 production rate than clastic rocks. In addition, the heat generation rate of the Triassic 172 mudstone samples (G3, G4, and G5) from the HC1 borehole is lower than that of the 173 Paleozoic mudstone samples (G9, G10) from the KQ3 borehole, which is a reflection of 174 the regional and stratigraphic differences.

175 In general, the Upper Yangtze region, especially the inner Sichuan Basin, has a 176 relatively low heat production rate. For example, the Qaidam Basin on the west side of the Qinghai-Tibetan plateau reaches an average heat production rate of 2.0-2.2  $\mu W~m^{\text{-3}}$ 177 178 [27]. However, the Sichuan Basin is a foreland basin that began to form on top of the 179 ancient cratonic basement during the final phase of the Middle Triassic, and the Emeishan 180 basalts eruption during the Late Permian formed the distinctive Large Igneous Province 181 on the outer rim of the basin. The relatively low heat production rates of ancient carbonate rocks and basaltic rocks (~0.11  $\mu$ W m<sup>-3</sup>; 0.63 ± 0.12  $\mu$ W m<sup>-3</sup>) [16, 28] may be 182 183 one of the reasons why the rock formations of the Paleozoic and basaltic provenance 184 areas of the Sichuan Basin have relatively low heat production rates.

185 Figure 4 shows the heat contribution of heat-generating elements. It is evident that 186 both measurement methods produce highly consistent results, showing that, in general, U 187 and Th have greater contributions and K has a lower contribution. Samples S13 and C13 188 show a discrepancy in heat-contributing elements; in the former, U has the highest 189 contribution rate, whereas in the latter, Th and K contribute more. The lithology of this 190 sample group was algal dolomite, and so it can be inferred that during sample processing, 191 the main contributing source in S13 was marine sedimentss, in which U is often the 192 dominant contributor (97%, [29]), whereas the main contributing source in C13 was algae. 193 Naturally, the heat generation rate of the rock and the contribution of the heat production 194 rate of each element varies in different regions. For example, in some regions in Egypt, 195 the contribution of the heat production rate of U in metamorphic rocks and pyrogenic 196 rocks can reach up to 51-76% [16], whereas the contribution of the heat production rate 197 of U in sedimentary rocks is 62-69.6%, as compared to 26.9-34% in Th and less than 5% 198 in K [30]. However, in the Alps-Apennines boundary zone, potassium contributes to heat 199 generation on average by 17% [29]. It can be also realized that the results of the 200 contribution rate of the three radioactive element were very uniform (|D| < 0.3), while the RHPs were greater than 1.0  $\mu$ W m<sup>-3</sup>. 201

202 In Fig. 5, it can be seen that samples with different lithologies differ more from each 203 other, while samples with the same lithology are consistent. Dolomites (C12, C15, and 204 S13) are mainly gathered in region A, where the contribution rate of U ranges from 82.5% 205 (S15) to ~100% (C12), while the contribution rate of Th is < 13%. Mudstone (C3, C4, 206 C5, C7, C9, and C10) is mainly gathered in region B, where the contribution of U is <207 40%, while the contribution of Th is > 45%. The distribution of siltstone (region C) is 208 closer to mudstone, and forms a transitional distribution from region A to nearby region 209 B. Compared with mudstones, the contribution of U in siltstone is greater, while the 210 contribution of Th is smaller, as shown in Fig. 6. We calculated the average contribution 211 of the heat production rate of U and Th as 50% and 47% in mudstones respectively, and 212 that of the heat production rate of U and Th as 47.4% and 41.6% in siltstones respectively. 213 The vertical distribution of U, Th, and K and of the element ratios (U/K, U/Th, and 214 Th/K) reflects the sedimentation environment and history [19, 31-34]. In this study, the 215 ratio of U/Th is 0.20 in mudstone and 0.30 in siltstone, possibly due to the fact that

216 uranium migrates easily into the external environment, and compared with siltstone 217 deposits, mudstone deposits are more susceptible to damage due to exhaustive weathering.

218 The distribution of radiogenic heat production in the continental lithosphere is very 219 important to geothermic and lithospheric studies. But continental crust rocks show a high 220 variability in radiogenic heat production (RHP) due to their petrogenesis and the complex 221 differentiation and redistribution processes affecting the whole lithosphere. The trace 222 element character of U and Th and their association with accessory minerals makes it 223 difficult to establish a direct relationship between RHP and lithology. These factors 224 impede any efforts to find a simple depth-dependent function for RHP within the 225 continental crust [35]. However, the measurement of the samples from the boreholes 226 made it possible to observe the distribution of the heat production in the upper crust in the 227 research area. With the using of the heat production rate dataset, the heat flow 228 contribution of the upper crust can be estimated, and the lithospheric thermal structure 229 can also revealed.

K has a relatively low heat production contribution, which is negligible for heat production rate calculations in the shallow parts of the earth's crust [16]. However, in deeper parts of the earth, including the core, K is assumed to be an important heatproducing element [36]; therefore, the surface heat flow, which also includes mantle heat currents, contains the heat produced by K in the deeper parts of the earth.

#### 235 Conclusions

The following conclusions were obtained through the comparison of the  $\gamma$ -ray spectrometry and the ICP-MS radioactive nuclide measurement results and the heat production rate result analysis.

Other than the measurement differences caused by the lithological changes, most of the compared samples had consistent radioactive nuclide content values and heat production rate calculation results, demonstrating that both methods of measuring radioactive nuclides ( $\gamma$ -ray spectrometry and ICP-MS) are highly stable. The result provide a reference for the uniformity between the two experimental methods of rock sample analysis. It also proved the rationality of the integrate use for the heat production rate data that analysed by the two different methods.

246 The rock heat production rate and the contribution of the heat production rate of 247 each heat-producing element are primarily related to lithology. Carbonate rock, especially dolomite, has a lower heat production rate, ranging from 0.23-0.63  $\mu$ W m<sup>-3</sup>, with U being 248 the primary contributor and the other two heat-producing elements having smaller 249 250 contributions. Mudstone and siltstone have higher heat production rates,  $1.73 \pm 0.46 \mu W$  $m^{-3}$  and 2.04  $\pm$  0.49  $\mu$ W  $m^{-3}$ , respectively, with U and Th being the primary contributors. 251 252 In mudstone, thorium's contribution rate (50.0%) is greater than uranium's (47.0%), 253 while in siltstone, uranium's rate (47.4%) is greater than thorium's (41.6%).

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266 Fig. 1 Sampling borehole locations and stratigraphic column of the research area







269 Fig. 2 Comparison of the heat production rate values of the S-group (analyzed by ICP-





273 Fig. 3 Comparison of the average heat production rate values of the more consistent S-

274 group (analyzed by ICP-MS and ICP-OES) and C-group (analyzed by γ-ray spectrometry)

samples.



Fig. 4 Heat production rate contribution of radioactive heat-producing elements in the
 selected S-group (analyzed by ICP-MS and ICP-OES) and C-group (analyzed by γ-ray
 spectrometry) samples



Fig. 5 Ternary diagram of heat production rates of radioactive elements in samples



**Fig. 6** Comparison of heat production rates and heat production contributions of various

- 287 elements in siltstone and mudstone
- 288

- 289 Tables
- 290

291	Table 1	Sample	e informa	tion: S-sam	ples measured	d using	ICP-MS at	nd ICP-OES,	C-
								,	

292 samples measured using γ-ray spectrometry

Group No.	Borehole	Strata	Lithology	Sample No.	Sampling code	Depth/m
				S1	DS8-06	2416.67
G1	DS8	$J_2s$	sandstone		DS8-05	2642.69
				CI	DS8-07	2517.26
				S2	FG21-08	3747.28
G2	FG21	$T_3x$	siltstone		FG21-02	3785.74
				C2	FG21-03	3784.64
				62	HC1-07	5844.5
				33	HC1-09	4776.8
G3	HC1	$T_2zg$	mudstone		HC1-15	4943.61
				C3	HC1-17	4776.8
					HC1-19	4372.83
				S 4	HC1-16	3784.5
G4	HC1	$T_3z$	sandstone	54	HC1-27	2268.8
				C4	HC1-32	3783.4
				S5	HC1-33	1316.27
					HC1-36	3568.5
G5	HC1	$T_3z$	mudstone	C5	HC1-38	3565.8
				CS	HC1-39	3321.4
					HC1-40	3320.3
G6	ні 2	I z	shall limestone	S6	HL2-05	2736.75
00	111.2	JIZ	shen innestone	C6	HL2-01	2746.2
				\$7	HL2-04	2734.96
G7	HL2	$J_1 z$	shale	37	HL2-06	2728.87
				C7	HL2-07	2734.96
				58	HYX6-04	989.9
					HYX6-07	893.5
G8	HYX6	K	siltstone		HYX6-02	1106.71
				C8	HYX6-06	1013.38
					HYX6-08	988.8
					KQ3-10	432.36
				S9	KQ3-34	336.8
G9	KQ3	$S_{1-2}wx$	mudstone		KQ3-39	114.36
				C9	KQ3-22	422.89
				0,	KQ3-23	421.79
G10	КОЗ	Si awx	lime mudstone	S10	KQ3-25	377.08
	ngo	01-2WA	inite industoire	C10	KQ3-12	438.56
G11	LS1	T <sub>i</sub> i	dolomite	S11	LS1-04	6942.62
	2.51	• IJ	doronnite	C11	LS1-05	6946.93
G12	LS1	Tal	dolarenite	S12	LS1-09	6512.76
	1.01	1 21	uolarenne	C12	LS1-16	6013.26
G13	LS1	T-1	algae dolomite	\$13	LS1-11	6323.11
015	LOI	1 21	argae doronnite	515	LS1-17	5993.37

				C13	LS1-22	6312.55
G14			mudstone	S14	MP1-02	1364.4
	MD1	La			MP1-03	1365.5
	IVIPI	33b		C14	MP1-04	1367.8
					MP1-05	1368.9
G15	71.10	TT 1	dolomito	S15	ZL12-08	1332.43
	ZL12	1 <sub>2</sub> 1	doioinite	C15	ZL12-16	1331.33

Sample	Sampling	ρ	U	Th	K	A <sub>U</sub>	A <sub>Th</sub>	A <sub>K</sub>	$A_0$	As
No.	code	(g/cm <sup>3</sup> )	$(\mu g/g)$	$(\mu g/g)$	(%)	$(\mu W/m^3)$	$(\mu W/m^3)$	$(\mu W/m^3)$	$(\mu W/m^3)$	$(\mu W/m^3)$
S1	DS8-06	2.61	1.39±0.07	6.71±0.34	1.08±0.05	0.35±0.01	0.45±0.01	0.10±0.00	0.89±0.02	0.89
S2	FG21-08	2.70	2.58±0.13	8.62±0.43	1.39±0.07	0.66±0.01	0.60±0.01	0.13±0.00	1.39±0.03	1.39
62	HC1-07	2.68	1.74±0.09	11.10±0.56	1.95±0.10	0.44±0.01	0.76±0.01	$0.18 \pm 0.00$	1.39±0.03	1 4 4
33	HC1-09	2.71	$2.05 \pm 0.10$	11.00±0.55	2.22±0.11	0.53±0.01	0.76±0.01	$0.21 \pm 0.00$	1.50±0.03	1.44
<b>S</b> 4	HC1-16	2.57	2.66±0.13	14.50±0.72	3.71±0.19	0.65±0.01	$0.95 \pm 0.02$	0.33±0.01	1.94±0.04	1.70
34	HC1-27	2.70	2.27±0.11	9.73±0.49	2.14±0.11	0.58±0.01	0.67±0.01	0.20±0.00	1.46±0.03	1.70
S5	HC1-33	2.57	2.78±0.14	14.40±0.72	3.73±0.19	0.68±0.01	$0.95 \pm 0.02$	0.33±0.01	1.96±0.04	1.96
S6	HL2-05	2.69	$0.48 \pm 0.05$	2.34±0.12	0.45±0.02	0.12±0.01	0.16±0.00	$0.04 \pm 0.00$	0.33±0.01	0.33
67	HL2-04	2.81	2.85±0.14	13.20±0.66	2.05±0.10	0.76±0.01	$0.95 \pm 0.02$	0.20±0.00	1.91±0.03	2.07
57	HL2-06	2.72	2.83±0.14	16.80±0.84	3.42±0.17	0.73±0.01	1.17±0.02	0.32±0.01	2.23±0.04	2.07
60	HYX6-04	2.69	3.36±0.17	13.90±0.70	3.02±0.15	0.86±0.02	0.96±0.02	0.28±0.01	2.10±0.04	2.05
30	HYX6-07	2.71	4.08±0.20	10.60±0.53	2.24±0.11	$1.05 \pm 0.02$	0.74±0.01	$0.21 \pm 0.00$	2.00±0.04	2.03
	KQ3-10	2.74	3.01±0.15	18.20±1.82	4.13±0.21	0.79±0.01	1.28±0.05	0.39±0.01	2.46±0.07	
S9	KQ3-34	2.79	4.98±0.25	17.50±0.88	2.24±0.11	1.32±0.02	1.25±0.02	0.22±0.00	2.79±0.05	2.71
	KQ3-39	2.80	3.58±0.18	21.30±1.06	4.08±0.20	$0.95 \pm 0.02$	1.53±0.03	0.40±0.01	2.88±0.05	
S10	KQ3-25	2.77	2.72±0.14	17.00±1.70	2.19±0.11	0.72±0.01	1.21±0.04	0.21±0.00	2.13±0.06	2.13
S11	LS1-04	2.74	$0.38 \pm 0.04$	$0.67 \pm 0.07$	0.22±0.01	0.10±0.01	0.05±0.01	$0.02 \pm 0.00$	0.17±0.02	0.17
S12	LS1-09	2.67	2.32±0.12	0.36±0.04	0.20±0.02	0.59±0.01	$0.02 \pm 0.00$	0.02±0.00	0.63±0.01	0.63
612	LS1-11	2.74	4.96±0.25	0.09±0.01	0.03±0.00	1.29±0.02	$0.01 \pm 0.00$	$0.00\pm0.00$	1.30±0.02	1.02
515	LS1-17	2.75	2.85±0.14	0.04±0.00	$0.01 \pm 0.00$	$0.75 \pm 0.01$	$0.00\pm0.00$	$0.00\pm0.00$	$0.75 \pm 0.01$	1.03
S14	MP1-02	2.65	0.86±0.09	1.21±0.06	2.87±0.14	0.22±0.01	$0.08\pm0.00$	0.27±0.01	0.56±0.02	0.56
S15	ZL12-08	2.86	1.25±0.06	$0.781\pm0.08$	0.15±0.02	0.34±0.01	0.06±0.00	0.02±0.00	0.41±0.01	0.41

Table 2 The calculated values for radioactive heat-producing element content and sample heat production rates using ICP-MS and
 ICP-OES\*

\*Notes: The data was analyzed at the Beijing Research Institute of Uranium Geology, while the experiment was run according to the QC reference material GBW07106. The blank values of both U and Th were adopted as 0.003ng/ml. The relative standard deviation (RSD) of the individual data was given by estimated as: RSD<5%, while the concentration > 1ppm; RSD<10%, while the concentration

34 5 6

< 1ppm.

#### 7 **Table 3** The calculated values for radioactive heat-producing element content and sample heat production rates using gamma-ray

8 spectrometry*	ĸ
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Sample	Sampling	ρ	23	<sup>18</sup> U	233	<sup>2</sup> Th	<sup>40</sup> K		Au	A <sub>Th</sub>	A <sub>K</sub>	$A_0$	Ac
No.	code	$(g/cm^3)$	(dpm/g)	$(\mu g/g)$	(dpm/g)	$(\mu g/g)$	(dpm/g)	(%)	$(\mu W/m^3)$	$(\mu W/m^3)$	$(\mu W/m^3)$	$(\mu W/m^3)$	$(\mu W/m^3)$
C1	DS8-05	2.63	2.29±0.05	3.07±0.06	3.09±0.03	12.67±0.14	54.82±0.88	3.01±0.05	0.77±0.01	$0.85 \pm 0.00$	0.28±0.00	$1.90\pm0.01$	1 77
CI	DS8-07	2.63	2.30±0.05	3.08±0.07	2.59±0.04	10.63±0.16	30.67±0.83	$1.69 \pm 0.05$	$0.77 \pm 0.01$	$0.72 \pm 0.00$	$0.16\pm0.00$	$1.64 \pm 0.01$	1.//
<b>C</b> 2	FG21-02	2.63	2.26±0.05	3.03±0.07	$2.85 \pm 0.04$	11.70±0.16	27.34±0.78	$1.50\pm0.04$	$0.76\pm0.01$	$0.79 \pm 0.00$	$0.14 \pm 0.00$	$1.68 \pm 0.01$	1.29
C2	FG21-03	2.65	$1.81 \pm 0.05$	2.43±0.06	$0.82 \pm 0.02$	3.37±0.10	5.02±0.47	$0.28 \pm 0.03$	$0.61 \pm 0.01$	0.23±0.00	0.03±0.00	0.87±0.01	1.28
	HC1-15	2.85	1.23±0.05	1.65±0.07	1.76±0.04	7.21±0.15	26.16±0.89	$1.44 \pm 0.05$	$0.45 \pm 0.01$	0.53±0.00	$0.14 \pm 0.00$	$1.12\pm0.01$	
C3	HC1-17	2.67	1.86±0.05	2.49±0.07	3.30±0.05	13.54±0.19	55.21±1.13	$3.04 \pm 0.06$	0.63±0.01	0.93±0.01	$0.28 \pm 0.00$	$1.84 \pm 0.01$	1.57
	HC1-19	2.76	2.13±0.05	2.86±0.06	2.69±0.04	11.03±0.15	44.41±0.90	$2.44 \pm 0.05$	0.75±0.01	$0.78 \pm 0.00$	0.23±0.00	$1.77 \pm 0.01$	
C4	HC1-32	2.57	2.06±0.05	2.75±0.07	3.37±0.04	13.84±0.17	61.42±1.08	3.38±0.06	$0.67 \pm 0.01$	0.91±0.00	$0.30\pm0.00$	$1.89\pm0.01$	1.89
	HC1-36	2.85	1.97±0.05	2.63±0.07	3.11±0.04	12.76±0.18	67.00±1.20	$3.68 \pm 0.07$	$0.71 \pm 0.01$	0.93±0.01	$0.37 \pm 0.00$	2.01±0.01	
C5	HC1-38	2.77	2.26±0.05	3.03±0.07	3.53±0.04	$14.50\pm0.18$	64.69±1.11	3.56±0.06	$0.80\pm0.01$	$1.03 \pm 0.01$	$0.34 \pm 0.00$	2.17±0.01	1.05
05	HC1-39	2.77	2.14±0.05	$2.87 \pm 0.07$	$2.50\pm0.04$	10.26±0.16	59.54±1.07	3.27±0.06	$0.76 \pm 0.01$	0.73±0.00	$0.32 \pm 0.00$	$1.80\pm0.01$	1.95
	HC1-40	2.87	2.12±0.05	$2.84\pm0.07$	2.74±0.04	11.26±0.17	42.09±0.97	2.31±0.05	$0.78 \pm 0.01$	$0.83 \pm 0.00$	0.23±0.00	$1.83 \pm 0.01$	
C6	HL2-01	2.76	0.58±0.03	$0.77 \pm 0.04$	$0.26 \pm 0.02$	$1.09 \pm 0.07$	1.27±0.34	$0.07 \pm 0.02$	$0.20\pm0.00$	$0.08 \pm 0.00$	$0.01 \pm 0.00$	0.29±0.01	0.29
C7	HL2-07	2.77	1.30±0.04	1.75±0.05	$1.72 \pm 0.03$	7.05±0.12	13.81±0.59	$0.76 \pm 0.03$	$0.46 \pm 0.01$	$0.50\pm0.00$	$0.07 \pm 0.00$	$1.04 \pm 0.01$	1.04
	HYX6-02	2.89	2.02±0.04	2.70±0.06	$2.80\pm0.04$	11.49±0.15	32.72±0.78	$1.80\pm0.04$	$0.74 \pm 0.01$	$0.85 \pm 0.00$	$0.18\pm0.00$	1.77±0.01	
C8	HYX6-06	2.71	2.16±0.04	2.89±0.06	2.62±0.03	10.76±0.13	33.16±0.71	$1.82 \pm 0.04$	$0.75 \pm 0.01$	$0.75 \pm 0.00$	$0.17 \pm 0.00$	$1.66 \pm 0.01$	1.93
	HYX6-08	2.68	3.34±0.06	4.47±0.08	3.23±0.04	13.25±0.18	57.72±1.10	3.17±0.06	$1.14\pm0.01$	0.91±0.01	$0.30\pm0.00$	2.35±0.01	
CO	KQ3-22	2.77	2.77±0.05	3.71±0.07	4.41±0.04	18.12±0.18	60.13±1.00	3.31±0.06	$0.98 \pm 0.01$	$1.29\pm0.01$	0.32±0.00	$2.58\pm0.01$	2 (0
09	KQ3-23	2.8	2.29±0.05	3.06±0.07	4.83±0.05	19.83±0.20	72.39±1.17	$3.98 \pm 0.06$	$0.82 \pm 0.01$	$1.42 \pm 0.01$	$0.39 \pm 0.00$	2.63±0.01	2.60
C10	KQ3-12	2.77	2.35±0.05	3.15±0.07	4.69±0.05	19.25±0.19	76.72±1.13	4.22±0.06	0.83±0.01	1.37±0.01	$0.41 \pm 0.00$	$2.60\pm0.01$	2.60
C11	LS1-05	2.72	0.67±0.07	$0.89 \pm 0.06$	_	_	_	_	0.23±0.01	_	_	0.23±0.01	0.23
C12	LS1-16	2.77	3.29±0.04	4.41±0.06	_	_	_	_	$1.16\pm0.01$	_	_	$1.16\pm0.01$	1.16
C13	LS1-22	2.74	1.44 ±0.06	1.93±0.08	$1.07 \pm 0.04$	4.38±0.15	20.63±0.92	$1.13 \pm 0.05$	$0.50\pm0.01$	0.31±0.00	$0.11 \pm 0.00$	$0.92 \pm 0.01$	0.92
	MP1-03	2.65	2.23±0.02	2.99±0.03	3.28±0.03	13.48±0.12	44.56±0.91	$2.45 \pm 0.05$	$0.75 \pm 0.00$	0.91±0.00	0.23±0.00	$1.90\pm0.01$	
C14	MP1-04	2.56	2.33±0.06	3.12±0.08	$2.86\pm0.05$	11.75±0.19	19.98±0.83	$1.10\pm0.05$	$0.76\pm0.01$	0.77±0.01	$0.10\pm0.00$	$1.63 \pm 0.01$	1.76
	MP1-05	2.76	2.08±0.04	2.78±0.06	2.93±0.03	12.04±0.14	34.63±0.71	1.90±0.04	0.73±0.01	$0.85 \pm 0.00$	$0.18 \pm 0.00$	1.76±0.01	
C15	ZL12-16	2.64	1.32±0.04	1.77±0.05		—	2.71±0.37	0.15±0.02	$0.44 \pm 0.01$	—	$0.01 \pm 0.00$	0.46±0.01	0.46

9 \*Notes: The measurements of the radioactivity concentrations were carried out at the Lab of Cosmogenic Nuclide Chronology, Institute of Geology and Geophysics, Chinese Academy of Sciences, according to the QC standard GBW08304. Dpm/g is the radioactivity specific activity (unit: the number of decayed 10 atoms per gram of sample per minute).  $\sigma$  is the standard deviation (unit: dpm/g = 1/60 Bq/g). "—" indicates sample radioactivity below the minimum detection 11 12 limit when the calculation value is "0". Content calculation principle is the number of decayed atoms per second in a radioactive element or isotope (radioactivity 13 specific activity is the radioactivity of a nuclide per unit mass of material), expressed by the formula  $A=\lambda N$  where A is radioactivity specific activity,  $\lambda$  is the decay constant ( $\lambda = \ln 2/T$ , where T is the half-life of a nuclide) and N is the number of decayed nuclides in a unit of mass. For details on the method of obtaining 14 the nuclide content of a material based on radioactive specific activity could be found from references [16-21]. A<sub>U</sub>, A<sub>Th</sub>, A<sub>K</sub> indicates the heat-production rate of 15 the U, Th and K respectively, while A<sub>C</sub> indicates the heat-production rate of each S-sample. 16

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