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1 **Intergranular diffusion rates from the analysis of garnet surfaces:**
2 **implications for metamorphic equilibration**

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4 Tim J. Dempster, Shona Symon¹ & Peter Chung

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6 School of Geographical and Earth Sciences, University of Glasgow, Glasgow G12

7 8QQ, UK.

8 ¹ Present address: Department of Civil and Environmental Engineering,

9 University of Strathclyde, Glasgow G1 1XJ, UK.

10

11 **ABSTRACT**

12 Novel approaches to garnet analysis have been used to assess rates of
13 intergranular diffusion between different matrix phases and garnet
14 porphyroblasts in a regionally metamorphosed staurolite-mica-schist from the
15 Barrovian-type area in Scotland. X-ray maps and chemical traverses of planar
16 porphyroblast surfaces reveal chemical heterogeneity of the garnet grain
17 boundary linked to the nature of the adjacent matrix phase. The garnet preserves
18 evidence of low temperature retrograde exchange with matrix minerals and
19 diffusion profiles documenting cation movement along the garnet boundaries.
20 Garnet-quartz and garnet-plagioclase boundaries preserve evidence of sluggish
21 Mg, Mn and Fe diffusion at comparable rates to volume diffusion in garnet,
22 whereas diffusion along garnet-biotite interfaces is much more effective.
23 Evidence of particularly slow Al transport, probably coupled to Fe³⁺ exchange, is
24 locally preserved on garnet surfaces adjacent to Fe-oxide phases. Ca distribution
25 on the garnet surface shows the most complex behaviour, with long wavelength

heterogeneities apparently unrelated to the matrix grain boundaries. This implies that the Ca content of garnet is controlled by local availability and is thought likely to reflect disequilibrium established during garnet growth. Geochemical anomalies on the garnet surfaces are also linked to the location of triple junctions between the porphyroblasts and the matrix phases, and imply enhanced transport along these channels. The slow rates of intergranular diffusion and the characteristics of different boundary types may explain many features associated with the prograde growth of garnet porphyroblasts. Thus minerals such as quartz, Fe-oxides and plagioclase whose boundaries with garnet are characterized by slow intergranular diffusion rates appear to be preferentially trapped as inclusions within porphyroblasts. As such grain boundary diffusion rates may be a significant kinetic impediment to metamorphic equilibrium and garnet may struggle to maintain chemical and textural equilibrium during growth in pelites.

Keywords: garnet, intergranular diffusion; porphyroblast surfaces; retrograde equilibration; grain boundaries.

42

43 INTRODUCTION

44 The transport of chemical components within rocks is recognized as a rate
45 limiting process during metamorphism that facilitates the growth and
46 dissolution of minerals and allows equilibrium to be established between phases
47 (Mueller *et al.*, 2010; Carlson *et al.*, 2015a). Volume diffusion within minerals is
48 known to be relatively slow in metamorphic conditions as demonstrated by the
49 preservation of growth zoning of cations in many metamorphic porphyroblasts
50 (e.g. Kohn, 2003). However there is often a general assumption that the rates of

51 intergranular diffusion are relatively fast (Joesten, 1991; Dohmen & Milke,
52 2010). In part this contributes to the belief that thermodynamic equilibrium
53 between minerals is readily attained during prograde metamorphism. Despite
54 this assumption, evidence for disequilibrium in metamorphic rocks suggests that
55 kinetic factors, such as transport along grain boundaries, may be important rate-
56 controlling steps in attainment of metamorphic equilibrium and porphyroblast
57 growth (Carlson, 1989; Carlson, 2002; Pattison & Tinkham, 2009; Mueller *et al.*,
58 2010; Pattison *et al.*, 2011; Carlson *et al.*, 2015b). The uncertainty associated
59 with rates of intergranular transport is compounded by the relative paucity of
60 estimates for grain boundary diffusion rates in either natural samples or from
61 diffusion experiments relevant to common rock types and geochemical species
62 (Brady, 1983; Florence & Spear, 1995; Dohmen & Milke, 2010; Marquardt *et al.*,
63 2011; Okudaira *et al.*, 2013; Bromiley & Hiscock, 2016). Other characteristics of
64 grain boundary interfaces, such as permeability and effective solubility of
65 components in the intergranular medium are recognized as important controls
66 on reaction mechanisms (Dohmen & Chakraborty, 2003), but some of these are
67 also rather poorly constrained.

68 Garnet is the most commonly studied metamorphic phase, in part because of it
69 has relatively sluggish volume diffusion rates for divalent cations (Chakraborty &
70 Ganguly, 1992; Carlson, 2006; Vielzeuf *et al.*, 2007; Ganguly, 2010). Hence garnet
71 has an ability to preserve changing chemistry during growth and potentially to
72 record the evolution of pressure and temperature during metamorphic events.
73 However many studies have demonstrated evidence of volume diffusion in
74 garnet at elevated temperatures (Woodsworth, 1977; Yardley, 1977) and
75 associated exchange during retrograde cooling (Tracy, 1982; Dempster, 1985;

76 Ehlers *et al.*, 1994; Florence & Spear, 1995). Recognition of such diffusive re-
 77 equilibration is important for the assessment of peak metamorphic P-T
 78 conditions (Spear, 1991; Florence & Spear, 1991; Kohn & Spear, 2000; Caddick *et*
 79 *al.*, 2010); metasomatic changes within the effective whole rock composition
 80 (Spear, 1988; Florence & Spear, 1993); and, geospeedometry and the
 81 determination of the duration of metamorphic events (Lasaga, 1983; Ague &
 82 Baxter, 2007; Caddick *et al.*, 2010; Müller *et al.*, 2015).

83 The study and identification of minerals using thin sections (Sorby, 1851) has
 84 become an essential and integral part of geoscience (Vernon, 2004) and the
 85 technique is at the heart of many, perhaps most, petrological advances. However
 86 the dominance of this approach has created an ingrained and inevitable
 87 emphasis on the study of the interiors of the minerals. Arguably this has
 88 occurred at the expense of understanding grain boundary processes.

89 Using a new approach to mineral analysis, this study attempts to assess the rates
 90 of intergranular diffusion along garnet grain boundaries. We use traditional
 91 techniques to analyze garnet composition in 1D profiles perpendicular to the
 92 grain boundaries in combination with 2D X-ray mapping of sections through the
 93 central parts of the porphyroblasts to monitor volume diffusion. In a novel
 94 approach we also map compositional gradients in one and two dimensions on
 95 the exposed surfaces of porphyroblasts to constrain rates of chemical transport
 96 along garnet-matrix grain boundaries.

97

98 **GEOLOGICAL SETTING AND PETROGRAPHY**

99 The Late Proterozoic Dalradian schists in the Scottish Highlands experienced
 100 peak regional metamorphic conditions during the Ordovician (Oliver *et al.*, 2000;

101 Dempster *et al.*, 2002), although parts of the succession have also experienced
102 earlier Precambrian deformation and metamorphism (Dempster & Jess, 2015).
103 Sample GL774 is of a staurolite garnet mica schist from the kyanite zone in the
104 Barrovian type area (Barrow, 1893) in Glen Effock at the north end of Glen Esk,
105 eastern Scotland (UK grid ref NO 4238377169). The schist contains some <1 cm
106 thick quartz-rich layers but generally is biotite- and muscovite-rich with large
107 (up to 3 mm diameter) garnet porphyroblasts (Fig. 1) and smaller 1 mm long
108 staurolite concentrated within the thicker micaceous layers. In thin section view,
109 garnet porphyroblasts are typically equant with planar gently curving grain
110 boundaries (Fig. 1). Some have locally irregular margins characterized by
111 embayments (up to 1 mm wide) filled with aggregates of granoblastic plagioclase
112 and quartz, with minor muscovite and staurolite. Micas are strongly aligned and
113 typically partially wrap around the garnet porphyroblasts, which contain
114 abundant small inclusions, mostly of quartz and Fe-oxides that define a weak
115 alignment at a high angle to the external fabric elements (Fig. 1). The outer ~100
116 μm edge of the garnet porphyroblasts is typically inclusion free (Fig. 1). The
117 sample lacks major retrograde alteration although Fe-staining is present on
118 some grain boundaries and small clusters of unaligned retrograde chlorite after
119 biotite are locally present. The schist contains 27% quartz, 26% biotite, 15%
120 muscovite, 13% garnet, 11% plagioclase, 3% staurolite and 3% chlorite and
121 minor Fe-oxides with accessory apatite, zircon and tourmaline.
122 Conditions of peak regional metamorphism are estimated as ~600 °C and 0.6
123 GPa (Harte & Hudson, 1979; Vorhies & Ague, 2011) and evidence from garnet
124 zoning profiles and X-ray maps suggests that re-equilibration of garnet zoning
125 profiles via volume diffusion has occurred in the rocks (Dempster, 1985; Vorhies

126 & Ague, 2011; Viete *et al.*, 2011). Garnet zoning profiles are progressively
127 smoothed during metamorphism and in this area, localized low Mg/Fe garnet
128 rim compositions are associated with exchange occurring at low temperature
129 with adjacent biotite (Dempster, 1985). Estimated biotite and garnet exchange
130 reaction geothermometry based on “rim-rim” compositions suggests that
131 equilibration may continue from peak temperature conditions to temperatures
132 of less than 500 °C (Dempster, 1985). Some studies have suggested that the
133 rocks in this area experienced very short times close to maximum temperatures
134 (Oliver *et al.*, 2000; Ague & Baxter, 2007). However the lack of well-constrained
135 ages of deposition or information on the nature of original compositional zoning
136 in garnet and apatite mean that well-constrained prograde thermal histories for
137 these rocks are not available.

138

139 **METHODS**

140 A combination of conventional polished thin section analysis and the analysis of
141 flat crystal surfaces and associated matrix surfaces was used to characterize the
142 sample. The mineral surfaces were produced by snapping of thin 4mm rock
143 slices (cf. Lawther & Dempster, 2009). This thickness proved to be optimum for
144 creating a clean snap perpendicular to the rock cleavage, thus exposing
145 porphyroblasts surfaces and intact adjacent matrix. Typically slices were cut
146 perpendicular to the rock cleavage and then snapped, also perpendicular to the
147 cleavage. Initial inspection of the snapped surfaces was done using a binocular
148 microscope and apparently suitable surfaces of porphyroblast and intact
149 adjacent matrix were selected. Pairs of slices were then stuck together with resin
150 with the porphyroblast and matrix “hole” in close proximity. Rock surfaces were

151 coated with relatively thick carbon layers (ca. 60 nm) to ensure electron
152 conductivity across the irregular surfaces. Samples were initially characterized
153 using a FEI Quanta 200F field emission environmental scanning electron
154 microscope operated at 20 kV to allow further selection of the most appropriate
155 porphyroblast surfaces. These were then analysed using a Carl Zeiss Sigma VP
156 electron microscope at 20 kV with Oxford instruments X-Max 80 energy
157 dispersive spectrometry with overnight, typically 15 hour, analysis to acquire X-
158 ray maps and traverses with minimized errors. Analyses of mineral surfaces are
159 done using energy dispersive spectrometry because the geometry of the
160 analytical system is less sensitive to surface irregularity and orientation
161 variations.

162 Porphyroblasts that form positive relief areas and the adjoining matrix “hole”
163 were both characterized using secondary and backscattered electron images.
164 The topographic surfaces of porphyroblasts and matrix were examined to ensure
165 that the snap occurred without any significant loss of intervening material. As
166 such most of the exposed garnet surfaces, and all of those analyzed, represent
167 grain boundaries between porphyroblasts and the adjacent matrix. Digital
168 elevation models were produced using Alicona Mex 3D software from stereo-
169 pairs of secondary electron images of the garnet porphyroblasts. This allowed
170 flat garnet surfaces with minimum relief to be selected for analysis. This also
171 enabled the garnet surface to be leveled at the analytical working distance and
172 oriented to avoid shielding by adjacent matrix minerals on the irregular surface
173 to ensure clear line of sight to the X-ray detector and hence optimizing the X-ray
174 yield. Surface profiles were obtained from the digital elevation models for some
175 surfaces so that topographic artifacts of the geochemical traverses could be

176 identified. Several surfaces were also analyzed in a variety of orientations to
177 ensure that none of the compositional variations are artifacts of the geometry of
178 surface. Backscattered electron images of the matrix surface adjacent to the
179 exposed garnet surface were superimposed onto the images of the garnet surface
180 so that chemical maps and compositional traverses across the garnet surfaces
181 could be directly linked to the locations of adjoining matrix phases and the
182 position of their grain boundaries. Grain boundaries between different minerals
183 were easy to identify using backscattered electron images and some of those
184 between minerals of the same type could be identified when an internal
185 structure such as cleavage could be used to identify individual grains. Those
186 between minerals lacking a prominent internal structure, such as quartz, are
187 more difficult to identify using the imaging techniques employed in this study.
188 Compositions of the garnet surfaces were assessed in traverses perpendicular to
189 the orientation of adjacent grain boundaries in the matrix to characterize a range
190 of different grain boundary types. Typically surfaces selected to avoid
191 topographic steps and surface blemishes yielded the chemical profiles that are
192 thought to represent the composition of the garnet surface most reliably.

193

194 **RESULTS**

195 **Internal zoning of garnet**

196 Geochemical analysis of polished thin sections of the interior of garnet
197 porphyroblasts reveal them to be zoned with smooth chemical variations from
198 core to rim (Fig. 2). They are characterized by relatively high Mn content cores
199 (Fig. 2b), with progressively increasing Mg/Fe from core towards the rim with
200 local decrease in Mg/Fe immediately adjacent to the rim in the outer ~50 μm .

201 Cores of the porphyroblasts are also relatively Ca-rich (Fig. 2c), although zoning
202 patterns for Ca are less obviously concentric than the other major divalent
203 cations. The edge of the garnet porphyroblasts is characterized by a low Mg
204 content in the outer 10-20 μm adjacent to biotite (Fig. 2d), but rim compositions
205 are variable and correlated with adjacent mineral type. Generally garnet adjacent
206 to quartz lacks well-developed edge-related zoning. Mn- and Ca-contents are
207 typically slightly elevated towards the outer edge of the garnet (Fig. 2a,b). The
208 analysis of the absolute edge of the porphyroblast is limited by the interaction
209 volume of the electron beam and proximity to grain boundary. Hence analyses
210 within 2 μm of the grain boundary are in part artifacts of the boundary.

211 **Surface morphology of garnet**

212 Secondary electron images of snapped rock samples show that garnet surfaces
213 preserve a combination of extensive flat surfaces, and slightly more irregular
214 surfaces containing a variety of small steps and growth facets (Fig. 3a,b). There is
215 no evidence from the surface morphology of the porphyroblasts that they might
216 represent multiple crystals (cf. Whitney *et al.*, 2008). The growth steps often
217 mirror the topography of the matrix on the opposite surface (Fig. 3d). Although
218 plagioclase grain boundaries often show porous surfaces characterized by ~ 5
219 μm pores (cf. Lawther & Dempster, 2009), most other matrix phases have
220 smooth planar boundaries. Growth steps and facets on garnet surfaces (Fig. 3b)
221 may coincide with matrix grain boundaries but more frequently the topography
222 of the porphyroblast surface is independent of the geometry of the grain
223 boundary network in the adjacent matrix (Fig. 3d). As such grain boundaries
224 between minerals on the matrix surface frequently coincide with perfectly
225 planar parts of the garnet surface (Fig. 3b,c). Some growth steps create small

226 ridges and have an influence on the chemistry of the garnet recorded across that
 227 step, however wherever possible traverses were selected in an attempt to avoid
 228 significant topographic irregularity. The effect of slight variations in the
 229 topography of the surface on the garnet chemistry depends in part on the
 230 orientation of the traverse relative to the spectrometer but is typically most
 231 marked for the lower atomic number elements and is interpreted to reflect
 232 preferential X-ray absorption (Reed, 2005). The detailed small-scale topographic
 233 features of these surfaces (Figs. 3b & 4) are hidden in conventional thin section
 234 petrography and disguised by an apparently gently curving geometry displayed
 235 by the porphyroblasts shapes (Fig. 1). These garnet and matrix surfaces are often
 236 very clean in secondary electron images. However, locally either surface may
 237 show bright BSE image deposits of Fe-oxides (Fig. 4), which may occur in small
 238 isolated spots or as more continuous thin films, other surfaces may have small
 239 clusters of S-rich deposits. Wherever possible the traverses were sited to avoid
 240 these surface deposits. The Fe-rich grain boundary deposits appear as thin
 241 discontinuous brown-orange streaks on some grain boundaries within thin
 242 sections. The thin films of Fe-oxide coating garnet surfaces appear to have little
 243 effect on the garnet compositions analysed other than when coarser grained
 244 aggregates of oxides are present (e.g. Fig. 6b). In such instances analysed
 245 compositions show more variability, especially in the low atomic number
 246 elements. The mineral volume activated under the electron beam is largely just
 247 below the surface (Reed, 2005) and the influence of the coarser deposits may be
 248 more of a topographic effect on X-ray absorption.

249 **General chemical characteristics of the garnet surfaces**

250 The X-ray maps of garnet surfaces show a patchy distribution of divalent cations,
251 especially those of Mg and Mn, that is typically spatially linked to the nature of
252 the adjacent matrix minerals. The variation in the garnet composition, expressed
253 as end member components, of the porphyroblast surface is approximately $\pm 6\%$
254 X_{prp} ; $\pm 2\%$ X_{sps} ; $\pm 6\%$ X_{alm} ; $\pm 2\%$ X_{grs} . Some compositional “boundaries” on the
255 garnet surface are sharp with locations closely matching the positions of phase
256 boundaries in the adjacent matrix (Fig. 5). Garnet surfaces adjacent to Fe-oxide
257 in the matrix shows some of the sharpest compositional “boundaries” with low
258 Al content only matched by high Fe in the garnet. The maps of chemical variation
259 on the garnet surface closely mirror the grain boundaries of the adjacent Fe-
260 oxide (Fig. 5). As such there is no evidence to suggest that significant
261 recrystallization of the matrix phases has occurred after the chemistry of the
262 garnet surface was established. Other garnet surfaces show more gradual
263 compositional changes perpendicular to the orientation of the mineral
264 boundaries in the matrix.

265 Overall the X-ray maps of Mg show the most pronounced variations on the
266 garnet surfaces that can be systematically linked to the location of matrix grain
267 boundaries (Fig. 6). Typically high Mg contents of garnet are matched by low Mn
268 and Fe contents. High Ca contents may be associated with elevated Mg contents
269 but such patterns are less consistent. Although typically the changes in surface
270 composition of garnet are very closely matched to the positions of grain
271 boundaries in the adjacent matrix some are apparently slightly offset by a few
272 microns. In part this may reflect the 3-D sub-surface geometry of the grain
273 boundary within the analytical volume and in part it is caused by distortion of

the electron images due to the different geometry of the porphyroblast surface and the matrix surface relative to the electron beam.

Geochemical characteristics of specific grain boundaries

Garnet-biotite-quartz (Mg-Mn-Fe variation)

Garnet adjacent to biotite has a consistently low Mg and high Mn and Fe contents relative to that adjacent to quartz or plagioclase (Fig. 7). Line scans of the Mg content of the garnet surface across the position of biotite-quartz grain boundaries within the adjacent matrix reveal that the garnet composition gradually changes in a smooth exponential profile marked by a increase in Mg/Fe along the garnet-quartz boundary away from the biotite and a decrease in Mg/Fe along the garnet-biotite boundary away from the quartz (Fig. 6). The compositional changes are typically more gradual within the garnet adjacent to biotite occurring over a distance of $\sim 50 \mu\text{m}$ and show a sharper transition in Mg content over a distance of $\sim 20 \mu\text{m}$ away from the quartz-biotite junction along the garnet-quartz boundary (Fig. 6b,c). Traverses of the garnet surface across quartz-biotite boundaries parallel to and perpendicular to the trace of the [001] cleavage in biotite show similar changes in garnet chemistry, although compositional changes appear to be spread over a longer distance in traverses parallel to biotite cleavage (Fig. 6c). Mn and Fe contents of garnet also show significant variation, mirroring the variation in Mg contents, linked to the adjacent matrix phase with elevated Mn next to biotite and profiles typically showing a smooth decrease in concentration along boundaries away from the biotite (Fig. 6). Although most of the analysed profiles show similar geochemical trends some differences are present. Some compositional traverses along quartz-garnet boundaries away from biotite appear to plateau at lower concentrations

299 of Mg (Fig. 6c). In other traverses Mn contents may show rather limited variation
 300 and the changes in Mg content of the garnet surface are dominantly matched by
 301 variation in the Fe content (Fig. 6b).

302 *Garnet-biotite-plagioclase (Mg-Mn-Fe variation)*

303 Garnet surfaces adjacent to biotite-plagioclase grain boundaries display similar
 304 characteristics to those adjacent to biotite-quartz boundaries with sharp
 305 exponential Mg-compositional gradients mirrored by more gradual Mn- and Fe-
 306 compositional profiles (Fig. 7). The Mg variation occurs over a very similar scale
 307 ($\sim 10 \mu\text{m}$) along the garnet-plagioclase boundary to the garnet-quartz
 308 boundaries.

309 *Garnet-muscovite-quartz (Mg-Mn-Fe variation)*

310 Garnet adjacent to muscovite typically has a composition that is intermediate
 311 between that adjacent to quartz and that adjacent to biotite. Quartz-muscovite
 312 boundaries typically lack systematic differences in garnet composition between
 313 the garnet-muscovite interface and the garnet-quartz interface. However, there
 314 are changes in the garnet composition that are directly linked to the position of
 315 the muscovite-quartz boundary itself. Locally elevated Mn-contents of the garnet
 316 surface are associated with quartz-muscovite-garnet triple junctions (Fig. 8).
 317 This geochemical anomaly is matched by lower Mg-contents, and this typically
 318 has a less smoothed sharper anomaly. The wavelength of the high Mn-, low Mg-
 319 anomaly varies from $20 \mu\text{m}$ (Fig. 8c) to $5 \mu\text{m}$ (Fig. 8d) between different grain
 320 boundaries and may be linked to the orientation of the boundary or the
 321 orientation of the muscovite lattice. These anomalies may display an asymmetry
 322 with geochemical variation in profiles along the muscovite-garnet boundary
 323 occurring over longer distances in comparison to that along the quartz-garnet

boundary. This is especially the case when the traverse is parallel to the cleavage of the muscovite (Fig. 8c). Apart from the changes in composition linked to the position of the matrix grain boundary itself, the garnet surface along quartz-muscovite interfaces appears to be characterized by smooth longer wavelength geochemical variations (e.g. Mg in Fig. 8d).

Ca variation on the garnet surface

The Ca-contents of the garnet surface show the least systematic variation with evidence of a lack of equilibrium characterized by smooth, long-wavelength, low-amplitude variations in composition (Figs. 6a & 7). These variations occasionally appear related to adjacent grain boundaries with sharp transitions over a distance of $\sim 10 \mu\text{m}$ (Fig. 6a) but in most traverses the Ca minima and maxima that occur on the garnet surface are unrelated to either the position of the grain boundaries in the matrix or the nature of the adjacent phase (Fig. 7). Garnet adjacent to plagioclase shows no obvious influence of exchange of Ca but, as with some other boundary types, gradual variation in the Ca content is preserved in garnet (Fig. 7).

340

INTERPRETATION

Smooth compositional profiles are retained in the interior of the garnet porphyroblasts and are interpreted as growth zoning with partial modification through volume diffusion (e.g. Kohn, 2003). Low Mg garnet rim compositions record post peak metamorphic exchange with adjacent matrix biotite (e.g. Tracy, 1982; Kohn & Spear, 2000). The presence of smoothly curving compositional profiles is indicative of exchange via volume diffusion rather than fluid-related chemical modification via coupled dissolution-reprecipitation processes. Such

349 garnet zoning is identical to that reported by Dempster (1985) and Viète *et al.*
350 (2011) from this area and confirms the localized retrograde exchange between
351 garnet and some other matrix phases.

352 **Intergranular diffusion along garnet surfaces**

353 Similarly smooth compositional profiles are recorded by the garnet surfaces,
354 which have partially equilibrated through intergranular diffusion.

355 Different elements show different diffusion profiles on the garnet grain
356 boundaries; different boundary types show different diffusion profiles; and, the
357 diffusion profiles may be partially dependent on the orientation of the boundary
358 or the phyllosilicate lattice. Some profiles show chemical variation directly
359 linked to the position of the immediately adjacent grain boundary (e.g. Mg and
360 Mn across the quartz-biotite boundaries). Other parts of the garnet surface show
361 chemical variation that is apparently not systematically linked to the
362 immediately adjacent grain boundaries (e.g. some Ca variation). Generally the
363 consistent profiles displayed by the same boundary types argues that diffusive
364 movements are subject to the same controls and hence that the intergranular
365 regions (cf. Brady, 1983) themselves have similar properties irrespective of their
366 orientation or the crystal lattice orientation.

367 Adjacent to Fe-oxides within the rock, rather than the grain boundary films, the
368 garnet surface is depleted in Fe and enriched in Al (Fig. 5). This may reflect
369 exchange in the trivalent lattice site of garnet. This exchange may be associated
370 with retrograde processes, as the Fe-oxide in this instance has a rather ragged
371 texture characteristic of formation during late alteration. The geochemical
372 boundaries of this modified garnet are exceptionally sharp and this implies low

373 rates of Al and Fe³⁺ intergranular diffusion (cf. Carmichael, 1969) and sluggish
374 movement relative to rates of Fe²⁺, Mg, Mn, and Ca grain boundary diffusion.
375 Mg variation on the garnet surface typically records the most complete and
376 clearly defined diffusion profiles associated with adjacent grain boundaries,
377 especially in the vicinity of biotite. As a consequence these are most amenable to
378 estimating intergranular diffusion rates. Mg contents may plateau at different
379 values along some garnet-quartz boundaries (Fig. 7). Low Mg contents of garnet
380 adjacent to quartz are relatively unusual (Fig. 6c) and may be caused by
381 interference with other diffusion profiles generated at grain boundaries in the
382 matrix near to the studied interfaces. Background gradients in Mg content of
383 garnet along and across muscovite-quartz grain boundaries may also reflect
384 proximity to biotite (Fig. 8c). Mn and Fe variation is typically inversely correlated
385 with the Mg contents, but Mn may be locally decoupled. Associated with the
386 geochemical anomalies at the quartz-muscovite-garnet triple junctions, Mn may
387 record relatively efficient intergranular diffusion in comparison to Mg. Thus in
388 comparison to Mg, Mn profiles have shallower gradients especially between
389 quartz and garnet (Fig. 8). The surface chemistry implies that Mn is transported
390 preferentially to the garnet, perhaps sourced from nearby biotite, along grain
391 boundaries between quartz and muscovite. This is apparently at odds with
392 suggestions of relatively sluggish intergranular diffusion of Mn (Carlson *et al.*,
393 2015a).

394 Ca shows complex long-wavelength, low-amplitude variations on the garnet
395 surfaces and lacks the well-defined diffusion profiles that characterize Mg, Fe
396 and Mn distribution. Adjacent to matrix plagioclase boundaries, garnet lacks
397 evidence of diffusive exchange of Ca, and the geochemical profiles along these

boundaries are similar to the garnet-quartz boundaries. Most Ca variation is unrelated to the nature of the immediately adjacent mineral phase. Some garnet surfaces preserve low amplitude Ca-anomalies (both negative (Fig. 9) and positive (Fig. 7, 8)) that may be broadly correlated with the positions of the triple junctions of mineral boundaries. Although the long wavelength variations point to effective Ca transport relative to other cations, there is a lack of equilibrium, and variation is typically not related to either the positions of grain boundaries or the nature of the matrix phase. This could in theory represent:

a) a *partial late equilibration with a Ca-bearing fluid phase* (Carlson *et al.*, 2015a) with relatively effective transport processes. Hence as with the behaviour of Mg and Mn, this may be indicative preferential transport Ca along some of the triple junction grain boundaries. Potentially this could be linked to the presence an interconnected fluid network (Watson & Brenan, 1987) locally delivering cations to and from the garnet surface at the triple junctions. The triple junction-related effects are not present on all such boundaries implying either a crystallographic or structural control on the effectiveness of transport in these zones. Individual planar interfaces show just as much variation in Ca content as the apparent triple junction effects, which implies that transport processes at these interfaces was independent of the nature of the grain interface. This model fails to explain how both peaks and troughs in Ca content of the garnet surface could be generated on apparently identical planar grain boundaries and is not compatible with the presence of fluid on such boundaries. It is also thought unlikely that fluids would be present during the retrograde cooling.

423 b) a *compositional variation established during the initial growth of the*
 424 *garnet*. A general prograde surface heterogeneity may in part be a
 425 reflection of the relatively distal nature of Ca sources in the pelites and
 426 consequently low diffusive flux. This contrasts with the behaviour of Mg,
 427 Fe and Mn, which all exchange directly across interfaces with the
 428 relatively abundant ferro-magnesium minerals in the matrix. These
 429 elements are also readily available due to efficient volume diffusion
 430 within nearby biotite, which doesn't act as a rate limiting step for
 431 transport. Ca is only present in a few other phases in pelites, such as
 432 plagioclase, which doesn't appear to be involved in exchange reactions,
 433 and apatite which is only present as an accessory phase. Hence larger
 434 transport distances are required for Ca equilibration and the Ca contents
 435 of garnet surfaces are perhaps most likely to be difficult to equilibrate in
 436 pelites. Consequently the Ca variation is believed to represent a prograde
 437 growth feature established at higher temperatures than the retrograde
 438 diffusion profiles that characterise the other divalent cations. This would
 439 explain low amplitude, long wavelength variations that are decoupled
 440 from the present geometry of adjacent matrix phases.

441 Garnet porphyroblasts may preserve evidence for sluggish Ca intergranular
 442 diffusion (Chernoff & Carlson, 1997) but other studies demonstrate preferential
 443 equilibrium at low temperatures relative to the other major cations. The latter
 444 has been linked to increased solubility of Ca in the presence of a fluid phase
 445 (Carlson *et al.*, 2015a). In the absence of such fluids, Ca equilibration may not
 446 occur until upper amphibolite facies conditions (Carlson, 2002). Intergranular
 447 diffusivity may drop by orders of magnitude if a network of fluids is absent

448 (Brenan, 1993). Fluids are likely to be largely absent during the establishment of
449 the retrograde diffusion profiles. However it seems that Ca distribution will be
450 more sensitive to other factors such as the presence of fluids and proximity to
451 sources of Ca in pelites. In general the relative rates of intergranular diffusion
452 along garnet-matrix interfaces suggest that Mn movement is more rapid than Mg
453 and Fe, and intergranular equilibration of Ca is most difficult in this rock.

454 **Estimation of intergranular diffusion rates**

455 To enable the calculation of absolute values for diffusion rate it is crucial to have
456 well constrained thermal histories of the samples. Such histories are not well
457 established for the prograde, post-garnet growth histories of these Dalradian
458 rocks, although evidence would suggest cooling histories were rapid (Dempster,
459 1985; Oliver *et al.*, 2000). A range of different diffusivity data exists for volume
460 diffusion in garnet with general agreement that Fe, Mg, and Mn diffusion
461 coefficients and activation energies are broadly similar and Ca diffusion is
462 significantly slower (Ganguly *et al.*, 1998; Chakraborty & Ganguly, 1992; Carlson,
463 2006; Vielzeuf *et al.*, 2007). However there remains uncertainty in the influence
464 of garnet composition (Ganguly, 2010). The uncertainty associated with volume
465 diffusion rates (Carlson, 2006; Ganguly, 2010) coupled to uncertainty in the
466 detailed thermal histories of these rocks precludes reliable absolute estimates of
467 intergranular diffusion rates in our study. However diffusion profiles may be
468 used to give an indication of relative diffusion rates through a comparison of
469 different element diffusion profiles and comparison with volume diffusion
470 profiles. The compositional changes on these surfaces are more gradual along
471 garnet-biotite boundaries than garnet-quartz boundaries. Hence intergranular
472 diffusion is faster along garnet-biotite boundaries.

473 The Mg-volume diffusion profiles can be compared to the Mg-profiles for
474 intergranular diffusion preserved on the garnet surfaces. Such an approach may
475 not be justified for all cations given that volume diffusion requires simple
476 multicomponent exchange (Borinski *et al.*, 2012), whereas the grain boundary
477 composition profiles potentially reflect more diffusive fluxes controlled by the
478 proximity to sources of cations from other phases (Mueller *et al.*, 2010). In
479 addition the grain boundary profiles themselves will represent an integrated
480 composition of garnet from a few microns below the surface rather than the
481 diffusion pathway itself. It is also important to note that Mg-diffusion in garnet
482 does not act independently and must be coupled to diffusion of at least one other
483 divalent cation. Typically in the measured profiles, Mg exchange is maintained by
484 both Mn and Fe variation. However the comparison undertaken here involves
485 volume diffusion profiles in garnet adjacent to biotite with profiles along the
486 garnet grain boundaries (with either quartz or plagioclase) away from adjoining
487 biotite. As such the biotite provides a proximal source/sink for Mg-Fe. This
488 approach (Fig. 9) demonstrates that Mg-diffusion profiles along the garnet-
489 quartz grain boundaries and garnet-plagioclase grain boundaries are very
490 similar to those of volume diffusion profiles within the garnet. These
491 observations from polyphase boundaries contrast with experimental studies of
492 Yb-Y coupled diffusion in garnet pairs (Marquardt *et al.*, 2011) in which grain
493 boundary transport is reported to be nearly 5 orders of magnitude faster than
494 volume diffusion. In terms of the total amount of chemically modified garnet (i.e.
495 the steepness of the diffusion profiles), Mg-diffusion along the plagioclase-garnet
496 interface appears to be slowest, followed by the volume diffusion of Mg within
497 garnet, whilst the rates of Mg-diffusion along the quartz-garnet interface are the

498 most effective of those presented in figure 9. Fitting diffusion curves to the
499 profiles along the garnet-biotite boundaries are harder because of the relative
500 lack of clearly defined diffusion profiles along the measured boundaries.
501 However given the available data on the relatively short “clean” traverses that
502 we have measured, we estimate that diffusion distances are at least 4 times
503 greater than those of volume diffusion within garnet.

504

505 **IMPLICATIONS**

506 Grain boundaries are the key zones through which elemental transport occurs
507 and metamorphic equilibration is facilitated. Hence intergranular diffusion has
508 been suggested as “the most common impediment to equilibration” during
509 metamorphism (Carlson, 2002) and as such is a key control on metamorphic
510 processes. Snapping rocks along the mineral grain boundaries reveals important
511 textural (Dempster *et al.*, 2006; Lawther & Dempster, 2009) and chemical
512 information regarding these zones. This study focuses on the chemical variations
513 along garnet grain boundaries and represents the first time that intergranular
514 diffusion profiles have been observed on natural mineral surfaces. It crucially
515 emphasizes the importance of a different analytical approach to petrological
516 research, one that is not based on thin section analysis.
517 Our study reveals that intergranular diffusion on garnet grain boundaries is
518 typically slow, occurring at similar rates to volume diffusion in garnet. Previous
519 studies of intergranular diffusion have suggested similarly sluggish transport
520 along grain boundaries (Florence & Spear, 1995; O’Brien, 1999) on the basis of
521 disequilibrium recorded in thin sections, although the role of fluid on the grain
522 boundary interfaces in these scenarios has been questioned (Dohmen &

523 Chakraborty, 2003). However, many investigations suggest that intergranular
 524 diffusion rates are faster by one or more orders of magnitude than rates of
 525 volume diffusion (e.g. Farver & Yund, 2000; Milke *et al.*, 2001). There is
 526 significant variation in diffusion rates of different cations for different grain
 527 boundary types. These types range from asymmetrical isolated boundaries
 528 (Mishin & Razumovskii, 1992) to fast grain boundaries (Eiler *et al.*, 1992) that
 529 respectively apply to the garnet-biotite-quartz boundaries and the movement
 530 along the triple junction network in our study.

531 Diffusion of divalent cations along boundaries involving quartz and plagioclase is
 532 slow in comparison to movement along garnet-biotite boundaries. The lack of
 533 Ca-exchange between plagioclase and garnet is notable. If as seems likely such
 534 behaviour also characterizes low temperature prograde metamorphism it would
 535 potentially compromise the application of geobarometers based on such
 536 equilibria (Holdaway 2001; Wu *et al.*, 2004). As such plagioclase appears to
 537 behave as a closed system in these conditions, much in the way that muscovite
 538 does in low-grade metamorphic environments (Dempster, 1992).

539 Crystallographic orientation also appears to play a role in the rate of
 540 intergranular transport with faster Fe-Mg diffusion parallel to the [001] cleavage
 541 in both muscovite and biotite. This is similar to, although less marked than, that
 542 reported for anisotropic volume diffusion in biotite (Usuki, 2002). Although
 543 preferential transport along some triple junctions of grain boundaries is
 544 recorded, ineffective transport along the planar boundaries limits the impact of
 545 these fast pathways in maintaining equilibrium on the garnet surface. Because of
 546 the sluggish nature of intergranular diffusion, garnet surfaces are heterogeneous
 547 and their chemistry is as much controlled by the availability, and proximity to

548 sources of cations, as it is to the rates of diffusion (Mueller *et al.*, 2010). The
549 abundance of Fe- and Mg- bearing phases and their availability in the matrix of
550 pelites means that equilibrium is more readily achieved for these cations during
551 garnet growth. However for typical pelites it is predicted that Ca is most likely to
552 preserve heterogeneities on the garnet surface due to the lack of available Ca
553 sources in the matrix.

554 Given the sluggish nature of volume diffusion in garnet, similarly slow grain
555 boundary diffusion is likely to be a significant kinetic influence during prograde
556 metamorphism, especially that associated with relatively low temperature
557 conditions. The low temperatures of retrograde exchange favour grain boundary
558 chemical heterogeneity, and the inferred lack of fluids, due to the lack of
559 significant retrogression, may also reduce length scales of equilibration.

560 Prograde metamorphism is characterized by additional factors that will promote
561 equilibrium, particularly fluid release by dehydration reactions (Bell & Cuff,
562 1989) and penetrative deformation (Dempster & Tanner, 1997). The latter will
563 potentially allow fluid access and whilst fluid presence may enhance diffusion in
564 intergranular zones, some studies suggest this may not be the case (Farver &
565 Yund, 1995). High permeability of metamorphic basement rocks may be short-
566 lived, even during prograde events (Yardley & Valley, 1997; Yardley, 2009),
567 relative to the time-scales over which diffusion is effective. Consequently the
568 sluggish intergranular diffusion recorded during retrograde processes may also
569 be applicable to prograde metamorphism.

570 Low rates of grain boundary diffusion will inhibit garnet growth, and equilibrium
571 at the growing edge of the porphyroblast. The initial stages of garnet growth at
572 lower temperatures should be more sensitive to the nature of adjoining phases

573 than during growth at higher temperature. As such overprint zoning (Yang &
 574 Rivers, 2001; Hirsch *et al.*, 2003) should be more commonly observed in
 575 examples of low temperature porphyroblast growth. Reports of overprint zoning
 576 in garnet porphyroblasts from the well-documented Harpswell Neck locality
 577 (Hirsch *et al.*, 2003) point to sluggish intergranular diffusion of Mn and Mg
 578 during the early stages of prograde growth. Published X-ray maps of garnets
 579 from this locality show that locally high Mn and low Mg contents are associated
 580 with areas of relatively few quartz inclusions within the poikiloblasts (Hirsch *et*
 581 *al.*, 2003; Carlson *et al.*, 2015a). If this observation is correct it suggests that
 582 sluggish diffusion in the quartz-rich parts of the matrix inhibited chemical
 583 equilibration at the garnet margin. Other studies also appear to show a
 584 correlation between a lack of equilibrium in the zoning profiles and inclusion-
 585 rich areas of garnet porphyroblasts (O'Brien, 1999). Perchuk *et al.* (2009)
 586 suggest that Mg-Fe volume diffusion may be slowed by the presence of inclusions
 587 in comparison to inclusion-free garnet and our results suggest that the grain
 588 boundaries of these inclusions will not provide an effective transport system
 589 within porphyroblasts.
 590 Intergranular diffusion is strongly controlled by the nature of the grain boundary
 591 and so garnet growth in layers with different modal proportions will occur at
 592 different rates. Initial growth of garnet may occur in finer grained chlorite-rich
 593 lithologies, and as such much of what garnet requires may be locally accessible.
 594 However disequilibrium garnet growth is reported in quartz-rich layers in
 595 comparison to more equilibrium growth and larger porphyroblasts in mica-rich
 596 domains (Spear & Daniel, 1998; Carlson *et al.*, 2015a). This would be consistent

597 with the more effective intergranular diffusion on phyllosilicates grain
598 boundaries in comparison to quartz grain boundaries.
599 The sharpest compositional changes on the garnet surfaces are associated with
600 garnet-Fe-oxide, garnet-quartz and garnet-plagioclase grain boundaries. Hence
601 intergranular diffusion will be sluggish during garnet growth adjacent to these
602 phases. These minerals typically form the most common type of inclusion in
603 garnet porphyroblasts and this may be a reflection of the difficulty in moving
604 components along these grain margins (Yang & Rivers, 2001). Hence locally
605 sluggish diffusion on the boundaries of garnet may have a fundamental control
606 on the textures of porphyroblast growth.

607

608 **SUMMARY**

609 New techniques of porphyroblast surface analysis provide a means to quantify
610 rates of intergranular diffusion on variety of different grain boundaries for a
611 range of elements. Such rates when combined with established techniques that
612 monitor scales of disequilibrium in garnet zoning and the distribution of
613 porphyroblasts provide a powerful combination of tools with which to assess of
614 the controls on metamorphic equilibration and porphyroblast growth. Models of
615 metamorphic behaviour based on relatively slow intergranular diffusion rates
616 would predict chemical disequilibrium of the garnet surface during
617 porphyroblast growth with overprint zoning of cations dominant at low
618 temperatures and in relatively quartz-rich lithologies. Divergence from such
619 behaviour would be expected if intergranular diffusion is not the dominant
620 kinetic impediment to the establishment of metamorphic equilibrium. Hence a
621 lack of correlation between porphyroblast size and lithology may point to the

importance of other kinetic factors. Equally an apparent lack of matrix-related “overprint” growth zoning in garnet argues that either volume diffusion has smoothed out such growth features or that deformation (Bell & Hayward, 1991; Dempster & Tanner, 1997) and fluids (Rubie, 1986; Jamtveit *et al.*, 1990; Pattison *et al.*, 2011; Carlson *et al.*, 2015a) may be of primary importance in enhancing elemental mobility during prograde metamorphism.

628

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632

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845
846 **FIGURE LEGENDS**

847 *Figure 1.* Transmitted light photomicrograph showing rounded morphology of
848 garnet porphyroblast containing Fe-oxide and quartz inclusions aligned at an
849 angle to the external muscovite-rich fabric. Note the presence of brown Fe-
850 staining on some fractures and grain boundaries.

851
852 *Figure 2.* X-ray maps for (a) Mg, (b) Mn, and (c) Ca variation within a garnet
853 porphyroblast. (d) Mg concentration traverses (X_{prp}) through the edge of a
854 garnet porphyroblast adjacent to biotite. Traverse is oriented perpendicular to
855 the garnet edge.

856
857 *Figure 3.* Low magnification electron microscope images of the surface of a
858 garnet porphyroblast and its adjacent matrix. (a) Secondary electron image of
859 garnet, dashed area shows the high magnification area shown in (b) and (c); (b)
860 High magnification secondary electron image of part of the garnet surface shown
861 in (a); (c) Backscattered electron image of the garnet surface shown in image (b);
862 (d) Backscattered electron image of the matrix surface adjacent to the garnet
863 surface shown in (b), this image is reversed so that grain boundary positions in
864 the matrix may be matched to positions on the garnet surface. Labels (Kretz,
865 1983) identify the matrix minerals.

866

867 *Figure 4.* Secondary electron image of garnet porphyroblast surface showing flat
 868 grain boundary areas, prominent ridges associated with growth steps and
 869 variable development of irregular Fe-oxide coating, shown by brighter patches
 870 (marked by arrows). Wherever possible chemical traverses of the garnet
 871 surfaces avoided these topographic features and later deposits.

872

873 *Figure 5.* X-ray maps of (a) Fe and (b) Al composition of garnet porphyroblast
 874 surface. Positions of matrix phase boundaries adjacent to the garnet surface are
 875 shown in figure (a) with outlined in white dashed line the position of an Fe-oxide
 876 grain within the adjacent matrix. Note that the high Fe content of the garnet
 877 surface adjacent to biotite in the lower left hand area of the X-ray map is due to a
 878 locally thick Fe-surface coating.

879

880 *Figure 6.* Major element cation traverses (in % of end member composition) of
 881 garnet surfaces across the position of quartz-biotite grain boundaries (shown by
 882 dashed red lines). Pairs of backscattered images of the garnet and the reversed
 883 image of the adjacent matrix show the positions of each traverse. Each pair of
 884 images is positioned above the corresponding compositional traverse. Position of
 885 the matrix grain boundary is marked on the line of traverse by the red dashed
 886 line. (a) Traverse (A-B) perpendicular to orientation of biotite cleavage; (b)
 887 Traverse (C-D) perpendicular to biotite cleavage trace. "Noise" in sections of this
 888 traverse adjacent to biotite reflect the presence of particularly coarse Fe-oxide
 889 surface deposits on the garnet-biotite boundary; (c) Traverse (E-F) parallel to

890 the orientation of the biotite cleavage trace. Note this traverse was analysed over
891 a shorter time period and as a consequence errors are a little larger.

892

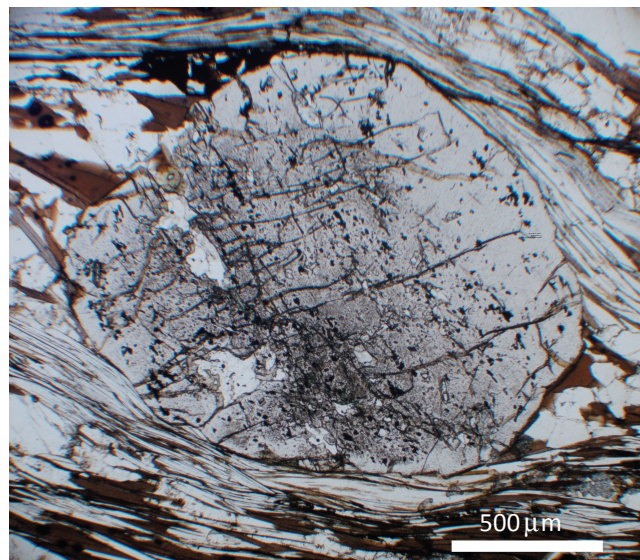
893 *Figure 7.* Major element cation traverses (in % of end member composition) of
894 garnet surface across an adjacent biotite-plagioclase grain boundary. Location of
895 traverse (G-H) shown in backscattered electron images of the garnet and the
896 reversed image of the adjacent matrix.

897

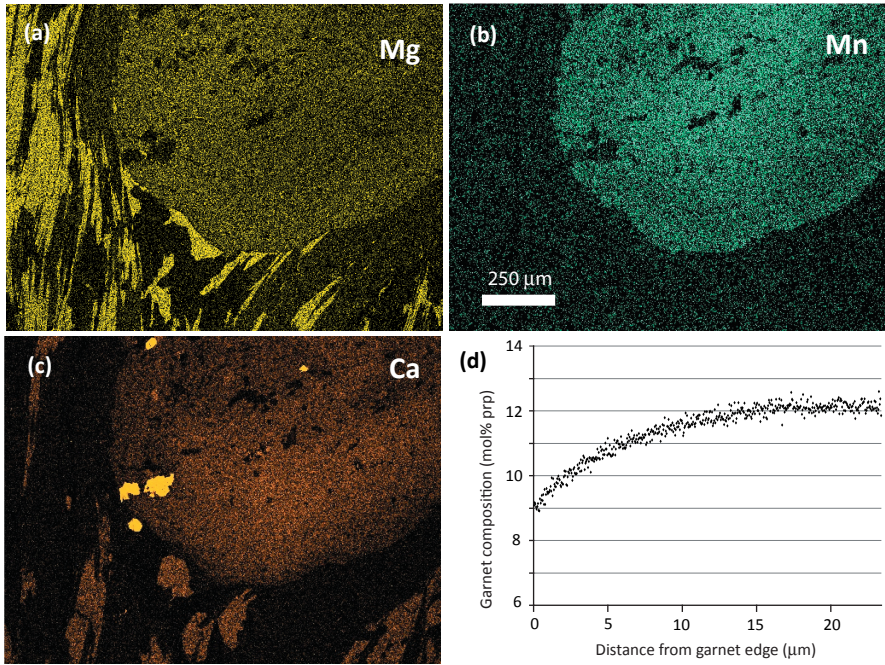
898 *Figure 8.* Major element cation traverses (in % of end member composition) of
899 garnet surface across adjacent quartz-muscovite grain boundaries. Location of
900 traverses shown in backscattered electron images of the garnet and the reversed
901 image of the adjacent matrix. (a) Traverse (I-J) parallel to the orientation of the
902 muscovite cleavage; (b) Traverse (K-L) perpendicular to the muscovite cleavage.

903

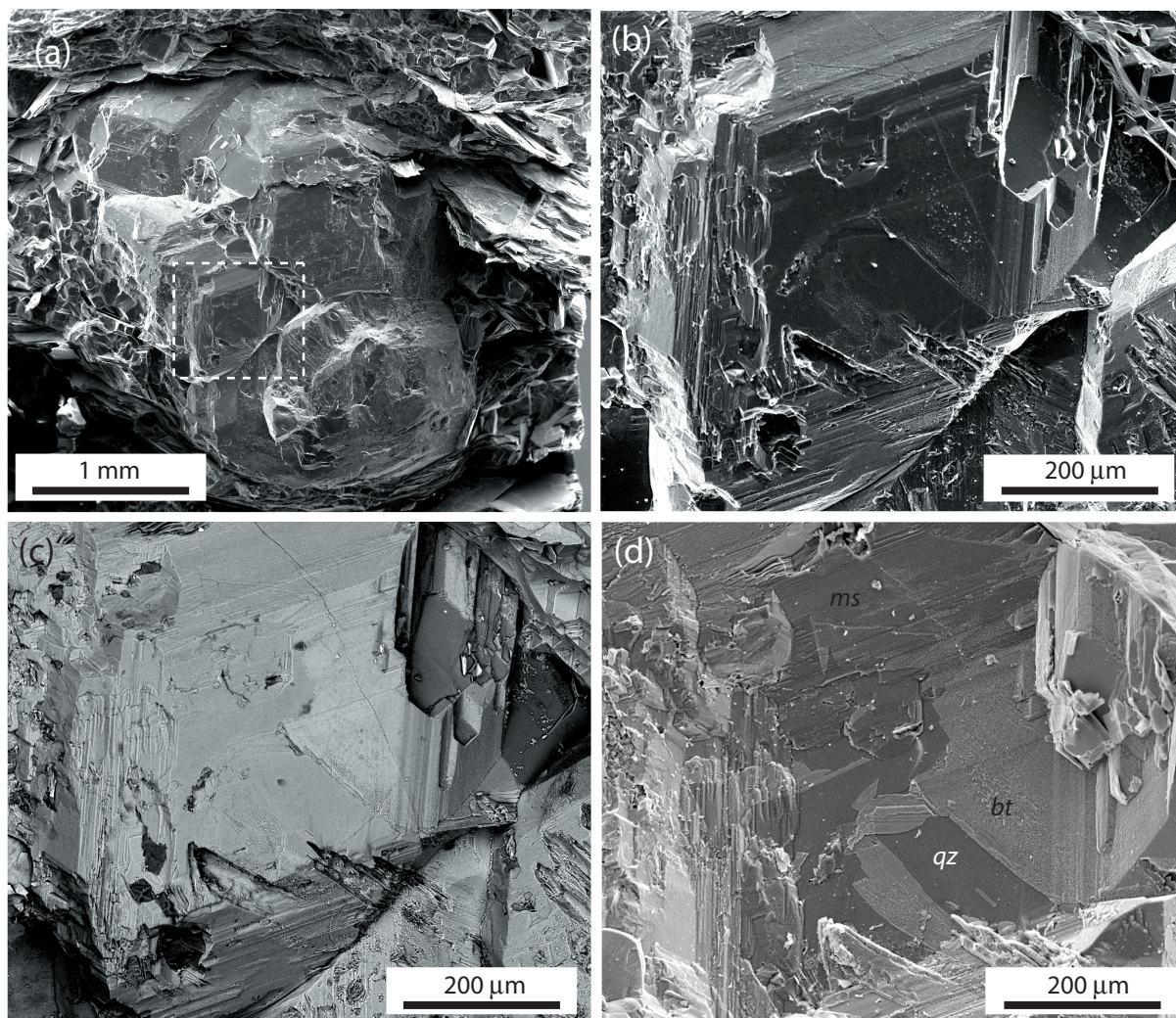
904 *Figure 9.* Comparison of Mg diffusion profiles for garnet grain boundaries with
905 volume diffusion profile (from Figure 2d) within porphyroblast (red dashed
906 line). Garnet surface diffusion profiles for garnet-quartz boundary (green dotted
907 line; A-B in figure 6) and garnet-plagioclase boundary (blue, dashed-dotted line;
908 G-H in figure 7). Inset shows schematic positions of diffusion profiles (coloured
909 arrows) in thin section view of grain boundaries with all traverses shown
910 oriented away from a garnet-biotite boundary. Volume diffusion profile shown is
911 truncated 2 μm from the grain edge due to potential uncertainty in the true
912 composition caused by the volume of interaction with the electron beam.



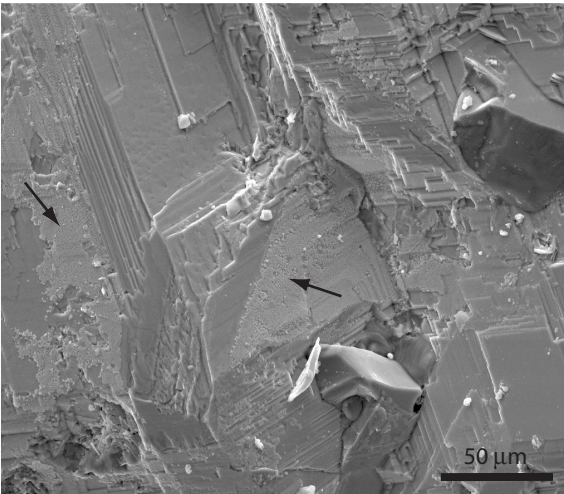
Dempster et al (Fig 1)



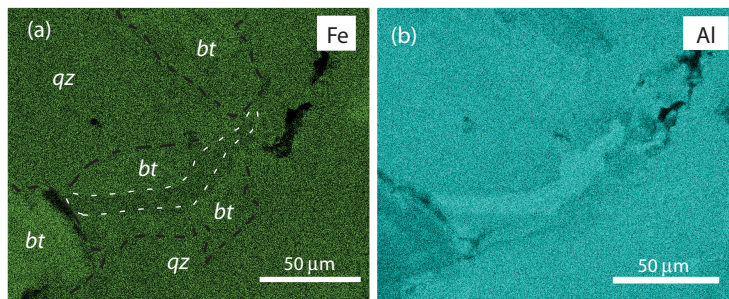
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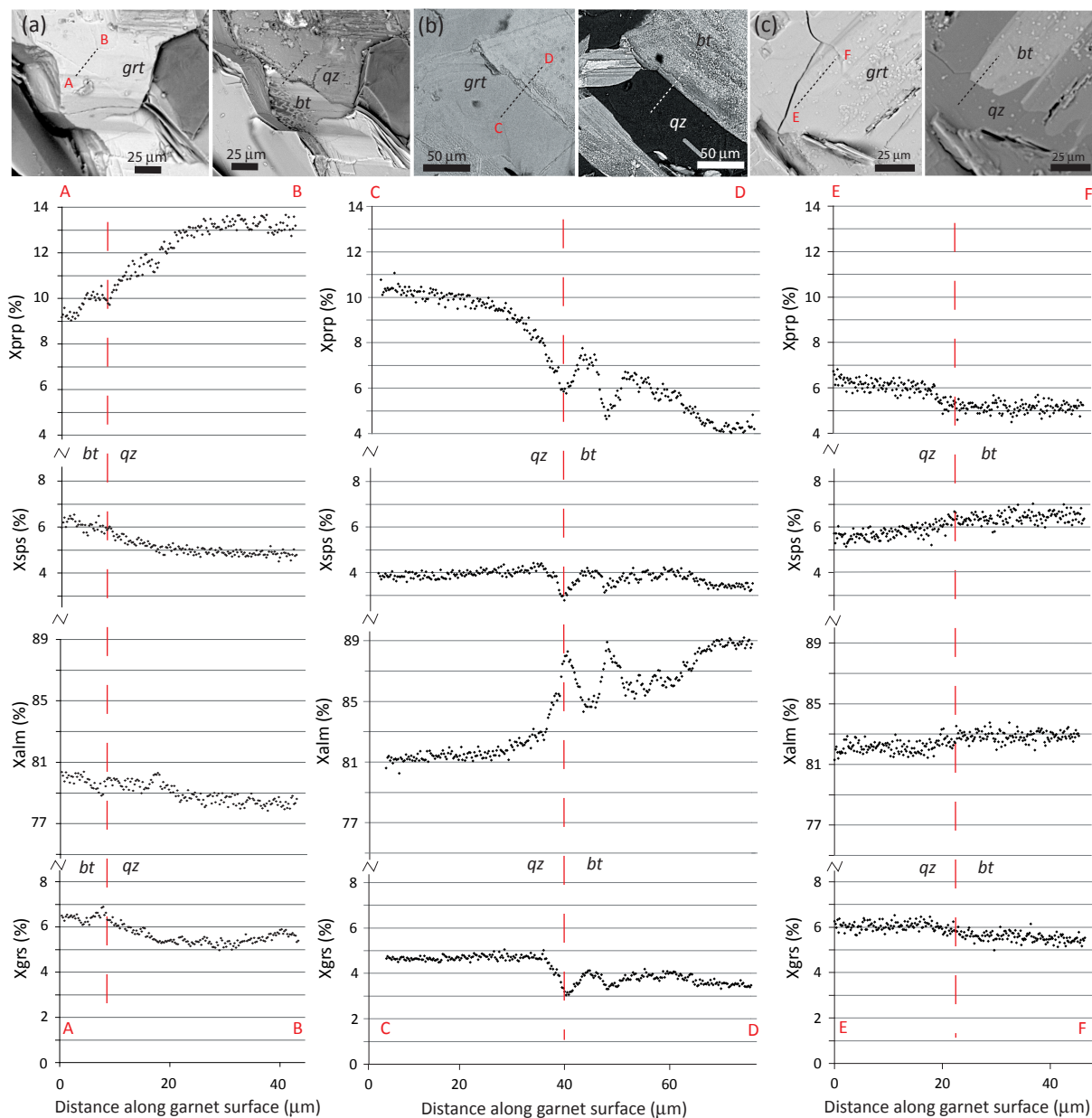
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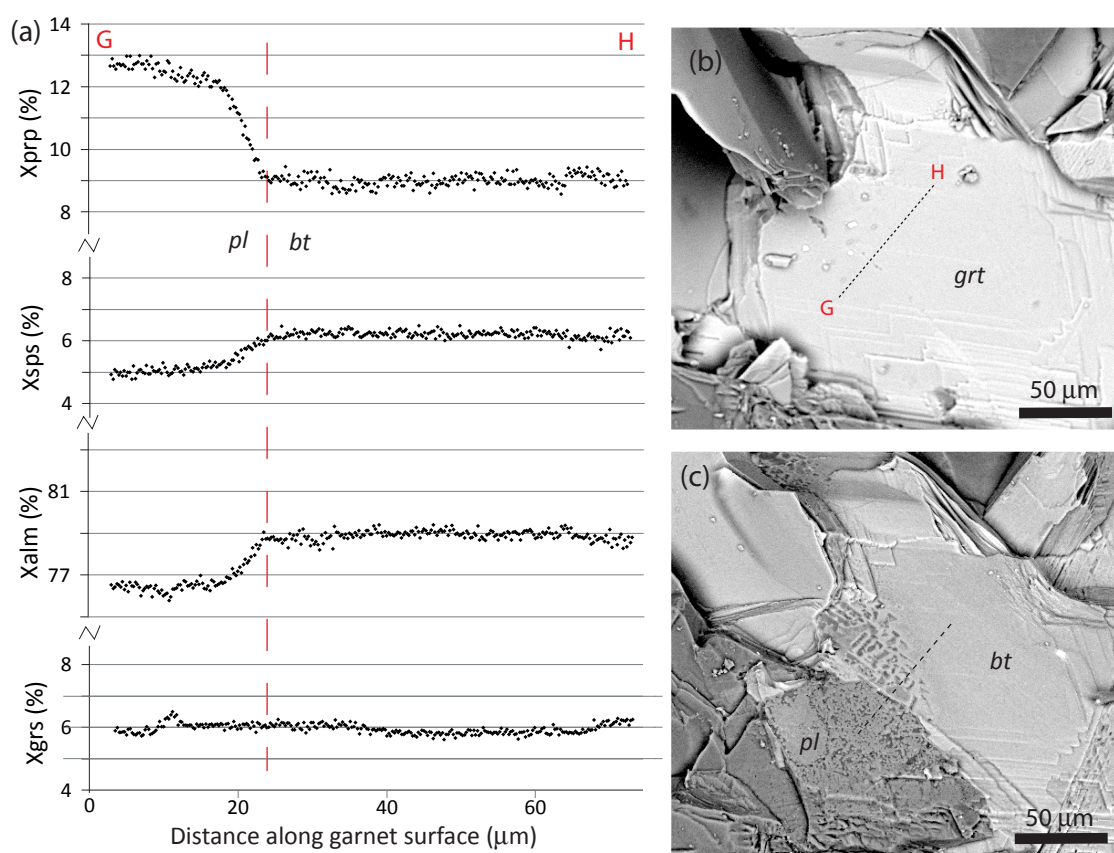
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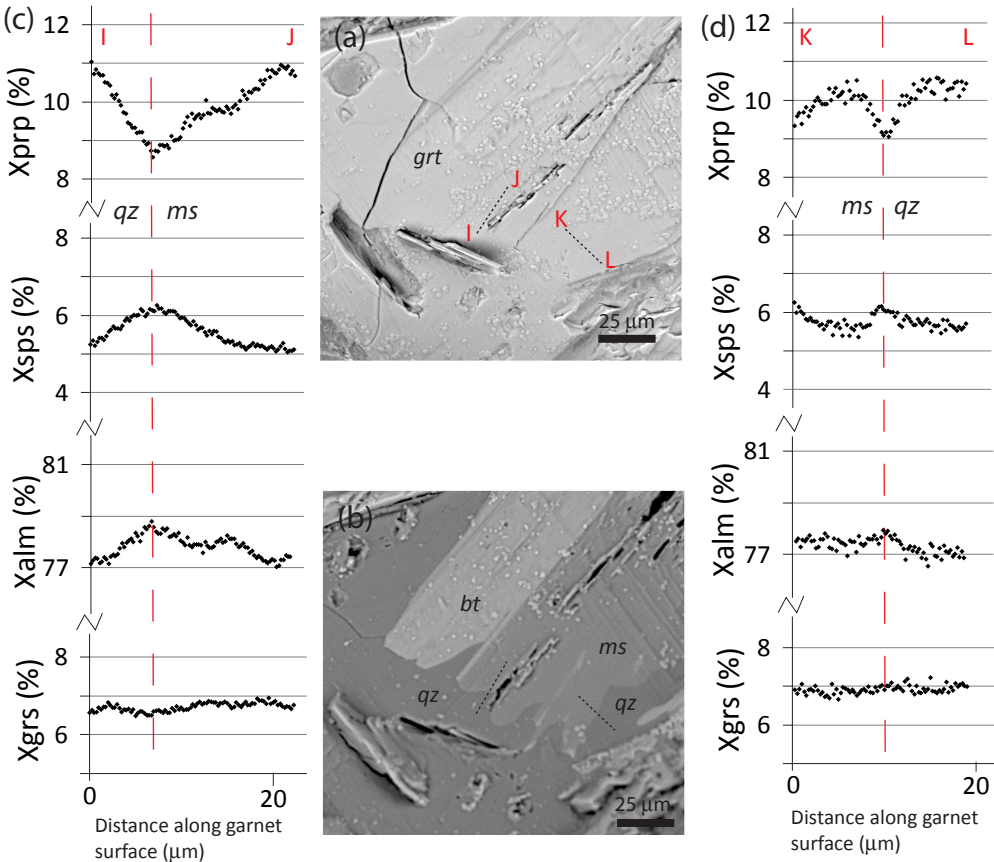
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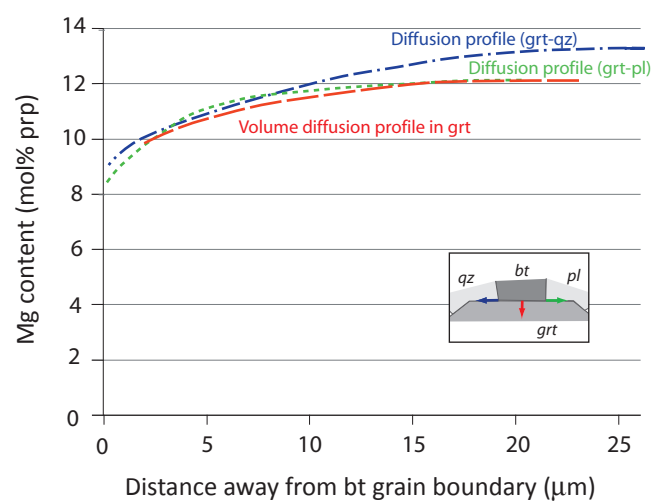
Dempster et al (Fig 6)



Dempster et al (Fig 7)



Dempster et al (Fig 8)



Dempster et al (Fig 9)