

1 Supplemental information

2

3 **Using the MitoB method to assess levels of reactive oxygen species in ecological studies**  
4 **of oxidative stress**

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## 18 Appendix S1: STANDARD OPERATING PROCEDURE FOR HPLC-MS TUNING

19 All quantification of the compounds MitoP,  $d_{15}$ MitoP, MitoB and  $d_{15}$ MitoB, using a high performance  
20 liquid chromatography system connected to a mass spectrometer (HPLC-MS), were conducted  
21 following standard operating procedures based on the following guidelines. We present how the  
22 method was conducted in our study as an example (but non-exclusive way) and then we present  
23 solutions to frequently encountered issues.

### 24 *Instrumentation and chemical*

25 To separate and introduce the compounds into the MS a liquid chromatography system is required,  
26 while quantification of the compound in the eluant coming from the HPLC requires a mass  
27 spectrometer. This can either be a HPLC system or an ultra-performance liquid chromatography  
28 (UPLC) system using a 2.0 mm column running at 200  $\mu$ l/min or a 4.6 mm column running at 1  
29 ml/min. The chromatographic column used in the present study is a standard reverse phase C 18  
30 column such as a hypersil gold column (150 mm x 2 mm, I.D. 2.0 $\mu$ m) with guard column. Standards  
31 at known concentration of isolated compounds are necessary to set up the quantification of the 4  
32 compounds. The concentration of the standards used to provide calibration curve is highly dependent  
33 on the system used. Table S1a gives a suggested range of starting concentrations for a conventional  
34 HPLC 4.6 mm column (flowrate 1 ml/min) and UPLC 2.0 mm column (flowrate 200  $\mu$ l/min).

35

### 36 *Mass spectrometric detection of the 4 compounds*

37 Two types of mass analysers have been used to detect Mito compounds in biological samples: one  
38 based on high resolution accurate mass (HRAM) <sup>1</sup>, and the other on tandem mass analysis (MS/MS) <sup>2</sup>.  
39 Mass spectrometric analysis was carried out using positive ion mode. To tune the HRAM or the  
40 MS/MS, directly infuse the isolated compound into the MS using an appropriately diluted stock  
41 solution containing only a single compound, starting at a mid-range concentration to visualize any low

42 level contaminants while not overloading the mass spectrometer with the compound of interest (e.g.  
43 Standard 4 in Table S1a). Detect the compound on the ion spectrum according to the accurate mass of  
44 the ions or the approximate one for the HRAM and MS/MS, respectively (Table S2a and b). This  
45 provides evidence of the purity of each standard and allows tuning of the mass spectrometer on each  
46 compound if needed. In the case of MS/MS analysis this procedure was used to optimise the  
47 fragmentation energy. The fragmentation energy used in our experiments carried out in an ion trap  
48 mass spectrometer (Thermo Fisher Scientific) was 55%.

49

#### 50 *Coupling of the chromatographic system to the mass spectrometer*

51 First, the stock solution should be checked for impurities by HPLC-MS analysis at high concentration.  
52 Second, a mixture of the 4 standards is assessed to confirm separation of the peaks of MitoB  
53 compounds from the MitoP compounds. In the setup described here there was approximately one  
54 minute separation between the pairs of peaks P *versus* B. Finally, an extended calibration curve is  
55 performed to ensure the linear range of the system and the minimum and maximum concentrations  
56 that can be quantified for each of the 4 compounds. Note that molecules that differ only in the degree  
57 of deuteration can show small differences in their interactions with the stationary phase during  
58 separation by HPLC, so that the deuterium compounds elute slightly more quickly than their isotope <sup>3-</sup>  
59 <sup>5</sup> – this is evident in figure 2 of the main article. For this reason, it is important to integrate the amount  
60 of the deuterated and undeuterated compounds over the whole peak area in place of peak height.

61

#### 62 *Sequence of the gradient in the mobile phase*

63 The mobile phase used in our analysis was based on solvent A: 98% water 2% acetonitrile in 0.1%  
64 formic acid and solvent B 100% acetonitrile with 0.1% formic acid. The gradient that allowed  
65 separation of the four MitoP, *d*<sub>15</sub>MitoP, MitoB and *d*<sub>15</sub>MitoB compounds was based on a starting  
66 solvent of 30% B, rising to 65% solvent B over 10 minutes prior to a wash and re-equilibration phase.

67 *Integration of the peak area*

68 The data was quantified by integrating the area under each peak. In the software package used in our  
69 experiments (Xcalibur version 2.0.7) the significant parameters used were:

70 Integration algorithm: Genesis

71 Percentage of highest peak: 1%

72 Minimum peak height (S/N): 2

73 S/N threshold: 0.5

74

75 *Possible issues:*

76 • If the pairs of compounds P and B overlap, the proportion of solvent B in the elution should  
77 be decreased to allow slower migration of the probes, which should improve separation  
78 without adversely affecting the peak width of the probes.

79 • If the concentration of the probe is too high this may result in a carry-over effect (causing  
80 cross-contamination between runs, as part of the sample is not completely washed off the  
81 column from the previous injection).

82 • If a peak in a sample is below the limit of quantification of  
83 the calibration curves, then its quantification cannot be made. A possible solution is to  
84 increase the volume of injection, but it is necessary to ensure that the volume injected does  
85 not overload the column for the other compounds causing asymmetric or flattened peaks.

86 • If only MitoP is below the linear section of the calibration curves, it may mean that the  
87 exposure duration was not long enough to display sufficient accumulation of MitoP in the  
88 tissue of interest and the size of the sample extracted. This can be overcome by increasing the  
89 exposure time duration and / or the amount of extracted sample.

90 • If no probe peaks are detected after injection of a sample while the compounds in the  
91 standards are seen, the issue may come from the sample itself. This can be because they  
92 contain high levels of compounds that can cause suppression of ionisation of the compounds

93 of interest, usually during or immediately after elution of the void volume peak (Fig. S1). The  
94 elution gradient should allow sufficient time for the ion current to re-stabilise before the  
95 probes elute into the mass spectrometer.

96 • Another ion suppression type problem can, and did occur, due to a very high level of an  
97 unknown compound in samples that suppressed detection of closely eluting probes<sup>6</sup>. We were  
98 unable to resolve this issue using chromatographic separation. If analysis is by full scan  
99 HRAM a possible solution is to split the scanned mass range so that this ion is not included  
100 for mass analysis. Alternatively, a selected ion monitoring approach can also work to  
101 eliminate this type of problem.

102

#### 103 Appendix S2: COMPARISON OF STANDARDS TO BUILD THE CALIBRATION CURVE

104 In our study, calibration curves were generated using standards prepared by serial dilutions of stock  
105 solutions of the four compounds, without any tissue sample, as in Salin, et al.<sup>1</sup>. Cochemé, et al.<sup>2</sup> in  
106 the original protocol prepared the standards for the calibration curve using a different approach: the  
107 standards were processed with a tissue sample and various amounts of MitoB and MitoP but constant  
108 amounts of deuterium compounds. We compared both types of calibration curves, hereafter called  
109 *calibration 1* and *calibration 2* for standards prepared according to Cochemé, et al.<sup>2</sup> and our study,  
110 respectively. For *calibration 1*, standards were prepared by serial 1:5 and 1:2 dilutions of a MitoB and  
111 MitoP stock solution (Table S1b) and were added to control samples of liver tissue from unexposed  
112 fish. Tissues were then homogenised with the MitoB standard, the MitoP standard and the spike.  
113 Standards were processed by following the same extraction protocol as was used for the experimental  
114 liver samples. For *calibration 2*, calibration curves were generated using standards of MitoB,  
115 *d*<sub>15</sub>MitoB, MitoP and *d*<sub>15</sub>MitoP, prepared by serial 1:5 and 1:2 dilutions of stock solutions of each of  
116 the four compounds in ethanol which were then directly added to a solution containing 20% ACN, 0.1  
117 % FA (table S1a). To test the consistency of the MitoP/MitoB ratios calculated from *calibration 1* and  
118 2, the quantification of probe levels in liver samples from 40 individuals exposed to MitoB for 24h  
119 were run in a single reading set along with standards for both *calibrations 1* and 2. For *calibration 2*,

120 calculation of the MitoP/MitoB ratio was done as described in the section “calculation of the  
121 MitoP/MitoB ratio” in the main text. For *calibration 1*, calculation of the MitoP/MitoB ratio differed.  
122 The ratios of AA MitoB/AA  $d_{15}$ MitoB and AA MitoP/AA  $d_{15}$ MitoP were calculated for the standards  
123 and the samples. The calibration curves of AA MitoB /AA  $d_{15}$ MitoB and AA MitoP/AA  $d_{15}$ MitoP  
124 against pmol of MitoB and MitoP was then generated. The MitoB and MitoP content for each liver  
125 samples from the 40 individuals exposed to MitoB was calculated by converting the AA MitoB/AA  
126  $d_{15}$ MitoB and AA MitoP/ AA  $d_{15}$ MitoP ratios into pmol using the appropriate calibration curve. The  
127 calculated values for the MitoP/MitoB ratio were slightly smaller when using *calibration 2* (Mean  $\pm$   
128 SE:  $0.0731 \pm 0.0063$ ) than when using *calibration 1* (Mean  $\pm$  SE:  $0.0877 \pm 0.0076$ ). However, the  
129 level of consistency of the MitoP/MitoB ratio between the values calculated from the two calibration  
130 curves was very high (Fig. S2, ICC  $r = 0.959$ ,  $p < 0.001$ ). For the rest of the study, MitoP/MitoB  
131 ratios were calculated using *calibration 2* since it allows as accurate a calculation as *calibration 1*  
132 while determining of standard curve and detection limits for each compounds, saving the time of the  
133 extraction step and removing the need to sacrifice additional animals for control tissues.

134

### 135 Appendix S3: REPEATABILITY OF THE QUANTIFICATION OF THE COMPOUNDS AND 136 MitoP/MitoB RATIOS

137 To assess the HPLC-MS repeatability, the quantification of probe levels in extracts of liver samples  
138 from 40 individuals exposed to MitoB for 24 h were run in duplicate in a single reading set, along  
139 with duplication of the calibration curve. Intra class correlation coefficients (ICC) were used to test  
140 for the consistency of the quantification of the compounds between the two measures from the same  
141 tissue extract. The resulting repeatability was high for the absolute area of the four compounds  
142 (MitoP: ICC  $r = 0.983$ ,  $P < 0.001$ ,  $d_{15}$ MitoP: ICC  $r = 0.703$ ,  $P < 0.001$ , MitoB: ICC  $r = 0.974$ ,  $P <$   
143  $0.001$ ,  $d_{15}$ MitoB: ICC  $r = 0.598$ ,  $P < 0.001$ ; in all cases  $n = 40$ ; Fig. S3A to S3D). The resulting  
144 repeatability in the calculated MitoP/MitoB ratios was also very high (ICC  $r = 0.878$ ,  $n = 40$ ,  $P <$   
145  $0.001$ ; Fig. S3E). Note that the MitoP content was below the detection limit of the HPLC-MS in two

146 samples (outliers 1 and 2). In both samples, the AA of MitoP was below the range of linearity of the  
147 standard curve; this was not due to a detection failure since the measurements of the MitoP deuterium  
148 spike in the samples were normal. Instead it is likely that too little probe was extracted, either because  
149 the tissue sample was too small (the liver sample from individual 1 was the smallest of those  
150 analysed) or the homogenization process was inadequate.

151

#### 152 Appendix S4: EXTRACTION OF THE MITOB AND MITOP FROM THE WATER.

153 To extract MitoB and MitoP from water, each water sample (780  $\mu$ L) was added to acetonitrile  
154 (ACN) and formic acid (FA) in order to obtain a final solution containing 60% (v/v) ACN and 0.1 %  
155 (v/v) FA. After centrifuging for 10 min at 16,000 g, the supernatant was centrifuged for a final 10 min  
156 at 4,560g in microcentrifuge filters. The filtered solution from the water samples was processed as  
157 done for the tissue samples from the drying step onwards <sup>1</sup>.

158

#### 159 **References**

- 160 1 Salin, K. *et al.* Individuals with higher metabolic rates have lower levels of reactive oxygen  
161 species in vivo. *Biol. Lett.* **11**, doi:10.1098/rsbl.2015.0538 (2015).
- 162 2 Cochemé, H. M. *et al.* Using the mitochondria-targeted ratiometric mass spectrometry probe  
163 MitoB to measure H<sub>2</sub>O<sub>2</sub> in living *Drosophila*. *Nature Protocols* **7**, 946-958,  
164 doi:10.1038/nprot.2012.035 (2012).
- 165 3 Turowski, M. *et al.* Deuterium Isotope Effects on Hydrophobic Interactions: The Importance  
166 of Dispersion Interactions in the Hydrophobic Phase. *Journal of the American Chemical*  
167 *Society* **125**, 13836-13849, doi:10.1021/ja036006g (2003).
- 168 4 Tanaka, N. & Thornton, E. R. Isotope effects in hydrophobic binding measured by high-  
169 pressure liquid chromatography. *Journal of the American Chemical Society* **98**, 1617-1619,  
170 doi:10.1021/ja00422a076 (1976).
- 171 5 Tayar, N. E., van de Waterbeemd, H., Gryllaki, M., Testa, B. & Trager, W. F. The lipophilicity  
172 of deuterium atoms. A comparison of shake-flask and HPLC methods. *International Journal*  
173 *of Pharmaceutics* **19**, 271-281, doi:[http://dx.doi.org/10.1016/0378-5173\(84\)90057-7](http://dx.doi.org/10.1016/0378-5173(84)90057-7) (1984).
- 174 6 Annesley, T. M. Ion Suppression in Mass Spectrometry. *Clinical Chemistry* **49**, 1041-1044,  
175 doi:10.1373/49.7.1041 (2003).

176

177

178 Table S1: Amount of the probes in the standards used to generate calibration curves of (a) MitoP,  
 179  $d_{15}$ MitoP, MitoB and  $d_{15}$ MitoB, and (b) of MitoP/ $d_{15}$ MitoP and MitoB/ $d_{15}$ MitoB.

180 a.

Standard	MitoP (pmol)	$d_{15}$ MitoP (pmol)	MitoB (pmol)	$d_{15}$ MitoB (pmol)
1	1313	100	5040	200
2	656.5	50	2520	100
3	131.3	10	504	20
4	65.65	5	252	10
5	13.13	1	50.4	2
6	6.565	0.5	25.2	1
7	1.313	0.1	5.04	0.2
8	0.6565	0.05	2.52	0.1
9	0.32825	0.025	1.26	0.05
10	0	0	0	0

181

182 b.

Standard	MitoB (pmol)	MitoP (pmol)	$d_{15}$ MitoB (pmol)	$d_{15}$ MitoP (pmol)
1	2520	656.50	100	50
2	504	131.30	100	50
3	252	65.65	100	50
4	50.40	13.13	100	50
5	25.20	6.57	100	50
6	5.04	1.31	100	50
7	2.52	0.66	100	50
8	0	0	100	50

183



184 Table S2: Mass of the four ions detected by (a) the high resolution accurate mass and (b) the tandem  
185 mass analysis.

186

187 a.

<b>Compounds</b>	<b>Accurate mass M+1</b>
<b>MitoP</b>	369.141
<b><i>d</i><sub>15</sub>MitoP</b>	384.234
<b>MitoB</b>	397.153
<b><i>d</i><sub>15</sub>MitoB</b>	412.247

188

189

190 b.

<b>Compounds</b>	<b>Approximate mass M+1</b>	<b>Mass daughter ions</b>
<b>MitoP</b>	369.2	183-185 and 260-263
<b><i>d</i><sub>15</sub>MitoP</b>	384.2	183-185 and 260-263
<b>MitoB</b>	397.2	191-195 and 275-279
<b><i>d</i><sub>15</sub>MitoB</b>	412.3	191-195 and 275-279

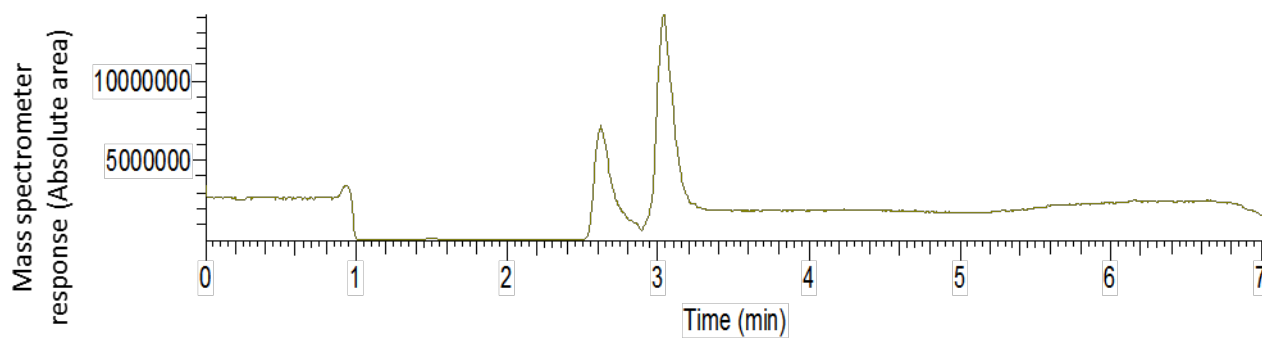
191

192 Table S3: Cross sectional assay of MitoB and MitoP content in the liver of brown trout exposed to  
 193 one of the compounds for different durations (n=4 per time point and per compound). Note that the  
 194 complete disappearance of MitoB and MitoP can be expected over time, however, a much longer time  
 195 course would be necessary to observe this phenomenon in this system.

<b>Injected compound</b>	<b>Exposure duration ± SE (hour)</b>	<b>Final concentration of compound ± SE (pMol/mg liver)</b>	<b>Generated MitoP ± SE; min-max (pMol/mg liver)</b>
<b>MitoB</b>	3.3 ± 0.1	68.03 ± 18.56	4.63 ± 3.55 (0.00-15.16)
<b>MitoB</b>	11.9 ± 0.1	84.21 ± 18.62	10.05 ± 6.65 (0.41-22.30)
<b>MitoB</b>	24.2 ± 0.0	75.40 ± 14.06	2.16 ± 1.20 (0.00-5.59)
<b>MitoB</b>	48.1 ± 0.0	62.42 ± 8.49	2.14 ± 0.44 (0.92-3.01)
<b>MitoB</b>	72.0 ± 0.0	63.70 ± 26.28	2.66 ± 0.82 (0.86-4.34)
<b>MitoP</b>	3.3 ± 0.1	93.77 ± 39.04	
<b>MitoP</b>	11.9 ± 0.1	44.72 ± 14.24	
<b>MitoP</b>	24.3 ± 0.0	127.16 ± 49.79	
<b>MitoP</b>	48.2 ± 0.0	72.91 ± 11.45	
<b>MitoP</b>	72.0 ± 0.0	76.92 ± 19.24	

196

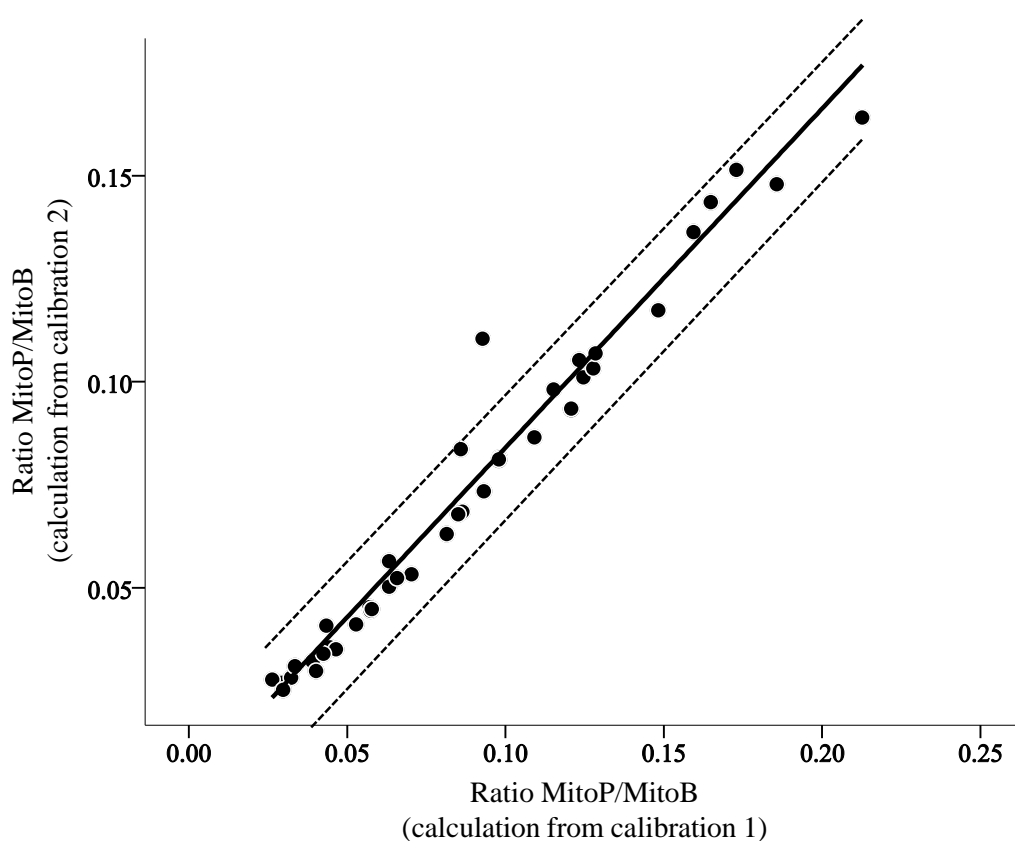
197 Fig. S1: Example of chromatograms from the HPLC-MS analysis of Mito compounds within samples  
198 of trout muscle with ion suppression phenomenon, i.e. a loss of signal between min 1 and min 2.5, just  
199 before the peak of interest. The same phenomenon could occur just at the same time as the peak  
200 immersion, so it would be necessary to split the spectrum of analysis into two spectra excluding the  
201 contaminant ion.



202

203

204 Fig. S2: Consistency of the MitoP/MitoB ratio as calculated using two different calibration  
205 approaches: calculated from standard curves for 4 compounds in the absence of tissue samples  
206 (*calibration 2*, y axis as in Salin, et al. <sup>1</sup>) or from standard curves based on the compounds MitoB and  
207 MitoP together with a fixed amount of each spike and with a tissue sample (*calibration 1*, x axis as in  
208 Cochemé, et al. <sup>2</sup>). The central line is the linear regression line and the two external lines represent the  
209 95% confidence interval of the data.



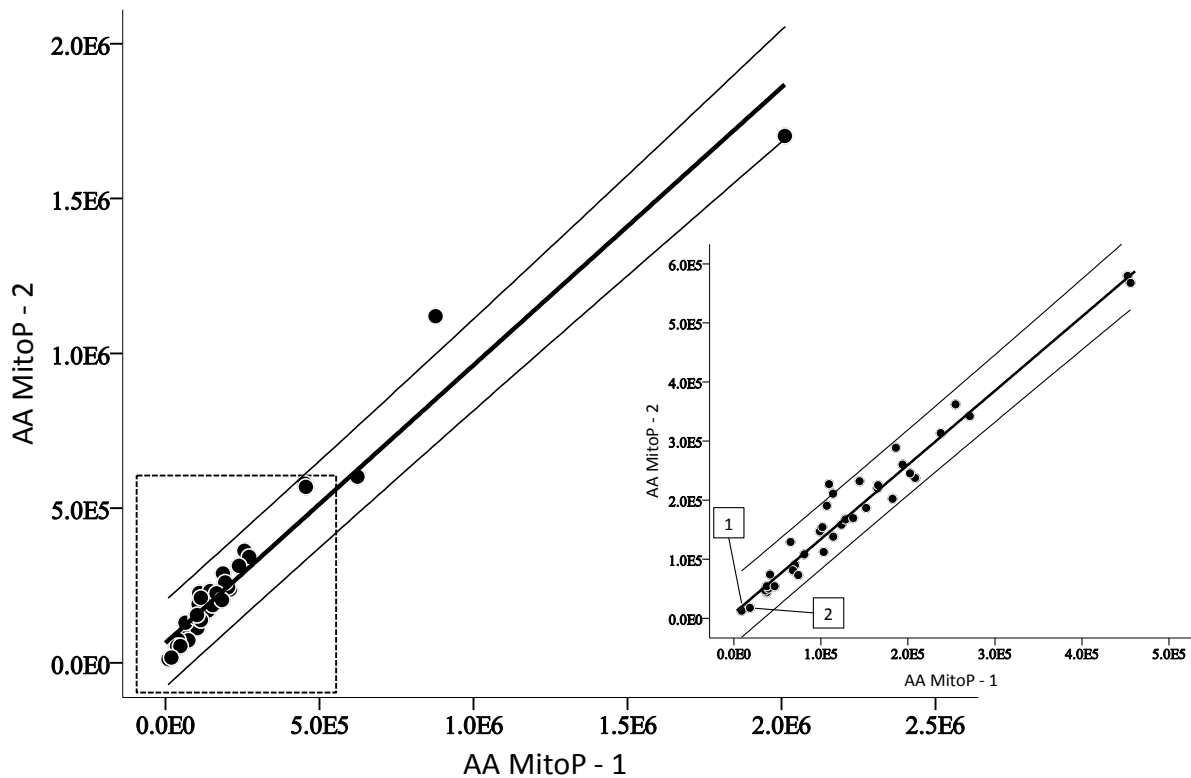
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212 Fig. S3: The repeatability of the quantification of probe levels by HPLC-MS was tested by running  
 213 duplicate extracts of liver samples from 40 individuals exposed to MitoB for about 24 h. The  
 214 replicability of HPLC-MS measurements was very high, as assessed by the relationships between two  
 215 quantifications of (A) MitoP, (B)  $d_{15}$ MitoP, (C) MitoB and (D)  $d_{15}$ MitoB absolute areas (AA) by  
 216 HPLC-MS. The values for MitoP and Mito B content (pmol) per sample were calculated from the  
 217 calibration curves and then corrected using the sample's own coefficients for extraction efficiency  
 218 calculated from the  $d_{15}$ MitoP and  $d_{15}$ Mito B content. (E) The replicability of HPLC-MS  
 219 measurements was confirmed by the strong similarity of the MitoP/MitoB ratios calculated for the  
 220 two samples taken from the same extract. The central thick line is the linear regression line and the  
 221 two external thin lines represent the 95% confidence interval of the data; data points labelled 1 and 2  
 222 refer to potential outliers discussed in the text. Inset of the graph A. shows the data points near the  
 223 origins.

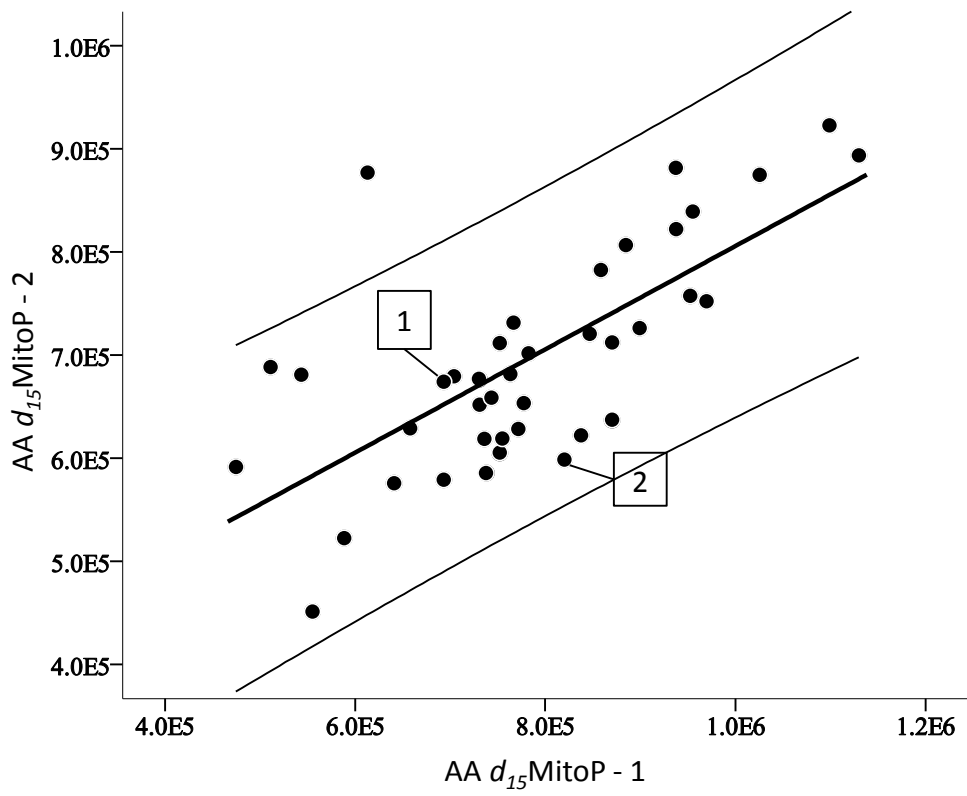
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225 A.



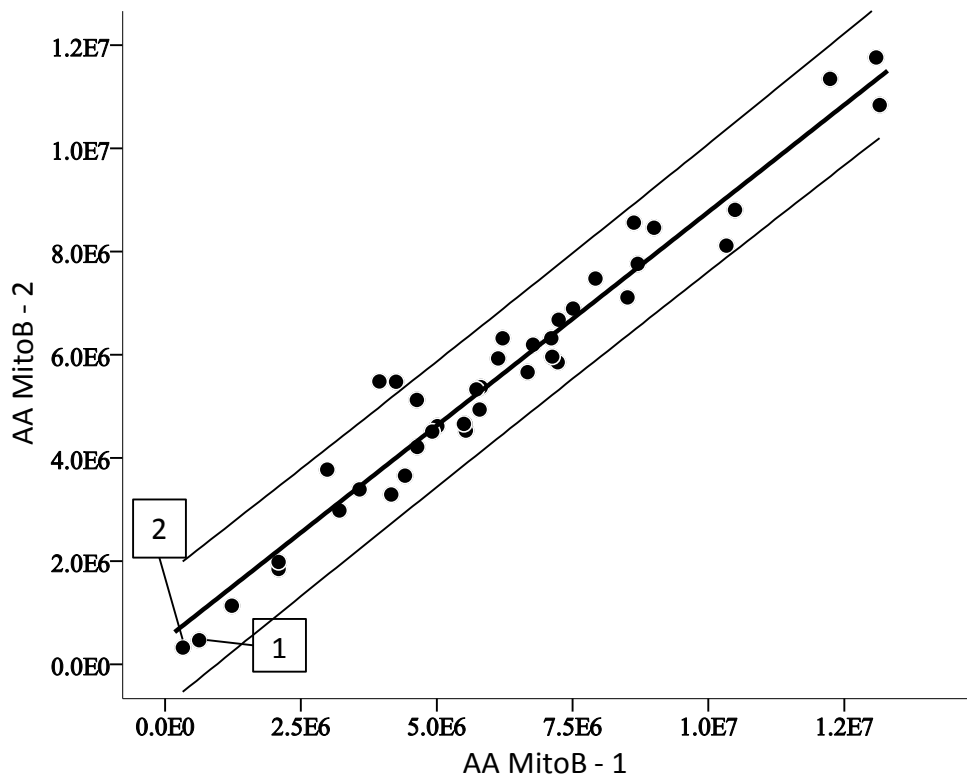
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227 B.



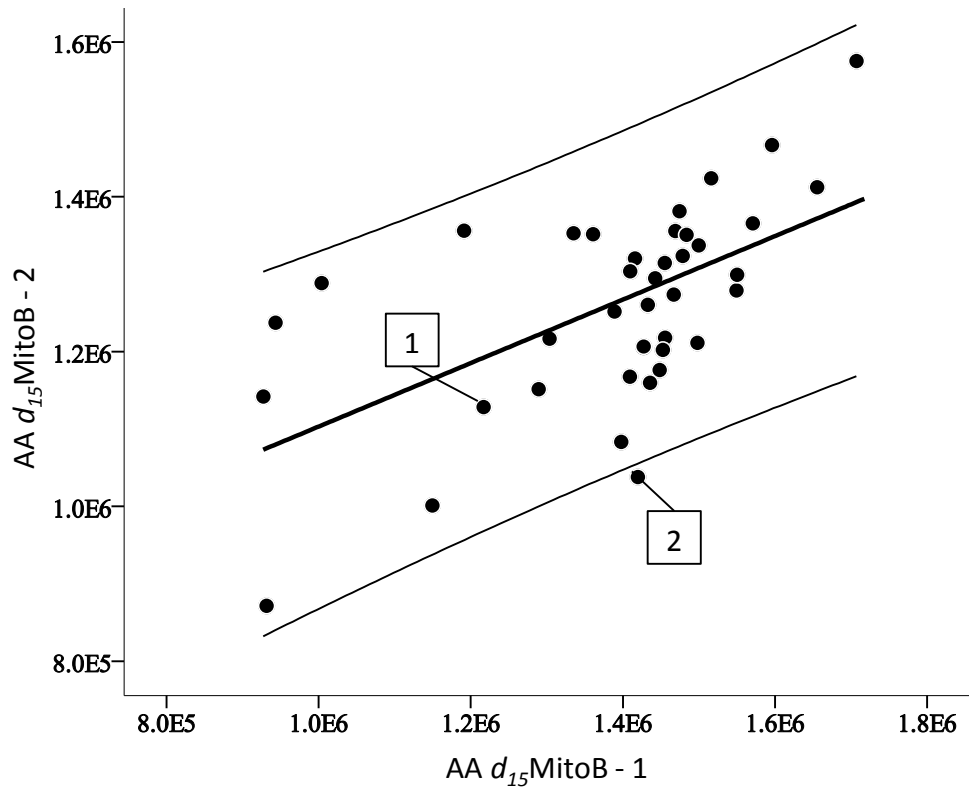
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229 C.



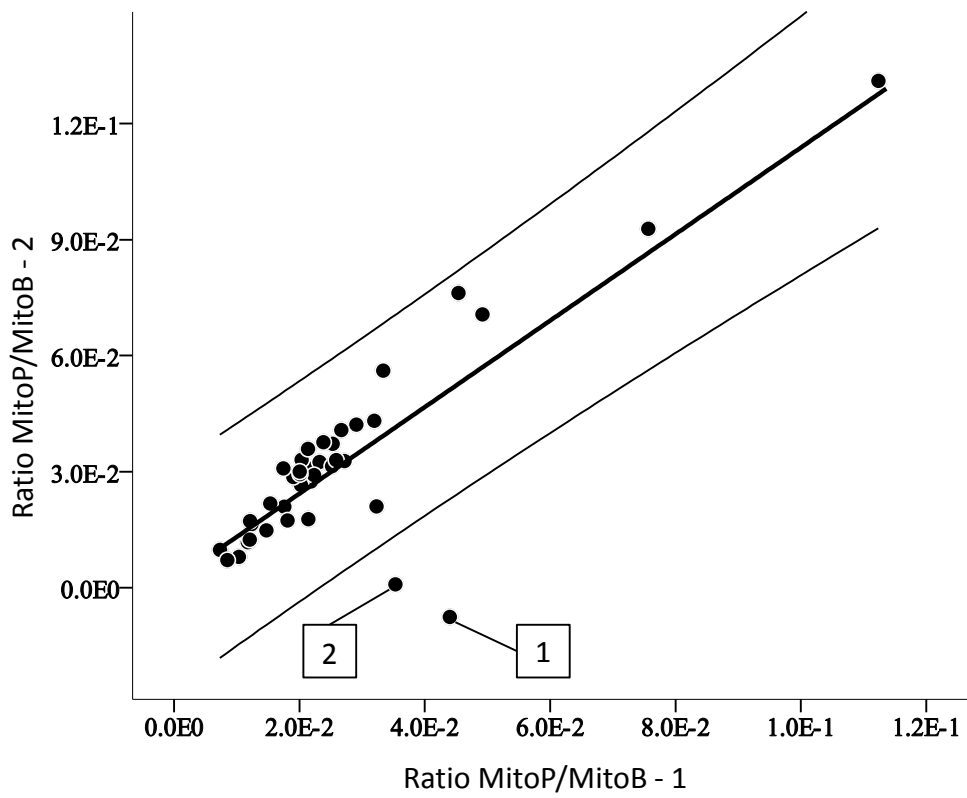
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231 D.



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233 E.



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