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# Use of magnesium-silicate-hydrate (M-S-H) cement mixes in 3D printing applications

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## Abstract

This study focused on the development of 3D printed building components incorporating MgO-SiO<sub>2</sub> binders. The effects of parameters such as MgO/SiO<sub>2</sub> and water/binder ratios and superplasticizer dosage were evaluated to produce 3D printable MgO-SiO<sub>2</sub> binders. The assessment of rheological and mechanical properties, which led to an optimum mix that satisfied the printability criteria, was supported by microstructural characterization. The best-performing mix in terms of printability and mechanical performance was characterized via x-ray diffraction (XRD), thermogravimetric analysis/derivative thermogravimetry (TG/DTG) and field emission scanning electron microscopy (FESEM) with energy dispersive spectroscopy (EDS). The selected mix exhibited thixotropic behaviour with an ability to recover ~80% of its original viscosity within 60 seconds of extrusion and a shape retention factor of ~1. The extrudability, shape retention and buildability aspects were successfully demonstrated by printing a 5-layered 3D structure without any deformations. The final product contained brucite and M-S-H as hydration products, which were the main contributor to mechanical performance and microstructural densification.

**Keywords:** 3D concrete printing; MgO; M-S-H; thixotropy; performance; microstructure

## 32 1. Introduction

33

34 Additive and automated manufacturing technologies involving cementitious materials such as 3D  
35 concrete printing have experienced continuous worldwide development over the past years. Digital  
36 concrete has rapidly evolved from an idea to a feasible alternative to conventional technologies  
37 with expanding application areas [1, 2]. The use of 3D printing can shorten the construction  
38 process, enhance resource efficiency, reduce construction waste, enable worker distancing (i.e.  
39 especially important in global pandemics such as COVID-19), and facilitate complex, high-quality  
40 and functional designs that would either be not feasible or too costly with conventional building  
41 methods. One of the aspects that needs to be considered for 3D-printed concrete mixes involves  
42 the development of binders with adequate rheological and stiffening properties to enable their  
43 continuous pumping during the initial stages, followed by shape retention after extrusion [3, 4].

44

45 Along with extrudability (i.e. pumpability), another property that is considered in the mix design  
46 of 3D printing is buildability. Pumpability refers to the effort needed for transferring the material  
47 from the mixer/reservoir to the nozzle; whereas buildability is dependent on the material yield  
48 stress and structural build-up properties [5-8]. These aspects can be assessed by measuring the  
49 yield stress and viscosity of the material. For a given layer to be stable after extrusion, an adequate  
50 yield stress must be reached after deposition, which can be controlled by the flocculation process  
51 [9]. Flocculation refers to the formation of clusters by the growth of cement particles in dispersion,  
52 contributing to the development of yield strength in the material. Accordingly, in order to achieve  
53 a stable filament, the printed material must show the ability to flocculate immediately after its  
54 deposition. In the case of the multi-layer deposition processes, gravity-induced stresses  
55 progressively increase with an increase in the number of layers. During this process, the rate of  
56 structural build-up needs to be controlled by the use of suitable admixtures, without sacrificing the  
57 inter-layer bond strength. These properties can be achieved via the optimisation of the mix design,  
58 resulting in a balance between continuous flowability during extrusion and proper structuration  
59 after deposition, thereby enabling the production of a robust 3D printed component [10].

60

61 Previous studies [11-15] have reported the key factors controlling the fresh properties (e.g.  
62 rheology) of 3D printable cement-based mixes. However, most of the mixtures designed for 3D

63 printing involve the use of Portland cement (PC) as the main binder component. The production  
64 of PC accounts for 5-7% of the global anthropogenic carbon dioxide (CO<sub>2</sub>) emissions. In this  
65 regard, reactive magnesium oxide (MgO)-based cements present some advantages over PC such  
66 as the lower calcination temperatures used during their production in comparison to PC (700°C vs.  
67 1450°C) and their ability to be extracted from waste sources (i.e. reject brine). The combination  
68 of MgO with a silica (SiO<sub>2</sub>) source leads to the formation of magnesium silicate hydrate (M-S-H),  
69 which provides strength within concrete formulations, without requiring any special curing  
70 arrangements such as accelerated CO<sub>2</sub> curing [16, 17]. The characteristics of M-S-H are controlled  
71 by several factors including the type and properties of binder components, mix design (i.e. initial  
72 MgO/SiO<sub>2</sub> ratio, water-to-binder ratio and superplasticizer content) and curing conditions [18, 19].  
73 Previous studies have looked into the effect of different raw materials [20, 21], role of  
74 superplasticizers [22, 23], durability and mechanical properties [17, 24] and microstructural  
75 characterization [25, 26] of the final products in the MgO-SiO<sub>2</sub> binders. With high initial  
76 compressive strengths reaching up to 70 MPa after 28 days of curing [27], MgO-SiO<sub>2</sub> binders have  
77 the potential to be used as structural or non-structural building components. These initially high  
78 strengths enable a higher shape retention and lower filament deformation during 3D printing.  
79 Additionally, due to their intrinsically low pH (~10.5) in comparison to PC (~12.5), MgO-SiO<sub>2</sub>  
80 binders can be utilized in applications involving natural fibre reinforcements [28, 29] and waste  
81 encapsulation [30, 31], which can widen the application spectrum of 3D printed components.

82

83 In line with the global academic and industrial initiatives on enhancing the sustainability of  
84 cement-based mixes, this paper focused on the development of 3D printed building components  
85 incorporating alternative binders such as MgO-SiO<sub>2</sub> binders with potentially lower environmental  
86 impacts. The use of MgO-SiO<sub>2</sub> binders in 3D printing, which has not been studied until now, was  
87 aimed to pave the way for the development of alternative binder systems for 3D printing  
88 applications and provide new insights on the printability aspects of these mixes. This was achieved  
89 via the examination of the key factors controlling the design of 3D printable MgO-SiO<sub>2</sub> binders  
90 such as the rheological properties of materials in the fresh state and printed region based on  
91 thixotropy, slump and flow diameter values. The effects of parameters such as MgO/SiO<sub>2</sub> and  
92 water-to-binder (w/b) ratios and superplasticizer (SP) dosage were assessed in order to produce  
93 sustainable 3D printable mixtures. Out of the formulations investigated in this study, the mix that

94 exhibited the best printability and mechanical performance was further characterized in terms of  
95 its composition (i.e. phase formations) and microstructural development via x-ray diffraction  
96 (XRD), thermogravimetric analysis/derivative thermogravimetry (TG/DTG) and field emission  
97 scanning electron microscopy (FESEM) with energy dispersive spectroscopy (EDS).

98  
99

## 100 **2. Materials and Methodology**

101

### 102 **2.1 Materials and mix preparation**

103

104 Reactive MgO (labelled as “calcined magnesite 92/200”) was provided by Richard Baker Harrison  
105 (UK), whereas microsilica (MS) (labelled as “940U”) was provided by Elkem Materials  
106 (Singapore). The chemical compositions and physical properties of these binder components are  
107 shown in Table 1. The superplasticizer (SP) used in the prepared formulations was sodium  
108 hexametaphosphate (SHMP), provided by VWR (Singapore).

109

110 Table 2 shows all the different mix compositions used in this study. Two different binder  
111 compositions were prepared, in which the MgO:MS proportions were set at 50:50 and 70:30. For  
112 each composition, two different water-to-binder (w/b) ratios (0.4 and 0.5) and SP dosages (1.5  
113 wt.% and 2.5 wt.% of the binder) were used, generating eight mix designs in total. These levels  
114 were determined based on the preliminary extrudability test results and minimum M30 grade  
115 concrete (compressive) strength requirement. Mix preparation initiated by dissolving SHMP in  
116 water to form a superplasticizer solution. After the SHMP was fully dissolved in water, MgO was  
117 added into the SHMP solution while mixing in a Hobart HL200 mixer. MS was then gradually  
118 added to the mix and the mixing continued for 5 minutes, until a thoroughly mixed paste was  
119 obtained.

120

121

### 122 **2.2 Methodology**

123

#### 124 **2.2.1 Slump and flow table**

125

126 The slump value of cement-based mixes is related to their static yield stress, whereas the flow  
127 diameter is related to viscosity [32]. Accordingly, to assess these fresh properties, a conical mould,  
128 as shown in Fig. 1(a), was used for the slump and flow table tests, in accordance with  
129 ASTM C230 [33]. The test procedure started by filling the prepared pastes into the mould, after  
130 which the excess material was taken out and the surface was smoothed with a trowel. This was  
131 followed by the removal of the mould, after which the change in height was measured and reported  
132 as the slump value. The flow table test setup is demonstrated in Fig. 1(b), which was performed in  
133 accordance with ASTM C1437 [34]. To measure the flow values, the flow table was dropped 25  
134 times after the mould was removed and the diameter of the paste was recorded.

135

136

### 137 **2.2.2 Thixotropy (viscosity recovery)**

138

139 Thixotropy, assessed by the viscosity recovery test, is a desirable aspect for 3D printing, referring  
140 to the state of high viscosity at low stress and vice versa. A constant shear rate of  $300 \text{ s}^{-1}$  was  
141 applied for the evaluation of the shear thinning property, while viscosity recovery was assessed  
142 via a three-stage approach, whose details were provided in [35]. The thixotropy of the prepared  
143 mixes was measured by a MCR 101 model rheometer from Anton Paar. The approach adopted in  
144 determining the shear rates and timings used during this 3-step analysis aimed to simulate the  
145 printing process, during which the prepared mix is initially at rest in the hopper, then experiences  
146 high shear during extrusion, followed by its resting on the print bed.

147

148

### 149 **2.2.3 3D printing**

150

151 The success of the printing process was evaluated according to two main aspects: (i) the effort  
152 needed to extrude the material through the printer nozzle, thereby assessing its extrudability and  
153 (ii) the stability of the layers forming the component, indicating its buildability. Accordingly, a 4-  
154 axis gantry printer, whose details were provided in [36], was employed in the printing of the  
155 designed mixes via the use of a nozzle that had a width of 30 mm and a layer height of 15 mm.

156 The printer was connected to a screw pump. The optimum speed for the mixes was determined by  
157 varying the print speed from 60 to 100 mm/sec, while the flow rate was kept constant, with the  
158 goal of maintaining the shape stability of the filaments. The shape retention factor (SRF) for  
159 selected mixes was obtained by dividing the width of the printed filament with the width of the  
160 rectangular nozzle (30 mm). A digital imaging technique was used to capture the width of the  
161 filament after the deposition of each layer. The obtained photos were analysed with “ImageJ”, an  
162 image processing program to accurately determine the width of the printed filament. Finally, a  
163 freeform design was printed by using the best flow rate and speed combination to validate the  
164 feasibility of using the proposed criteria and mix design in practice.

165

166

#### 167 **2.2.4 Green strength**

168

169 Uniaxial unconfined compression tests were performed on freshly prepared cylindrical samples  
170 with a diameter of 70 mm and a height of 140 mm, as shown in Fig. 2. These samples were kept  
171 in moulds for 15 minutes (i.e. replicating the approximate time interval from mixing to printing,  
172 to reveal the mix condition during the printing process) before the mould was carefully removed  
173 to expose the sample without any damage. The dimensions were chosen to avoid any variations on  
174 the results due associated with particle size, leading to a diagonal shear failure [37]. An INSTRON  
175 5969 test rig equipped with a 50 kN load cell was used at a loading rate of 0.6 N/s for strength  
176 testing, which was selected to simulate the loading steps in the printing process. Each test was  
177 conducted at ~60 to 70 seconds to avoid the influence of thixotropic build-up. The averages of the  
178 obtained values were presented in the load and deformation diagram.

179

180

#### 181 **2.2.5 Compressive strength**

182

183 Samples were cast into 50×50×50 mm cubic moulds and their average 28-day compressive  
184 strength values were reported in Table 4. For the 3D printed samples, blocks with the dimensions  
185 of 300×150×200 mm were printed by using a 20 mm circular nozzle at a print speed of 90 mm/sec  
186 and flow rate of 3.15 g/lit. A circular nozzle was preferred as the blocks were printed without any

187 nozzle rotation. 50×50×50 mm cubes were then extracted from these blocks by using a saw. The  
188 compressive strengths of 50×50×50 mm cubic samples after 28 days of ambient curing  
189 (temperature of 30±2°C and relative humidity (RH) of 60±5%) were measured by three directional  
190 uniaxial loading, in line with ASTM C109/C109-M13 [38]. The test was performed via the use of  
191 a Toni Technik Baustoffprüfsysteme machine, during which the loading rate was set at 55 kN/min.  
192 Each data point represented the average of four samples, accompanied with their standard  
193 deviation.

194  
195

## 196 **2.2.6 Characterization of reaction products (XRD, TG/DTG, FESEM-EDS)**

197  
198 The compositions and morphologies of the phases that formed were evaluated via microstructural  
199 tests in order to shed light on the mechanism behind the strength development of the 3D printed  
200 MgO-SiO<sub>2</sub> samples. To facilitate these investigations, segments carefully extracted from the 3D  
201 printed samples were kept in acetone for 3 days to stop hydration and vacuum dried before being  
202 ground down to pass through a 75 μm sieve in preparation for microstructural analyses.

203  
204 The different phases within the hydrated samples were identified and characterized via x-ray  
205 diffraction (XRD) and thermogravimetry/derivative thermogravimetry (TG/DTG) analyses. XRD  
206 was performed on the ground powder extracted from the best performing samples after 28 days of  
207 curing, by using a Bruker D8 ADVANCE instrument with a scanning rate of 0.02° 2θ/step from  
208 10° to 80° 2θ to identify the formed hydration products (e.g. M-S-H and brucite) as well as any  
209 unreacted phases. TG/DTG was carried out by a Perkin Elmer TGA 4000 equipment, which was  
210 operated under a constant nitrogen flow. Samples were heated from 30°C to 900°C at a heating  
211 rate of 10 °C/min. The morphology of the formed phases was observed via a JEOL JSM-7600F  
212 field emission scanning electron microscope (FESEM), at an accelerating voltage of 5.0 kV. In  
213 preparation for FESEM, the specimens were coated with platinum under 20 mA current for 30  
214 seconds. Energy dispersive spectroscopy (EDS) measurements were used to determine the  
215 elemental composition of the 3D printed specimens at 28 days. The accelerating voltage used  
216 during this analysis was 15.0 kV.

217

218

### 219 3. Results and Discussion

220

#### 221 3.1 Effect of mix design on slump, flow diameter and viscosity recovery

222

223 The results of slump and flow table tests of the prepared mixes were plotted in Fig. 3. As mix M1  
224 led to a very dry and stiff mix because of a combination of low w/b ratio (0.4) and SP dosage  
225 (1.5%) in the presence of 50% MS, it was not shown in Fig. 3 and did not qualify for the subsequent  
226 pumpability test due to discontinuous filament extrusion. Amongst the remaining mixes, M3 and  
227 M5 revealed low values of slump and flow diameter, which was associated with their low w/b  
228 ratios (0.4) and/or SP dosage (1.5%). On the other hand, mixes M2, M4, M6 and M8, all of which  
229 had high w/b ratios (0.5), resulted in similar or high slump and flow diameter values; whereas mix  
230 M7 led to intermediate outcomes due to its low w/b ratio (0.4) but high SP dosage (2.5%). The  
231 higher water content within these mixes (i.e. M2, M4, M6, M8) reduced the interaction and internal  
232 friction amongst the particles, thereby resulting in a higher flowability. However, previous  
233 research [39] indicated that mixes with high slump values could experience elastic collapse during  
234 3D printing, which would negatively affect the mechanical properties of the printed specimens by  
235 inducing porosity. Even though mixes M2 and M4 had similar slump values with mixes M5 and  
236 M7, their flow diameter values were significantly higher, which could result in challenges in  
237 retaining the printed shape and hinder the success of the overall 3D printing process. Therefore,  
238 mixes M3, M5 and M7 were chosen for further rheological assessment as a part of the evaluation  
239 of the suitability of these mixes for 3D printing.

240

241 Fig. 4 shows the shear recovery of the three mixes M3, M5 and M7 selected for the further  
242 investigation of their rheological properties. The results of viscosity recovery test were divided in  
243 three parts to mimic the 3D printing process, starting with (i) material at rest, (ii) material sheared  
244 or extruded at a high shear rate and (iii) material at rest again on the print bed. During this analysis,  
245 the viscosities before and after the extrusion process are important for the success of extrusion-  
246 based 3D printing. Accordingly, an ideal material should recover to its initial viscosity as soon as  
247 it is deposited on the print bed. The results shown in Fig. 4 revealed the ability of all three mixes  
248 to recover up to 80% of their original viscosities within 60 seconds of extrusion. Out of these

249 mixes, M3 exhibited slightly higher apparent viscosity values than the other two mixes with a  
250 reasonable recovery ability. This outcome could be an indication of the dense microstructure of  
251 M3, which was also shown by its low slump value and flow diameter in Fig. 3.

252

253

### 254 **3.2 3D printability and mechanical performance**

255

256 In this section, the printability of the three selected mixes was studied by using a 4-axis gantry  
257 printer. Table 3 shows the SRF values of these mixes that were extruded through a rectangular  
258 nozzle, indicating their shape stability. Amongst the three mixes, M3 revealed the lowest SRF  
259 value of  $\sim 1.0$ , followed by M5 ( $\sim 1.84$ ) and M7 ( $\sim 2.32$ ). The trend observed in the SRF values was  
260 in agreement with the rheological properties (i.e. slump and flow diameter values), in which M3  
261 demonstrated the lowest slump/flow diameter, followed by M5 and M7. The combination of these  
262 outcomes could explain the enhanced viscosity recovery of mix M3, which also exhibited the best  
263 shape stability, as indicated by its SRF of  $\sim 1.0$ . However, these differences amongst the three  
264 samples do not necessarily translate into poor shape stability for the other two mixes (M5 and M7)  
265 as SRF is highly influenced by printing parameters such as print speed, flow rate and nozzle height.  
266 Considering the higher slump and flowabilities of mixes M5 and M7 than those of mix M3, they  
267 may require a higher print speed ( $> 90$  mm/sec) to obtain a stable filament shape.

268

269 The 28-day compressive strengths of the selected samples are shown in Table 4. Out of the three  
270 mixes, mix M3 exhibited the highest 28-day compressive strength (44.3 MPa), which could be  
271 attributed to its denser microstructure, potentially linked with the higher degree of M-S-H  
272 formation (i.e. the composition of hydrate phases within mix M3 is shown in Section 3.4) [40].  
273 The differences in the binder composition of mix M3 (MgO:MS of 50:50) when compared to mixes  
274 M5 and M7 (MgO:MS of 70:30) could explain the increased formation of M-S-H in mix M3 as  
275 opposed to the predominant formation of brucite in the presence of 70% MgO in mixes M5 and  
276 M7 [26, 41]. This was in line with the findings of previous studies [42], where the direct correlation  
277 between the strength development of MgO-SiO<sub>2</sub> samples cured under different conditions and M-  
278 S-H formation was identified.

279

280 Considering its ideal thixotropic properties demonstrated by its viscosity recovery, lowest SRF  
281 and highest compressive strength amongst the formulations investigated under this study, mix M3  
282 was chosen for further analysis, during which it was used in the printing of a 3D component. The  
283 settings of the gantry printer were fixed at a print speed of 90 mm/sec and a flow rate of 3.15 g/lit,  
284 in line with the findings of previous studies [32]. The layer-by-layer extruded structure led to the  
285 building of the 3D component shown in Fig. 5. The printed component, whose dimensions were  
286 300 mm (L) x 150 mm (W) x 100 mm (H), indicated a good shape retention as well as buildability  
287 properties, highlighting the suitability of mix M3 to be used in 3D printing applications without  
288 any discontinuity and deformation.

289

290

### 291 **3.3 Green strength and directional mechanical properties**

292

293 The buildability of mix M3 was highly related to its initial (green) strength. To indicate this  
294 relationship and provide a comparison to conventional mixes, the performance of mix M3 was  
295 compared with a PC-based mix, which was prepared with the same w/b ratio and without any other  
296 additives. The load-deformation curves of these two mixes, shown in Fig. 6, revealed the higher  
297 load carrying capacity of mix M3 than PC, which could be attributed to the differences in the  
298 reaction products and the binding network they formed within each mix [43]. However, the  
299 performance of the PC mix could be improved by reducing the w/b ratio [44], possibly resulting  
300 in similar outcomes as mix M3.

301

302 Another inherent property that must be considered in the evaluation of a potential mix for 3D  
303 printing is directional mechanical strength, which is associated with the layer-wise approach of the  
304 printing process. Fig. 7(a) demonstrates the main orthogonal directions F1, F2 and F3; whereas  
305 the 28-day compressive strengths of 3D printed samples prepared by using mix M3 in F1, F2 and  
306 F3 directions are presented in Fig. 7(b). While the analysed samples revealed comparable strengths  
307 in F2 and F3 directions, the highest strength was achieved in F1 direction. Similar anisotropic  
308 behaviour in terms of mechanical properties was also observed in other alternative binder systems  
309 (e.g. glass fibre reinforced geopolymer samples) [45]. However, due to the identical orientations  
310 of F1 and F3, it was hard to attribute this outcome to different layer directions. These strength

311 variations could also be due to the motion patterns of the material during the printing process,  
312 resulting in alterations in the sample microstructure [46]. Regardless of these slight variations  
313 observed in different directions, the strength results recorded at 28 days were generally comparable  
314 in all directions and with the strength of the directly cast samples, which highlighted the robustness  
315 of these 3D printed samples for various uses including load bearing applications.

316

317

### 318 **3.4 Microstructural analyses (XRD, TG/DTG and FESEM-EDS)**

319

320 The crystalline and amorphous phases present in the 3D printed M3 mix at 28 days were identified  
321 by using XRD analysis, as shown in Fig. 8. The main hydration products identified in this system  
322 were M-S-H gel, as indicated by the amorphous humps at  $\sim 26^\circ$ ,  $35^\circ$  and  $60^\circ$   $2\theta$  [27]; and brucite  
323 ( $\text{Mg}(\text{OH})_2$ ), with the main crystalline peaks at  $18.6^\circ$  and  $38.0^\circ$   $2\theta$ . Other than the formed hydration  
324 products, the presence of unreacted raw materials was notable. Accordingly, unreacted MgO  
325 (periclase) revealed its main peaks at  $\sim 42.9^\circ$  and  $62.3^\circ$   $2\theta$  with high intensities, while unreacted  
326  $\text{SiO}_2$  had an amorphous hump centred at  $\sim 23^\circ$   $2\theta$ . The presence of significant amount of unreacted  
327 MgO indicated that potentially higher strengths could have been achieved if a higher degree of  
328 hydration was realized. In addition to the mentioned phases, magnesium carbonate ( $\text{MgCO}_3$ ),  
329 which was present as an impurity in MgO, was observed at  $32.6^\circ$  and  $53.9^\circ$   $2\theta$ .

330

331 TG and DTG plots of the 3D printed M3 mix at 28 days are presented in Fig. 9(a) and (b),  
332 respectively. As can be seen in Fig. 9(a), almost 20% loss in mass was observed, which took place  
333 in multiple stages, when the sample was subjected to high temperatures of up to  $900^\circ\text{C}$ . The DTG  
334 curves (Fig. 9(b)) revealed detailed information about the decomposed phases. Accordingly, the  
335 first decomposition was completed at  $\sim 200^\circ\text{C}$ , which could be attributed to the dehydration of the  
336 interlayer water of M-S-H [47, 48]. The following decomposition curve centred at  $\sim 400^\circ\text{C}$  was  
337 due to brucite dehydroxylation [26, 27]. The curve that partially overlapped with brucite  
338 dehydroxylation was associated with M-S-H dehydroxylation, which also continued at higher  
339 temperatures [20, 26], as can be seen in Fig. 9(b). The final decomposition step of the M-S-H gel  
340 took place at  $\sim 750^\circ\text{C}$  due to the dehydroxylation of the silanol groups [26, 48]. In addition to these  
341 hydrate phases, the decomposition of magnesite ( $\text{MgCO}_3$ ), which was present as an impurity

342 within MgO used as a binder, occurred at  $\sim 650^{\circ}\text{C}$ , revealing a minor curve in the DTG plot of the  
343 analysed sample. The decomposition patterns shown by the TG/DTG plots were in line with the  
344 XRD results, indicating the presence of M-S-H and brucite as the main hydrate phases,  
345 accompanied by undecomposed binder phases.

346  
347 Fig. 10 shows the FESEM image of the 3D printed M3 mix at 28 days. In line with the XRD and  
348 TG/DTG results, a majority of the sample microstructure was comprised of a widespread  
349 formation of M-S-H gel, which was the main contributor of strength in MgO-SiO<sub>2</sub> samples [49].  
350 Along with M-S-H, spherical particles indicative of unreacted SiO<sub>2</sub> were observed. Furthermore,  
351 plate-like hydrates smaller than 1  $\mu\text{m}$ , which resembled brucite crystals [50], could be seen in  
352 multiple locations within the microstructure. Overall, the continuous gel-like structure established  
353 by M-S-H covered the particle surfaces, leading to the formation of a dense microstructure, which  
354 contributed to mechanical performance, enabling the potential use of mix M3 in various building  
355 applications.

356  
357 EDS analysis was also performed on mix M3 after 28 days by using the 14 arbitrarily chosen areas  
358 within the region shown in Fig. 11(a). A representative elemental analysis of one of the areas  
359 indicated by the rectangular box (i.e. Spectrum 8) in Fig.11(a) was presented in Fig. 11(b). The  
360 average and standard deviations of the values obtained from the measurements taken from 14  
361 different areas were listed in Table 5. The elements O, Mg and Si were dominantly present within  
362 the measurement territory, highlighting the formation of M-S-H as well as brucite (i.e. composed  
363 of only Mg and O). The weight percentages of O, Mg and Si were around 52%, 21% and 21%,  
364 respectively. On top of these 3 elements, presence of  $\sim 6$  wt.% carbon (C) was observed, which  
365 could be attributed to magnesite that was present as an impurity within MgO and the partial  
366 absorption of CO<sub>2</sub> from the atmosphere during sample preparation and curing.

367

368

#### 369 **4. Conclusions**

370

371 This paper focused on the development of MgO-SiO<sub>2</sub> binders to be used in 3D printing  
372 applications. The effects of parameters such as water-to-binder (w/b) ratio, MgO/SiO<sub>2</sub> ratio and

373 superplasticizer (SP) dosage were assessed in order to produce sustainable 3D printable MgO-SiO<sub>2</sub>  
374 binders. The assessment of fresh properties was supported with mechanical testing and  
375 microstructural characterization. The following conclusions can be drawn from this study:

376

- 377 • The thixotropic behaviour of the developed mixes, which recovered 70-80% of their  
378 original viscosity within 60 seconds of extrusion, enabled the deposition of printed layers  
379 in 60 second intervals, without any pumpability issues.
- 380 • Out of the different formulations prepared in this study, the best-performing mix that  
381 satisfied the performance (i.e. high strength) and printability criteria (i.e. low slump and  
382 high flow) was determined.
- 383 • The selected mix (M3) indicated a SRF of ~1, highlighting its buildability potential, which  
384 was successfully demonstrated via the 3D printing of a 5-layered structure without any  
385 deformations.
- 386 • The anisotropic mechanical performance displayed by the printed samples, in comparison  
387 to the cast specimens, could be associated with the layer-wise fabrication approach adopted  
388 in the printing process. In spite of these differences, satisfactory strengths were obtained in  
389 all 3 directions.
- 390 • Microstructural characterization revealed the formation of brucite and M-S-H gel as the  
391 main hydration products, which were the main contributor to the mechanical performance  
392 observed within the prepared samples.

393

394 The findings of this research paved the way for the development of alternative MgO-SiO<sub>2</sub> binders  
395 for 3D printing applications and presented new insights on the printability aspects of these  
396 mixtures under the context of 3D printing, which have not been studied until now. Overall, the  
397 results of this study have shown that MgO-SiO<sub>2</sub> binders with proper mix designs are suitable to be  
398 used in 3D printing applications, both in terms of their rheological and mechanical properties.  
399 Considering the need for the optimization of these formulations for enhanced performance as well  
400 as an improved understanding of the mechanisms behind strength and microstructural  
401 development, further studies will focus on the enhancement of the performance of MgO-SiO<sub>2</sub>  
402 binders through an optimization of their fresh and long-term properties, with the overall goal of  
403 enabling their use in various building applications, including those involving 3D printing. Any

404 initiative taken towards this goal will facilitate the wide scale use of these alternative binders in  
405 different building components.

406

407

408 **Acknowledgements**

409

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## List of Tables

Table 1 Chemical composition and physical properties of the binder components.

Materials	Content (%)								SSA (m <sup>2</sup> /g)	LOI (%)
	MgO	CaO	SiO <sub>2</sub>	Al <sub>2</sub> O <sub>3</sub>	C	P <sub>2</sub> O <sub>5</sub>	Fe <sub>2</sub> O <sub>3</sub>	K <sub>2</sub> O		
MgO	>91.5	1.6	2.0	<0.7	-	-	-	-	35.7	6.3
MS	-	-	>90.0	-	-	-	-	-	17.0	2.2

Table 2 Mix compositions used in this study.

Mix	Binder composition		Water-to-binder (w/b) ratio	SP dosage (%)
	MgO (%)	MS (%)		
M1			0.4	1.5
M2	50	50	0.5	1.5
M3			0.4	2.5
M4			0.5	2.5
M5			0.4	1.5
M6	70	30	0.5	1.5
M7			0.4	2.5
M8			0.5	2.5

Table 3 Shape retention factors (SRF) of 3D printed MgO-SiO<sub>2</sub> binders.

Mix	SRF
M3	~1.0
M5	~1.84
M7	~2.32

Table 4 28-day compressive strengths of cast MgO-SiO<sub>2</sub> samples

Mix	Compressive strength (MPa)	Relative error (±)
M3	44.3	0.6
M5	38.3	1.0
M7	36.2	0.9

Table 5 Elemental composition (wt.%) of mix M3 after 28 days of curing

C	O	Mg	Si
$5.77 \pm 1.11$	$51.82 \pm 2.41$	$21.26 \pm 2.92$	$21.15 \pm 4.66$

**List of Figures**

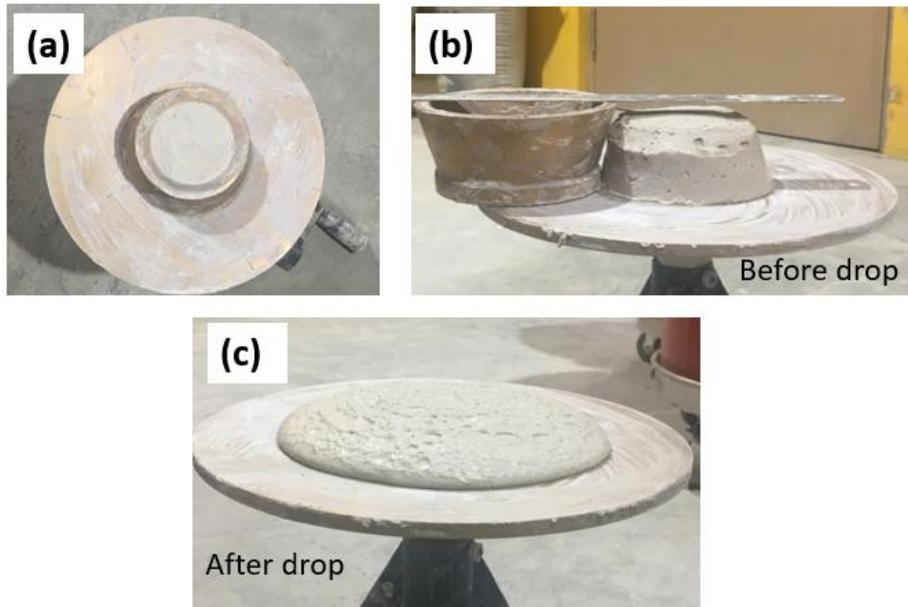


Fig. 1 Photos showing the (a) slump and flow table test setup; and behaviour of the MgO-SiO<sub>2</sub> pastes (b) before and (c) after drop

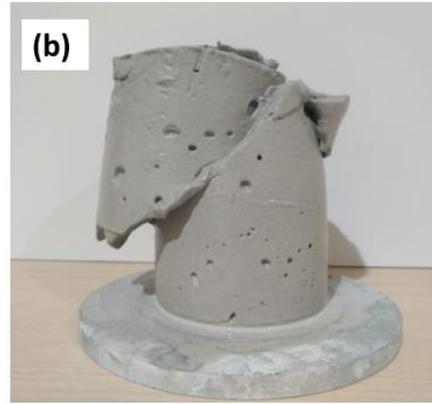


Fig. 2 Photos showing the (a) experimental setup and (b) typical shear failure of the MgO-SiO<sub>2</sub> pastes during uniaxial compression test

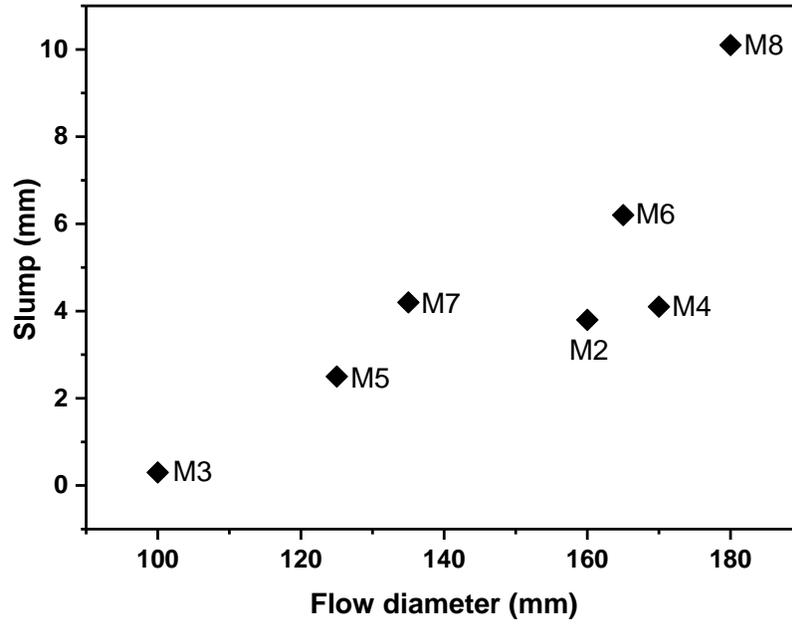


Fig. 3 Slump values and flow diameters of MgO-SiO<sub>2</sub> pastes

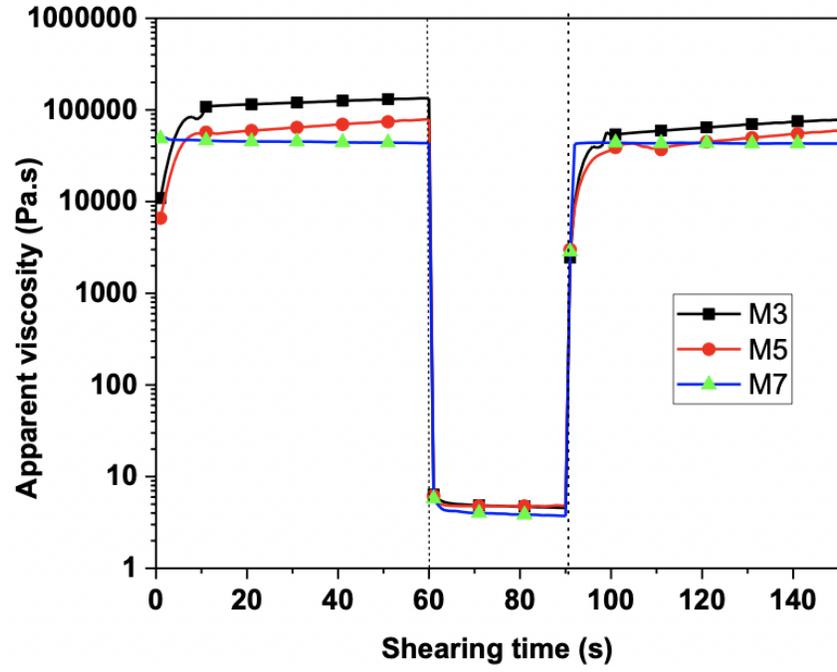


Fig. 4 Viscosity recovery of MgO-SiO<sub>2</sub> pastes



Fig. 5 3D printed section via the use of mix M3

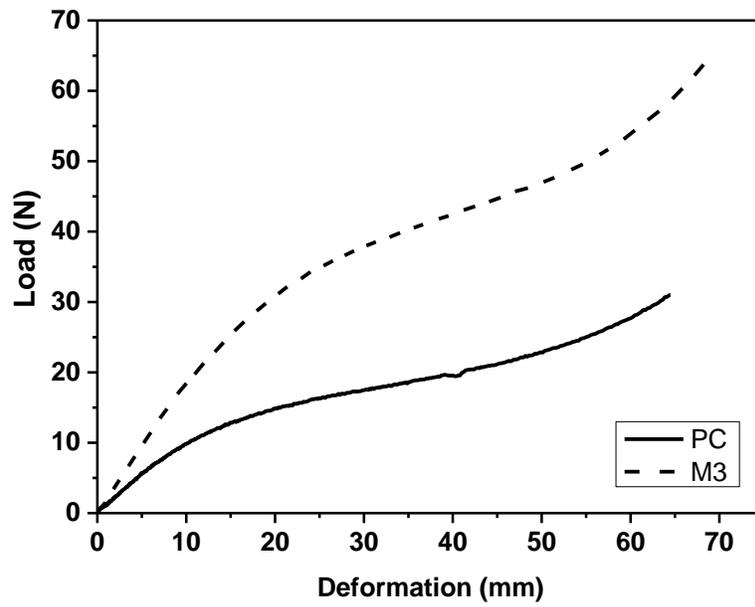


Fig. 6 Load-deformation curves of M3 (MgO-SiO<sub>2</sub>) and PC mixes after 15 minutes of initial contact with water

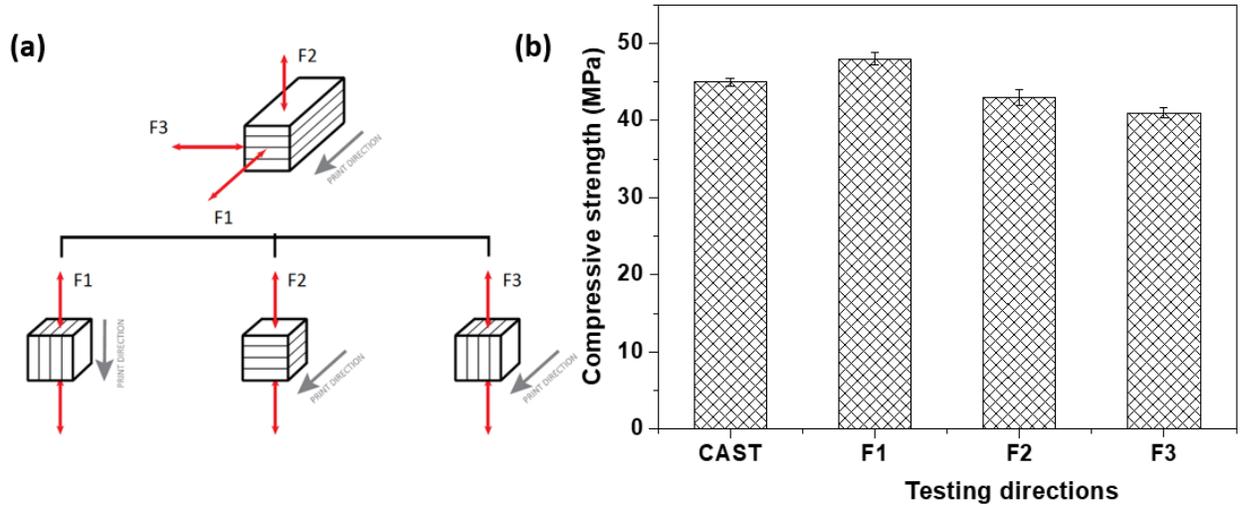


Fig. 7 Illustration of (a) the directional compressive testing orientations with respect to print direction, and (b) mechanical properties of cast and 3D printed mix M3 in 3 directions

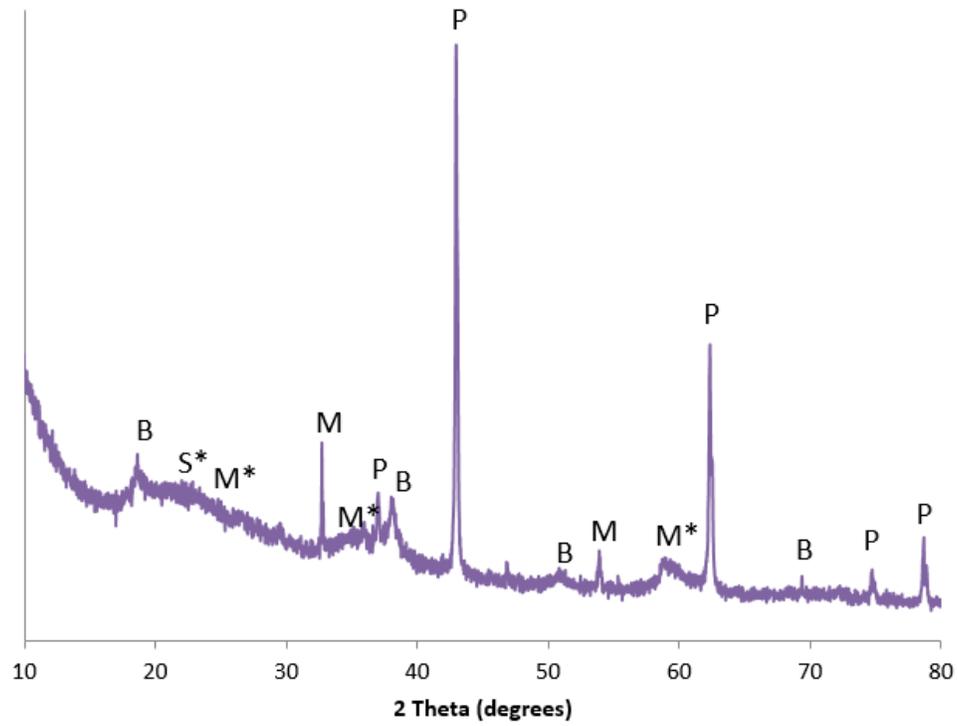
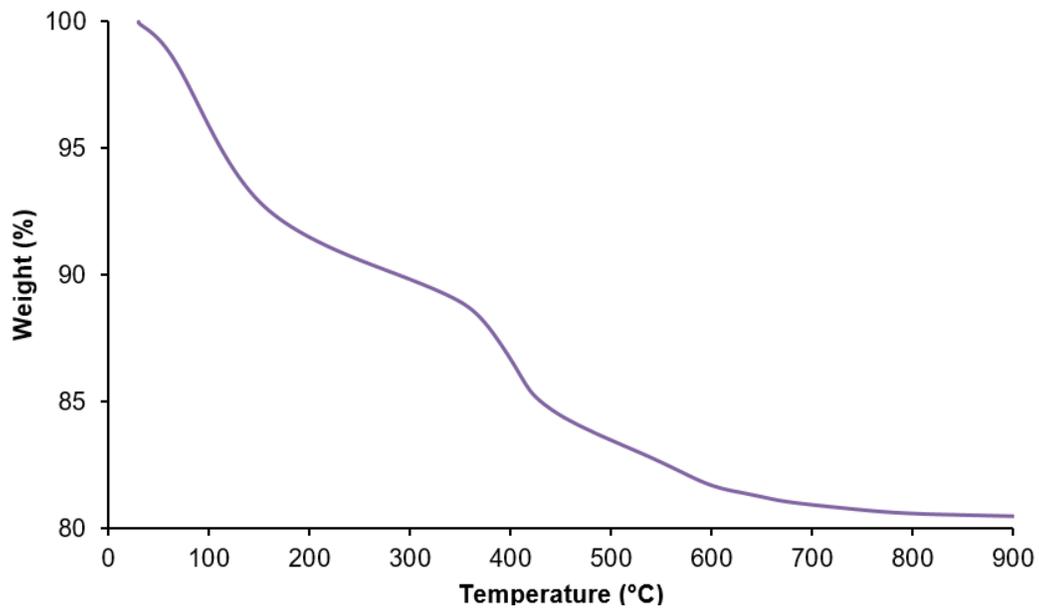
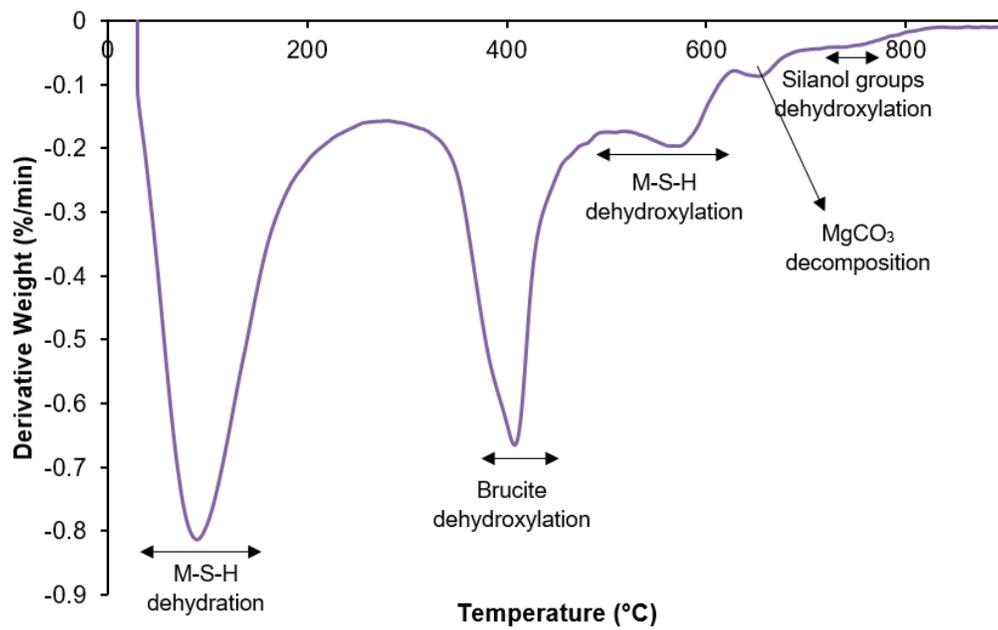


Fig. 8 XRD pattern of mix M3 at 28 days

(B: Brucite, P: Periclase, M: Magnesite, M\*: M-S-H, S\*: SiO<sub>2</sub>)



(a)



(b)

Fig. 9 Thermogravimetric analysis results showing (a) TGA and (b) DTG plots of mix M3 at 28 days

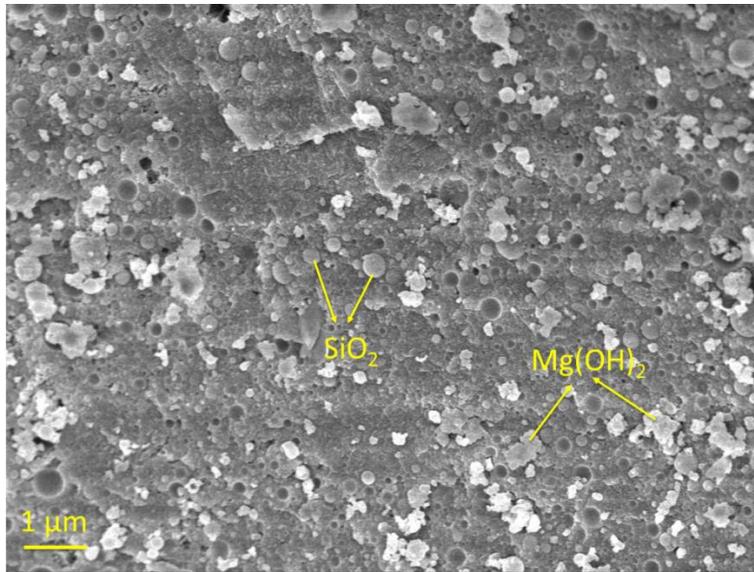
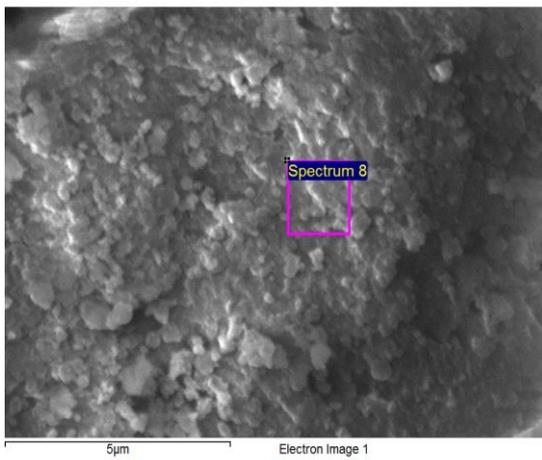
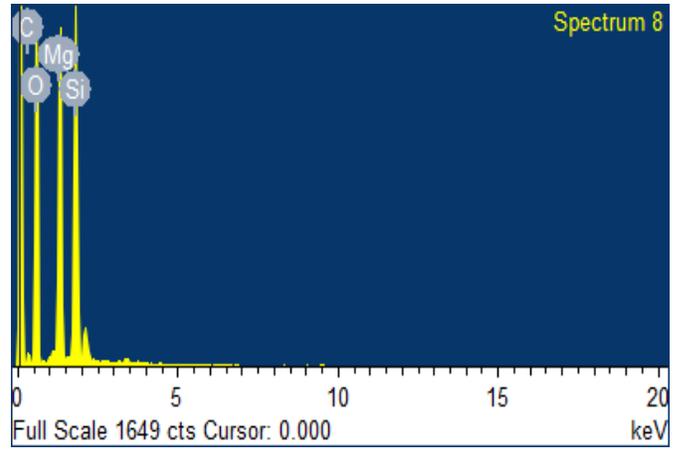


Fig. 10 FESEM image of mix M3 at 28 days



(a)



(b)

Fig. 11 Images taken during the EDS analysis of mix M3 at 28 days, showing: (a) region of the specimen, where multiple EDS analyses have been performed and (b) representative EDS elemental analysis of the spectrum indicated in Fig. 11(a)