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LOW-CA PYROXENES IN THE NWA 998 NAKHLITE METEORITE: REACTIVE PRODUCTS OF OLIVINE-PLAGIOCLASE MINERAL ASSEMBLAGE.

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Introduction: NWA 998 is a unique nakhlite meteorite as it has the texture and mineralogy typical of terrestrial gabbro [1]. It is fully crystallized with abundant augite and olivine forming the framework of the rock and an inter-cumulus assemblage comprised of plagioclase and K-feldspar, low-Ca pyroxene, Timagnetite, Cl-apatite and Fe-sulfide [2].

Another characteristic of NWA 998 is the significant presence of low-Ca pyroxenes, most of which are in chemical and textural disequilibrium with the other igneous minerals [3]. The formation of low-Ca pyroxenes in some of the nakhlites is poorly understood but may contain a record of re-equilibration reactions in response to changing physico-chemical conditions during late and/or post igneous processes [4].

Methods: Polished surfaces of slices of NWA 998 were examined using a Carl Zeiss Sigma analytical SEM with Oxford microanalysis for back-scattered electron imaging, energy dispersive X-ray element mapping and quantitative mineral analysis.

Results and discussion: Petrographic observations provide evidence that low-Ca pyroxenes developed along olivine-plagioclase grain boundaries and have progressively intruded into olivine, plagioclase and adjacent augite. Remnants of both olivine and plagioclase and inclusions of apatite are commonplace within large (up to 100 μ m long) low-Ca pyroxene grains. The crystallization of low-Ca pyroxene has locally erased the primary texture and structure of the rock. Their reactive textures reflect a return of mineralogical equilibrium from early stable mineral associations.

The low-Ca pyroxene crystals display compositional variations according to their petrological settings suggesting some degree of chemical interdependency between the reactant hosts and their associated products. The low-Ca pyroxenes topographically replacing olivine correspond to "pure" orthopyroxene (with or without vermicular magnetite), while those replacing plagioclase are pigeonite with augite lamellae.

Conclusions: The crystallization of low-Ca pyroxene resulted from mineralogical instability between the formerly juxtaposed olivine and plagioclase feldspar grains. This instability may have been caused by changes in temperature or oxygen fugacity (solid-state transformation) or because of the presence of a reactive silicate melt along grain boundaries. The residual melts in NWA 998 show enrichment trend towards more silica and alkali rich compositions, as underlined by the occurrence of both K-feldspar and "frozen" alkali-silica-rich material filling corner pores. Silicate melt droplets may have triggered the reactions, which need to be quantified through volume mass balance calculation. However, petrographic observations suggest that solid-state (sub-solidus) reactions have been the main mechanism for the development of low-Ca pyroxene within the reactive zone.

References: [1] Treiman A. H. 2005. Chemie der Erde 65:203-270. [2] Irving A. J. et al. 2002. Meteoritics and Planetary Science 37:A70. [3] Treiman A. H. and Irving A. J. 2008. Meteoritics and Planetary Science 43:829-854. [4] Harvey R.P. and McSween H. Y., Jr., 1992. Geochimica et Cosmochimica Acta 56:1655-1663.